



# 2022 New Jersey Air Quality Report

New Jersey Department of Environmental Protection



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<https://nj.gov/dep/airmon/>

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Photo: "Trenton Makes" Bridge, Trenton NJ

## EXECUTIVE SUMMARY

This report presents the New Jersey Department of Environmental Protection (NJDEP) air quality data for 2022, collected from NJDEP's extensive air monitoring network. The state of New Jersey has been monitoring air quality since 1965. During that time, as a result of state, regional and national air pollution reduction efforts, pollution levels have improved significantly.

In 2022, New Jersey had nine days on which there was an exceedance of either the ozone or fine particulate matter (PM<sub>2.5</sub>) National Ambient Air Quality Standards (NAAQS). Ozone pollution in New Jersey tends to be a warm-season problem, since it is formed in the presence of sunlight from other pollutants such as volatile organic compounds and nitrogen oxides. During the 2022 summer ozone season, seven days were "Unhealthy for Sensitive Groups" and one day was "Unhealthy" for ozone. And although particulate matter levels in New Jersey are usually good to moderate, in July a wildfire in the Pine Barrens caused the PM<sub>2.5</sub> concentration at the Brigantine air monitoring station to reach the "Unhealthy" level.

The Newark Firehouse monitoring station, which had been in operation since 2009, had to be shut down on September 26, 2022. NJDEP received notice the month before that construction of a new building was about to begin at the site, and the equipment had to be removed as soon as possible.

### What's in the Annual Air Quality Report

This report includes detailed chapters for ozone, particulate matter, sulfur dioxide, nitrogen dioxide, carbon monoxide and lead. These are the criteria pollutants, that is, those for which National Ambient Air Quality Standards (NAAQS), or criteria, have been set. The Clean Air Act requires the U.S. Environmental Protection Agency (USEPA) to set NAAQS for criteria pollutants at levels that are deemed protective of human health and the environment, based on current science. Other measurements made at New Jersey's air monitoring stations and discussed in this report include air toxics, ozone precursors, and chemical components of airborne fine particles.

The chapter on the Air Quality Index (AQI) describes a national air quality rating system based on the NAAQS, and discusses the overall quality of the state's air in 2022. Included is a detailed list of the nine days on which the AQI was over 100. This means that the NAAQS were exceeded, and the days were classified as "Unhealthy for Sensitive Groups" or "Unhealthy."

Figures 1-1 through 1-6 below illustrate the downward trends in concentrations of criteria pollutants in New Jersey over the past few decades by graphing the statewide design values for each pollutant. A design value is the actual statistic that is compared to a NAAQS. If this value exceeds the NAAQS at any site in the state, the state is determined to be in nonattainment. Design values for each of the criteria pollutants are described in detail in each pollutant-specific chapter of this report.

New Jersey is on track for meeting the ozone NAAQS (Figure 1-1), and will continue to implement control strategies to reduce ambient concentrations. Because ozone is formed in the presence of sunlight and high temperatures, the highest levels occur in the summer months. Ozone has been found to have serious health effects at lower levels than previously thought. In response, the United States Environmental Protection Agency (USEPA) periodically revises and lowers the NAAQS. USEPA lowered the ozone standard to 0.070 ppm in 2015 (effective in 2016).

Particulate air pollution less than 2.5 micrometers in diameter is referred to as fine particulate or PM<sub>2.5</sub>. These small particles can be inhaled deep into the lungs, and are known to have a greater impact on public health than larger particles, which were the focus of the earliest ambient air quality standards. Even though in 2022 there was one day on which the 24-hour PM<sub>2.5</sub> NAAQS was exceeded, monitoring data in New Jersey shows a steady decline in overall PM<sub>2.5</sub> levels, which are now in compliance with the NAAQS (Figure 1-2).

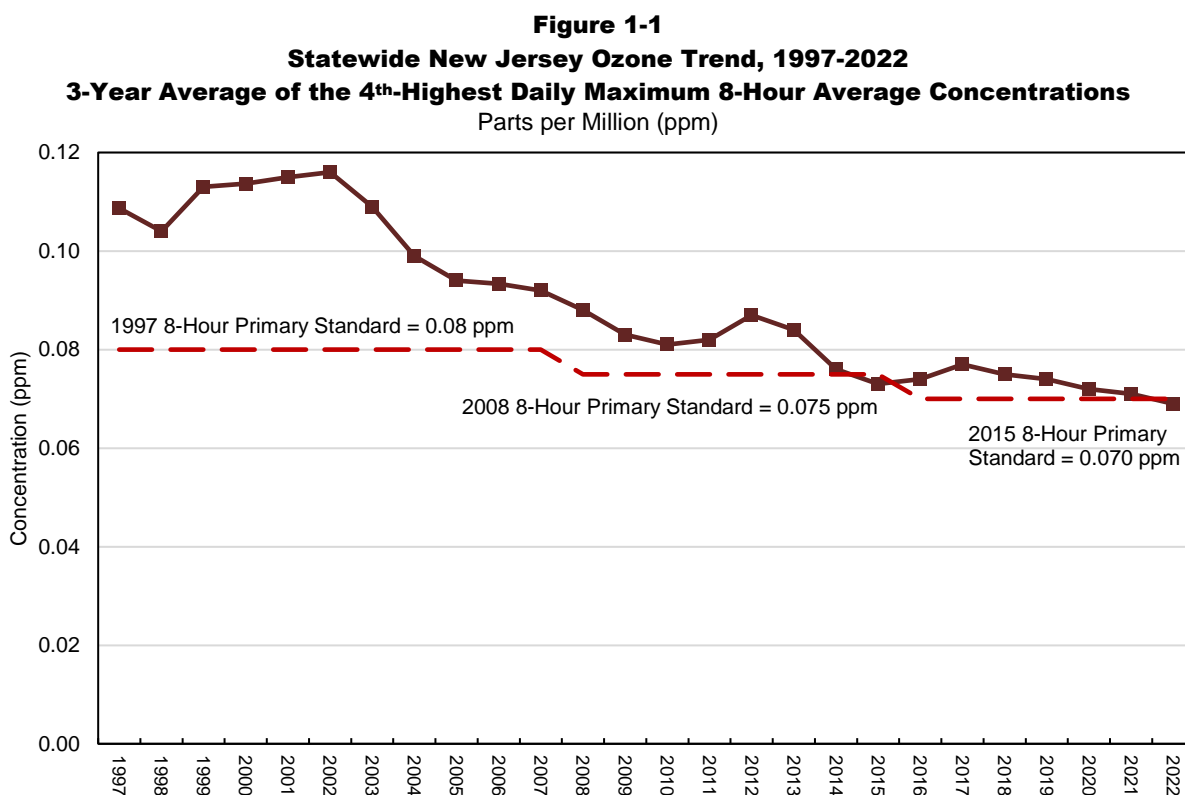
Nitrogen dioxide (NO<sub>2</sub>) is a reactive gas emitted primarily from motor vehicles. It is known to cause serious health problems, especially for sensitive individuals such as children, the elderly, and people with asthma. New Jersey has long been in compliance with the NAAQS for NO<sub>2</sub> (Figure 1-3).

The sharp increase and subsequent decrease in sulfur dioxide (SO<sub>2</sub>) concentrations in New Jersey shown in Figure 1-4 are attributable to a coal-burning facility across the Delaware River in Pennsylvania. NJDEP established the Columbia monitoring station in 2010 to determine that facility's impact on New Jersey's air quality. Exceedances of the SO<sub>2</sub> NAAQS were recorded that same year. Since the plant ceased operations under a court agreement, SO<sub>2</sub> levels in New Jersey have again dropped below the standard.

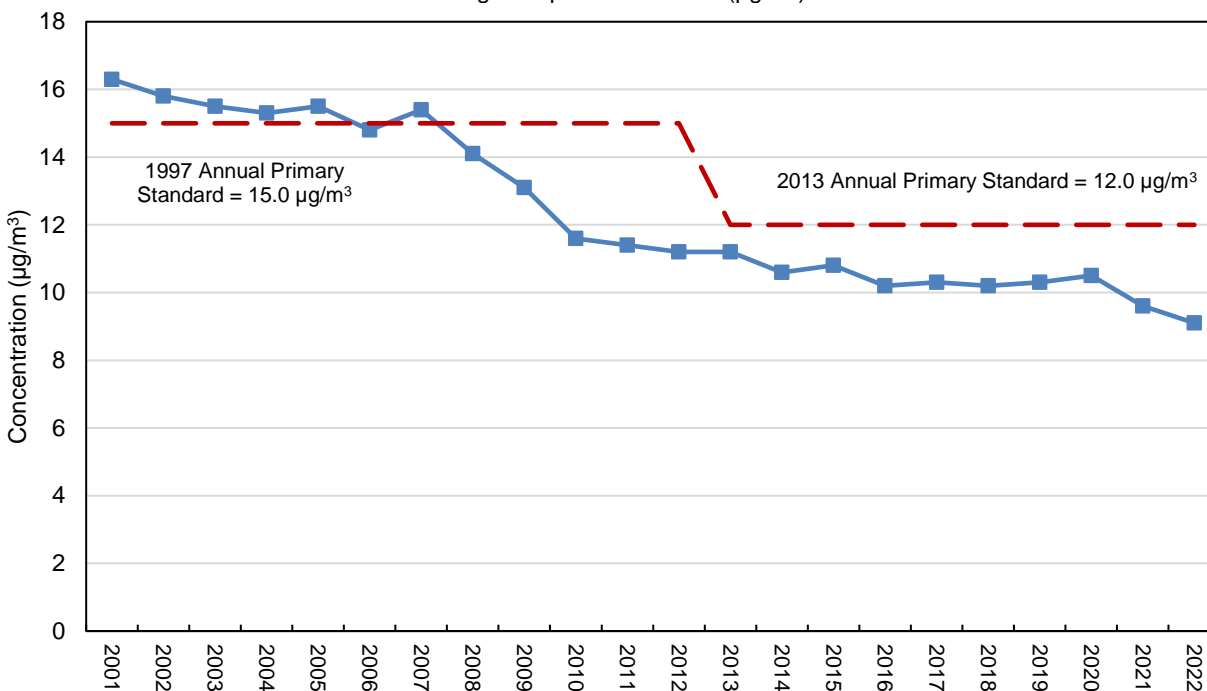
Outdoor concentrations of carbon monoxide can affect people with cardiovascular problems. Levels in New Jersey have been below the NAAQS for over twenty-five years (Figure 1-5).

Air concentrations of lead have dropped dramatically since a standard was established in 1978. The phase-out of leaded gasoline and removal of lead from paint and other products have had a measurable impact. The last exceedances of the NAAQS were in the early 1980s (Figure 1-6).

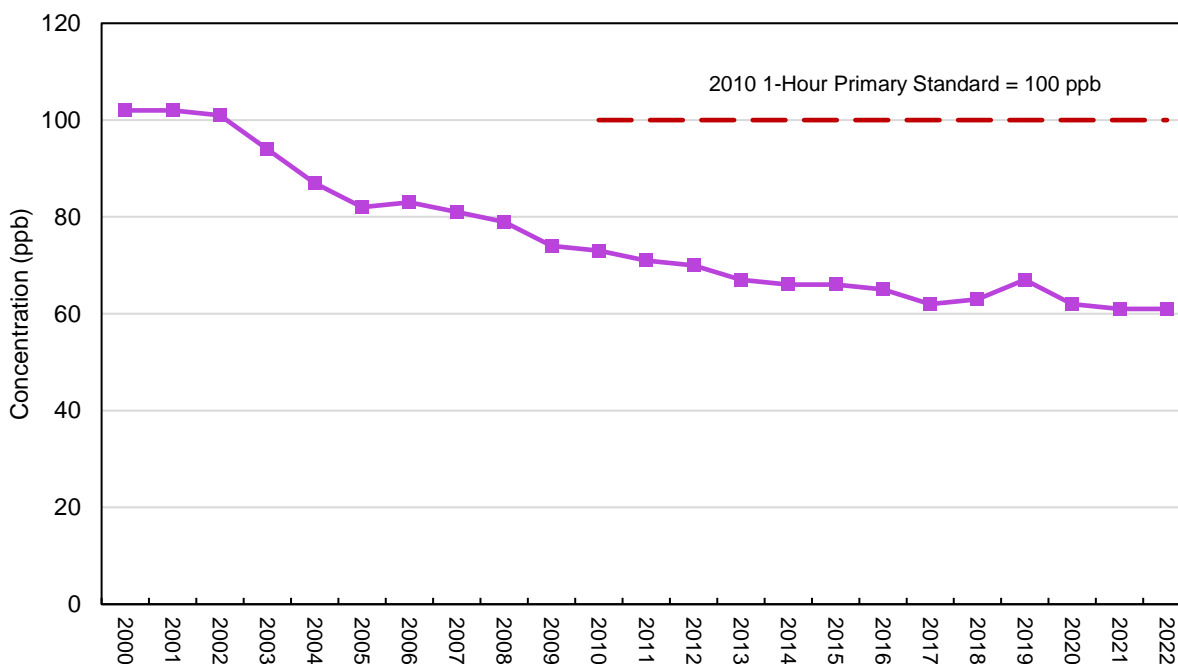
The Bureau of Air Monitoring website can be found at <https://nj.gov/dep/airmon>. Available information includes a table and map of current air quality readings, the daily air quality forecast, annual reports, trend graphs, and other data.



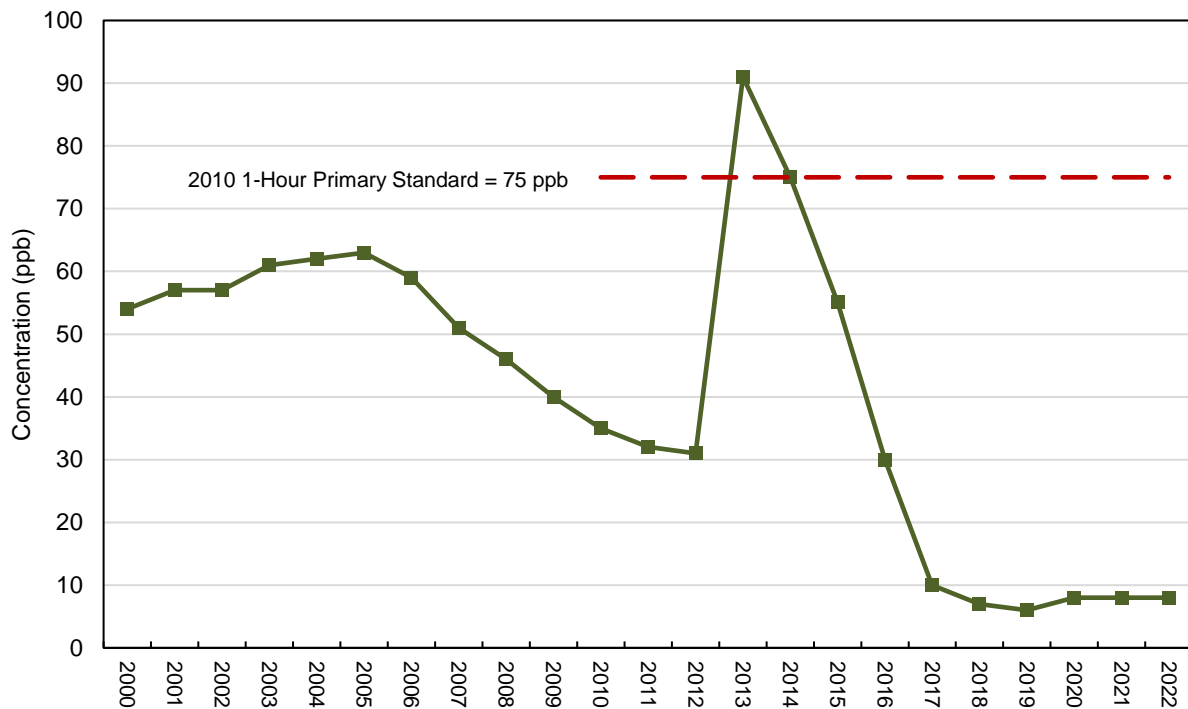
**Figure 1-2**  
**Statewide New Jersey Fine Particulate (PM<sub>2.5</sub>) Trend, 2001-2022**  
**3-Year Average of the Highest Annual Average Concentrations**  
 Micrograms per Cubic Meter (µg/m<sup>3</sup>)



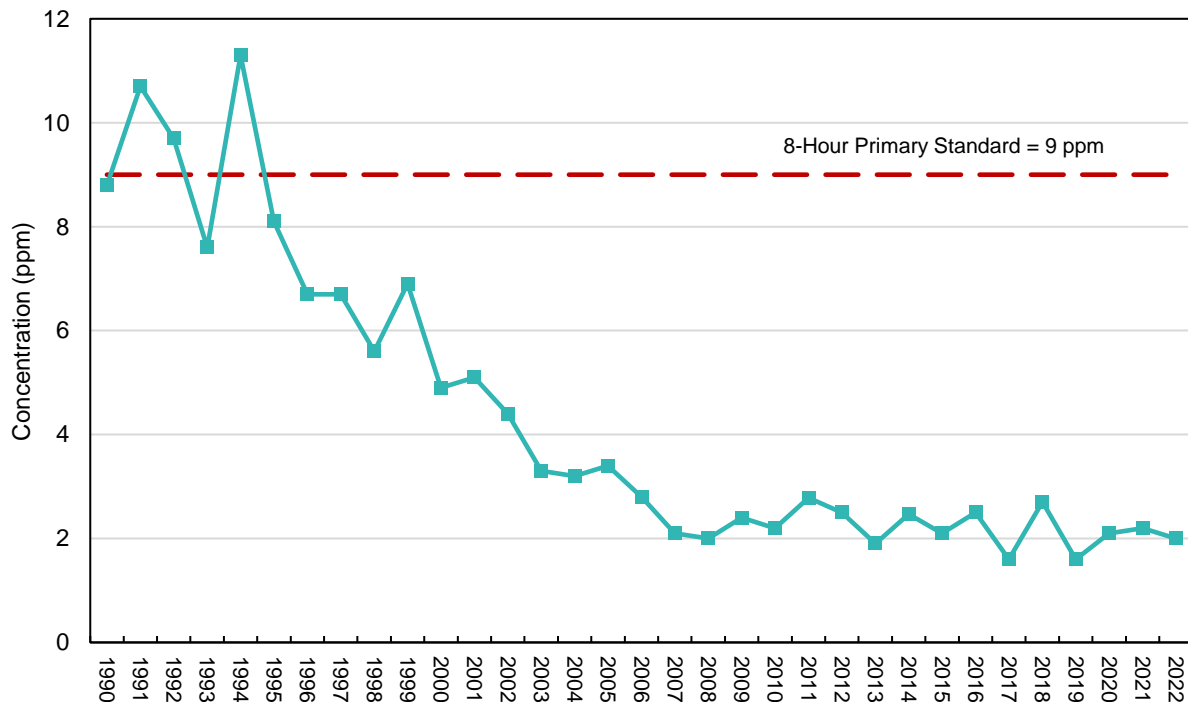
**Figure 1-3**  
**Statewide New Jersey Nitrogen Dioxide (NO<sub>2</sub>) Trend, 2000-2022**  
**3-Year Average of the 98<sup>th</sup> Percentile Daily Maximum 1-Hour Average Concentrations**  
 Parts per Billion (ppb)



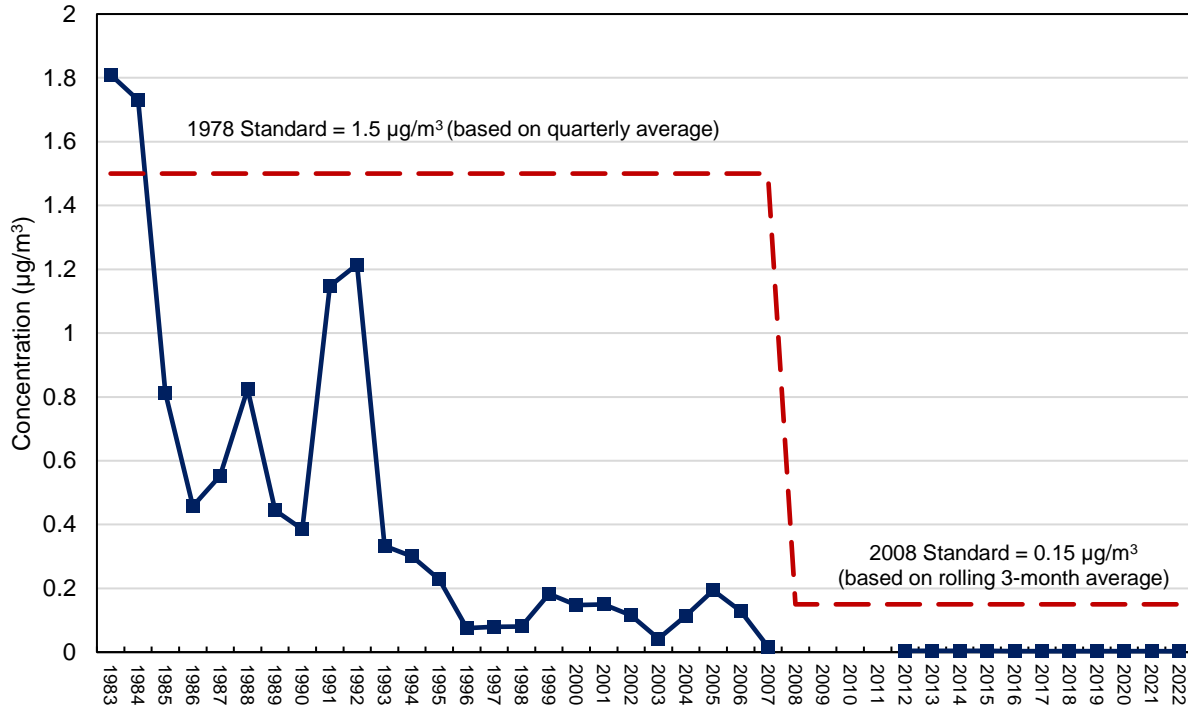
**Figure 1-4**  
**Statewide New Jersey Sulfur Dioxide (SO<sub>2</sub>) Trend, 2000-2022**  
**3-Year Average of the 99<sup>th</sup>-Percentile of Daily Maximum 1-Hour Average Concentrations**  
 Parts per Billion (ppb)



**Figure 1-5**  
**Statewide New Jersey Carbon Monoxide (CO) Trend, 1990-2022**  
**2<sup>nd</sup>-Highest 8-Hour Average Concentrations**  
 Parts per Million (ppm)



**Figure 1-6**  
**Statewide New Jersey Lead Trend, 1983-2022**  
**Highest 3-Month Average Concentrations**  
 Micrograms per Cubic Meter ( $\mu\text{g}/\text{m}^3$ )





# 2022 Air Monitoring Network

New Jersey Department of Environmental Protection

## NETWORK DESCRIPTION

In 2022 the New Jersey Department of Environmental Protection (NJDEP) Bureau of Air Monitoring (BAM) operated a network of thirty ambient air monitoring stations around the state. The monitoring stations vary in the number and type of monitors at each site. New Jersey's air monitoring program is primarily focused on the measurement of pollutants for which National Ambient Air Quality Standards (NAAQS) have been established, also known as criteria pollutants. Criteria pollutant monitoring is regulated by the United States Environmental Protection Agency (USEPA), which prescribes the design and siting of the monitoring networks, the acceptable monitoring methods, and the minimum quality assurance activities. Only data which meet USEPA requirements can be used to determine compliance with the NAAQS. There are six criteria air pollutants: ozone ( $O_3$ ), particulate matter (PM), nitrogen dioxide ( $NO_2$ ), sulfur dioxide ( $SO_2$ ), carbon monoxide (CO), and lead (Pb). Because particulate matter includes a wide range of contaminants, there are separate NAAQS for two different size fractions of particles. There are NAAQS for fine particles, less than 2.5 microns in size, also referred to as  $PM_{2.5}$  (1 micron = one millionth of a meter), and another NAAQS for inhalable particles, less than 10 microns in size, referred to as  $PM_{10}$ .

### Criteria Pollutants

In New Jersey,  $O_3$ ,  $NO_2$ ,  $SO_2$  and CO are measured using USEPA-approved real-time monitoring methods, and data for these pollutants are continuously transmitted to a central data acquisition system. Once an hour, the Bureau of Air Monitoring posts this air quality data to its website (<https://nj.gov/dep/airmon>) and to the USEPA's Air Now website ([www.airnow.gov](http://www.airnow.gov)). Data is subsequently reviewed and certified, and is available from USEPA's Air Quality Database at <https://www.epa.gov/outdoor-air-quality-data>.

$PM_{2.5}$  is measured with both 24-hour filter-based samplers and real-time continuous samplers. Filter-based samplers are used to determine lead and  $PM_{10}$  concentrations as well. Filters must be installed and removed manually, and brought to the BAM lab to be weighed and analyzed. NJDEP is gradually replacing most of its filter-based  $PM_{2.5}$  samplers with real-time samplers, so that the current air quality from those sites can be reported on the BAM website, and also to reduce the need for manpower and time.

Establishment of "near road" stations were required as part of the 2010 revisions to the  $NO_2$  NAAQS. The Fort Lee Near Road monitoring station was established to comply with these requirements, for  $NO_2$  and other monitors in large urban areas with high vehicular traffic. These stations, located within 50 meters of a major roadway where peak hourly  $NO_2$  concentrations are expected to occur, measure the relative worst-case population exposures that could occur in the near-road environment.

The Newark Firehouse monitoring station has been part of the USEPA's National Core Multipollutant Monitoring Network (NCore). NCore is a program that integrates several advanced measurement systems for gaseous pollutants, particles, and meteorology. This includes total reactive oxides of nitrogen,  $NO_y$ . Unfortunately, the Newark Firehouse station had to be shut down in September of 2022, but BAM hopes to establish a new NCore site in Newark by the end of 2023.

### Other Pollutants

Along with criteria pollutants, the NJDEP also measures "non-criteria pollutants," or pollutants that do not have health-based National Ambient Air Quality Standards. Certain non-criteria pollutants are grouped together by their purpose or collection method.



The Rutgers University monitoring site is part of USEPA's Photochemical Assessment Monitoring Station (PAMS) Program. PAMS measures non-criteria pollutants that are important in the formation of ozone. Since most ozone is not directly emitted from sources but forms in the atmosphere when volatile organic compounds and oxides of nitrogen react in the presence of sunlight, it is important to know the levels of these "precursor" pollutants. In addition, PAMS requires monitoring of  $\text{NO}_y$ , and various meteorological parameters.

Other non-criteria pollutants monitored by BAM include some commonly emitted by motor vehicles and other combustion sources: benzene, toluene, ethylbenzene, xylenes (measured with a "BTEX" analyzer), and black carbon (measured with an aethalometer).

Five air monitoring stations collect samples of  $\text{PM}_{2.5}$  that are analyzed to determine the chemical makeup of the particles. These are part of USEPA's Chemical Speciation Network (CSN). This data is used in helping to identify the primary sources of particles, and in assessing potential health effects.

Volatile organic compounds (VOCs) are collected and analyzed at four monitoring sites. These non-criteria pollutants are classified as "air toxics," pollutants that have potential health effects but for which NAAQS have not been established. They can be carcinogenic or have other serious health effects, and are very diverse in their chemical composition.

Cattus Island and Washington Crossing, are part of the National Atmospheric Deposition Network. BAM staff collect precipitation samples and ship them to a national laboratory for analysis of acids, nutrients, and base cations. Mercury is measured at Elizabeth Lab and Rutgers.

A number of sites within the air monitoring network also take measurements of meteorological parameters, such as temperature, relative humidity, barometric pressure, wind speed, wind direction, rain, and solar radiation.

Figure 2-1 shows the Ramapo seasonal ozone monitoring station in Passaic County.

## CHANGES TO THE NETWORK IN 2022

The most significant change to New Jersey's air monitoring network in 2022 was the abrupt shutdown of the Newark Firehouse station. In August 2022, BAM received notice that construction of a new building was about to start next to the Newark Firehouse air monitoring station, and that our equipment would have to be removed as soon as possible. Data collection ended there on September 26, and the station was subsequently shut down and dismantled. The search for a replacement site in Newark is ongoing.

BAM is in the process of replacing its filter-based  $\text{PM}_{2.5}$  monitors with continuous monitors. These provide 1-hour data in real time, which can then be displayed on BAM's air quality website. BAM plans to complete the switch during the next few years. In 2022, real-time  $\text{PM}_{2.5}$  monitoring started at Union City High School and Paterson. In 2021, both sites had remained off-limits for reasons related to Covid-19 shutdowns, and the filter-based instruments could not be accessed.

The locations of all the monitoring stations that operated in 2022 are displayed on the map in Figure 2-2. Table 2-1 lists the parameters that were measured at each site. More information about the monitoring stations can be found in Appendix A.

**Figure 2-1**  
**Ramapo Air Monitoring Station**



**Figure 2-2**  
**New Jersey Air Monitoring Sites in 2022**



**Table 2-1**  
**2022 New Jersey Air Monitoring Network Parameters**

Monitoring Parameter		O <sub>3</sub>	PM <sub>2.5</sub> (Filter-based)	Real-Time PM <sub>2.5</sub>	PM <sub>10</sub>	NO <sub>2</sub>	NOy	SO <sub>2</sub>	CO	Lead	Toxics	PM <sub>2.5</sub> -Speciation	O <sub>3</sub> Precursors (PAMS)	BTEX & Black Carbon	Visibility	Acid Deposition	Mercury	Meteorological <sup>a</sup>	Solar Radiation
1	Ancora State Hospital	X																	
2	Atlantic City		X																
3	Bayonne	X				X		X						X				X	
4	Brigantine	X	X	X				X							X	X			
5	Camden Spruce Street	X	X	X	X	X		X	X		X	X		X				X	
6	Cattus Island															X			
7	Chester	X	X			X		X			X	X							
8	Clarksboro	X	X																
9	Colliers Mills	X																	
10	Columbia	X		X		X		X										X	
11	Elizabeth							X	X										
12	Elizabeth Lab		X	X		X		X	X		X	X		X			X	X	
13	Flemington	X		X														X	
14	Fort Lee Near Road			X		X			X					X				X	
15	Jersey City					X		X	X										
16	Jersey City Firehouse		X	X	X														
17	Leonia	X																	
18	Millville	X		X		X													
19	Monmouth University	X																	
20	Newark Firehouse*	X	X	X	X	X	X	X	X	X		X		X				X	X
21	Paterson**			X															
22	Pennsauken		X																
23	Rahway			X															
24	Ramapo	X																	
25	Rider University	X		X														X	
26	Rutgers University	X	X	X		X	X				X	X	X				X	X	X
27	Toms River			X															
28	Trenton			X															
29	Union City High School**			X															
30	Washington Crossing			X												X			
	TOTAL	16	10	16	3	10	2	9	6	1	4	5	1	5	1	3	2	9	2

X - Parameter measured in 2022.

\*Site shut down as of 9/26/2022.

\*\*Continuous PM<sub>2.5</sub> monitor operated part of the year.

NO<sub>2</sub> usually includes NO and NO<sub>x</sub>.

a - Meteorological parameters include temperature, relative humidity, barometric pressure, wind direction & wind speed.

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# 2022 Air Quality Index

New Jersey Department of Environmental Protection

## What is the Air Quality Index (AQI)?

The Air Quality Index (AQI) is a national air quality rating system based on the National Ambient Air Quality Standards (NAAQS). An index value of 100 is equal to the primary, or health-based, NAAQS for each pollutant. This allows for a comparison of each of the pollutants used in the AQI. These pollutants are ozone, particulate matter, carbon monoxide, sulfur dioxide, and nitrogen dioxide. Although air concentrations of pollutants have been dropping over the past few years, the U.S. Environmental Protection Agency (USEPA) periodically reviews the NAAQS to make sure that they are protective of public health, and adjusts them accordingly in response to new research. The latest NAAQS revision, for ozone, occurred in October 2015.

Every morning an air pollution forecast for the current and following day is prepared by the New Jersey Department of Environmental Protection (NJDEP) using the AQI format. The forecast is provided to USEPA and is disseminated through the Enviroflash system to subscribers who sign up to receive air quality forecast and alert emails or texts ([www.enviroflash.info](http://www.enviroflash.info)). Anyone can view the forecast and current air quality conditions at USEPA's AirNow website ([www.airnow.gov](http://www.airnow.gov)) or at NJDEP's air monitoring webpage (<https://nj.gov/dep/airmon>).

In an effort to make the AQI easier to understand, a color code and descriptive interpretation are assigned to the numerical ratings (see Table 3-1). Table 3-2 contains suggested actions to take to protect public health for different AQI levels. For more information on the AQI, visit EPA's web site at [www.airnow.gov](http://www.airnow.gov).

**Table 3-1**  
**Air Quality Index Levels and Associated Health Impacts**

<b>AQI Level of Health Concern</b>	<b>Numerical Value</b>	<b>Meaning</b>	<b>Color Code</b>
<b>Good</b>	0 to 50	Air quality is considered satisfactory, and air pollution poses little or no risk.	<b>Green</b>
<b>Moderate</b>	51 to 100	Air quality is acceptable; however, for some pollutants there may be a moderate health concern for a very small number of people who are unusually sensitive to air pollution.	<b>Yellow</b>
<b>Unhealthy for Sensitive Groups</b>	101 to 150	Members of sensitive groups may experience health effects. The general public is not likely to be affected.	<b>Orange</b>
<b>Unhealthy</b>	151 to 200	Everyone may begin to experience health effects; members of sensitive groups may experience more serious health effects.	<b>Red</b>
<b>Very Unhealthy</b>	201 to 300	Health warnings of emergency conditions. The entire population is more likely to be affected.	<b>Purple</b>
<b>Hazardous</b>	301 to 500	Health alert: everyone may experience more serious health effects.	<b>Maroon</b>

**Table 3-2**  
**AQI Suggested Actions to Protect Health**

Air Quality Index Level	AQI Value Actions to Protect Your Health
<b>Good (1-50)</b>	None
<b>Moderate (51-100)</b>	Unusually sensitive individuals should consider limiting prolonged outdoor exertion.
<b>Unhealthy for Sensitive Groups (101-150)</b>	Children, active adults, and people with respiratory disease such as asthma should limit prolonged outdoor exertion.
<b>Unhealthy (151-200)</b>	Children, active adults, and people with respiratory disease such as asthma should avoid prolonged outdoor exertion: Everyone else should limit prolonged outdoor exertion.
<b>Very Unhealthy (201-300)</b>	Children, active adults, and people with respiratory disease such as asthma should avoid outdoor exertion. Everyone else should limit outdoor exertion.
<b>Hazardous (301-500)</b>	Everyone should avoid all physical activity outdoors.

Table 3-3 shows the pollutant-specific ranges for the AQI categories. These are set according to the corresponding NAAQS.

**Table 3-3**  
**AQI Pollutant-Specific Ranges**

Category	AQI Level	O <sub>3</sub>	PM <sub>2.5</sub>	NO <sub>2</sub>	SO <sub>2</sub>	CO
		(ppm) 8-hour	(µg/m <sup>3</sup> ) 24-hour	(ppb) 1-hour	(ppb) 1-hour	(ppm) 8-hour
<b>Good</b>	0-50	0.000-0.054	0.0-12.0	0-53	0-35	0.0-4.4
<b>Moderate</b>	51-100	0.055-0.070	12.1-35.4	54-100	36-75	4.5-9.4
<b>Unhealthy for Sensitive Groups</b>	101-150	0.071-0.085	35.5-55.4	101-.360	76-185	9.5-12.4
<b>Unhealthy</b>	151- 200	0.086-0.105	55.5-150.4	361-649	186-304	12.5-15.4
<b>Very Unhealthy</b>	201-300	0.106-0.200	150.5-250.4	605-1249	305-604	15.5-30.4
<b>Hazardous</b>	301-500	>0.200	250.5-500.4	1250-2049	605-1004	30.5-1004

Pollutants:

O<sub>3</sub> – Ozone  
 PM<sub>2.5</sub> – Fine particulate matter  
 NO<sub>2</sub> – Nitrogen dioxide  
 SO<sub>2</sub> – Sulfur dioxide  
 CO – Carbon monoxide

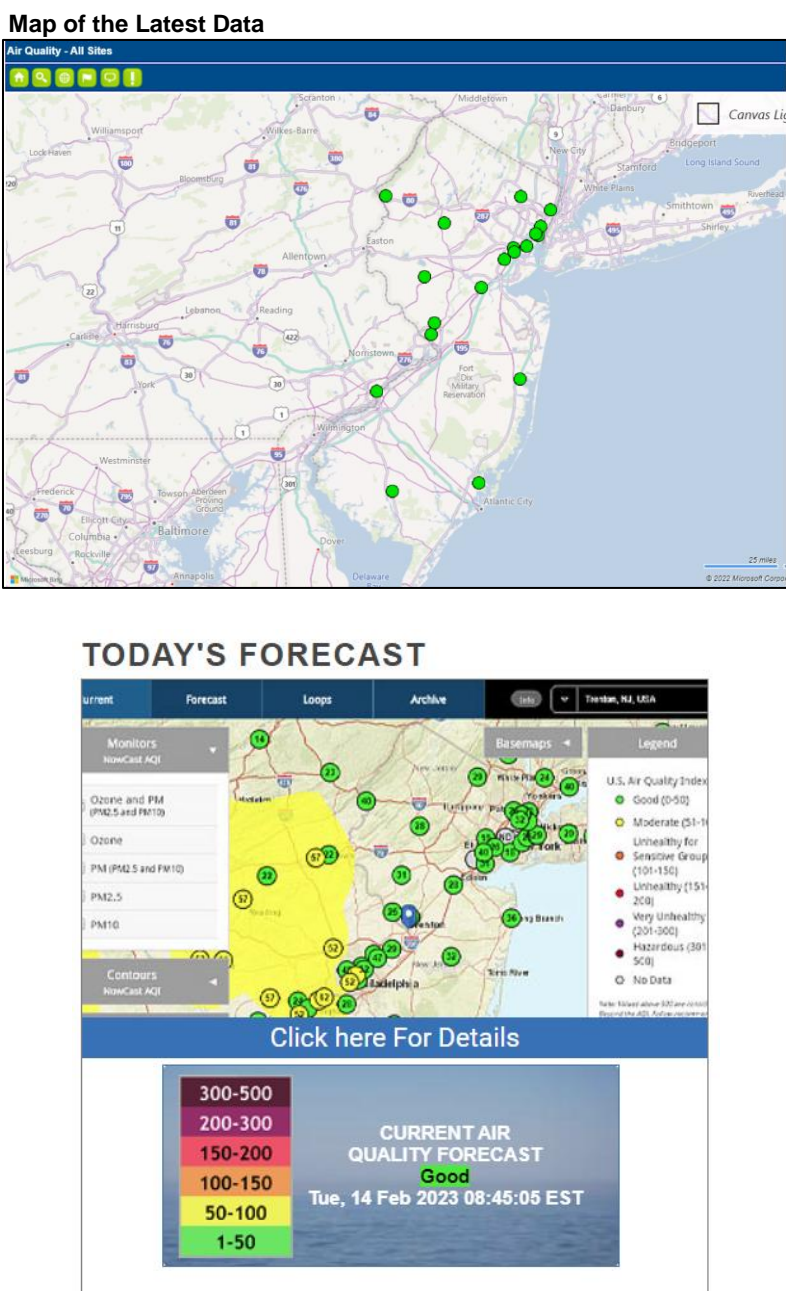
Units:

ppm – parts per million  
 µg/m<sup>3</sup> – micrograms per cubic meter  
 ppb – parts per billion



On days when the air quality is expected to reach the “Unhealthy for Sensitive Groups” range or above, cautionary statements similar to those in Tables 3-1 and 3-2 are provided as part of the forecast. These air quality alerts are issued through Enviroflash emails, are displayed on the AirNow and NJDEP air monitoring websites, and can also be viewed on the National Weather Service page for the Philadelphia/Mount Holly area (<http://airquality.weather.gov/>). Maps, tables, annual trends and other air quality information are also available on the NJDEP air monitoring web site, as shown in Figure 3-1 below.

**Figure 3-1**  
**Examples of Information Available on NJDEP’s Air Monitoring Website**  
<https://nj.gov/dep/airmon>



## 2022 New Jersey AQI Summary

Not all of New Jersey's monitoring sites have 365 (or 366) days of reported air quality index values. Certain ozone monitors only operate during "ozone season," from March through October. Also, not all monitoring sites measure all pollutants. Table 3-4 shows which pollutants are used to determine the daily AQI at different monitoring stations.

There is also an ozone monitor at Washington Crossing State Park that is managed by USEPA. Although it is not officially part of the NJDEP network and does not report to the BAM website, its data is included in determining exceedances in New Jersey.

**Table 3-4**  
**Pollutants Monitored at Each Air Quality Index Monitoring Site**  
**in New Jersey in 2022**

	Monitoring Site	Ozone	Particulate Matter	Carbon Monoxide	Sulfur Dioxide	Nitrogen Dioxide
1	Ancora State Hospital	√ (s)				
2	Bayonne	√			√	√
3	Brigantine	√	√		√	
4	Camden Spruce St.	√	√	√	√	√
5	Chester	√			√	√
6	Clarksboro	√ (s)				
7	Colliers Mills	√ (s)				
8	Columbia	√	√		√	√
9	Elizabeth			√	√	
10	Elizabeth Lab		√	√	√	√
11	Flemington	√	√			
12	Fort Lee Near Road		√	√		√
13	Jersey City			√	√	√
14	Jersey City Firehouse		√			
15	Leonia	√ (s)				
16	Millville	√	√			√
17	Monmouth University	√ (s)				
18	Newark Firehouse*	√	√	√	√	√
19	Paterson		√			
20	Rahway		√			
21	Ramapo	√ (s)				
22	Rider University	√	√			
23	Rutgers University	√	√			
24	Toms River		√			
25	Trenton		√			
26	Union City High School		√			

(s) – Seasonal operation only (March 1 through October 31).

\*Newark Firehouse shut down on 9/26/2022.



A summary of the 2022 AQI ratings for New Jersey is displayed in the pie chart in Figure 3-2 below. In 2022, there were 189 “Good” days (52%), 167 “Moderate” days (46%), seven “Unhealthy for Sensitive Groups” days (2%), and two “Unhealthy” days (1%).

**Figure 3-2**  
**2022 Air Quality Summary by Days**

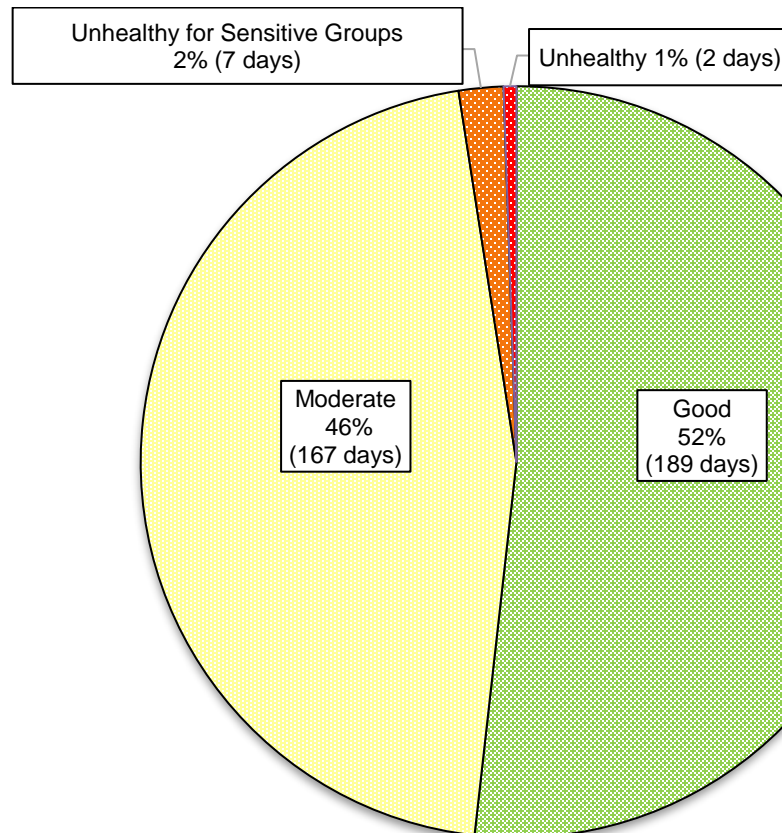


Figure 3-3 shows the distribution of AQI days since 2000. It should be noted that AQI ranges change whenever a NAAQS for a specific pollutant is revised. So even though improvement in AQI days appears to be somewhat erratic, to see how things really have improved, refer to the concentration trend graphs in the individual criteria pollutant reports or in the executive summary.

**Figure 3-3**  
**Number of Days in Each AQI Category Since 2000**

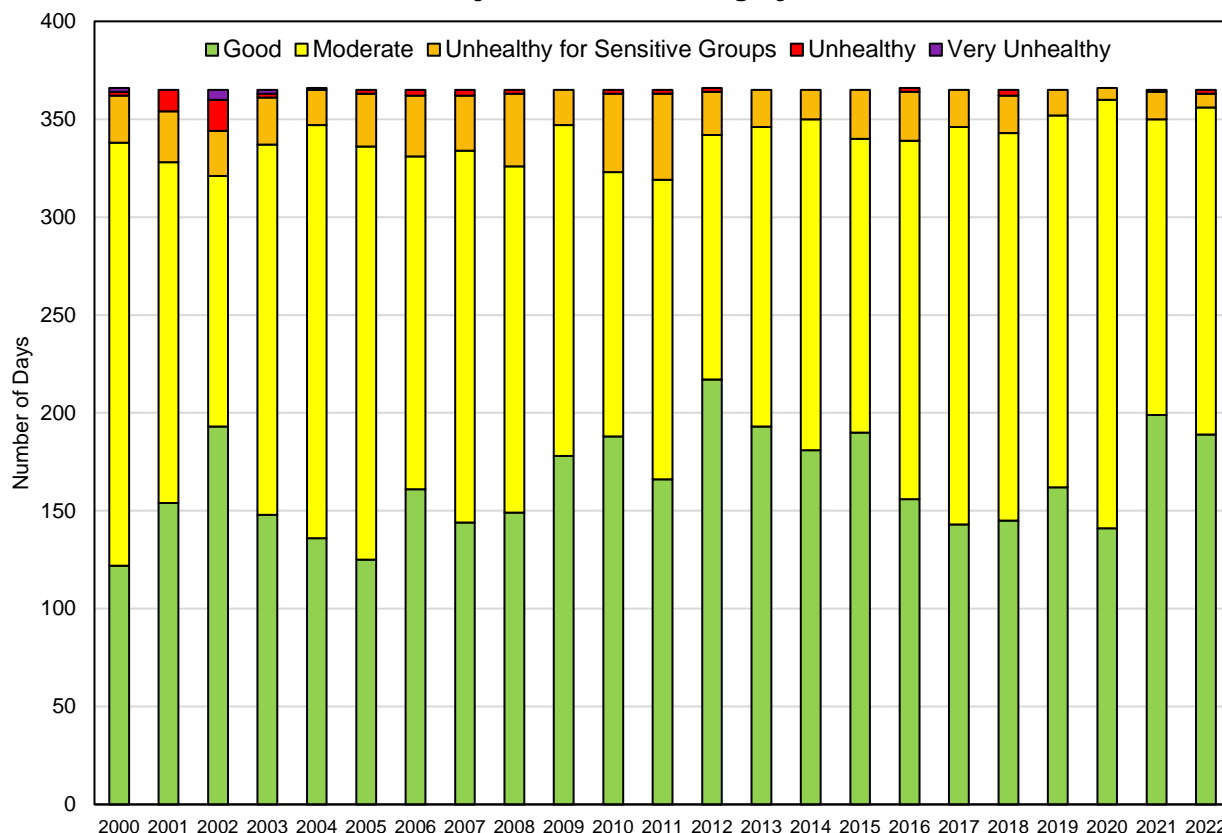


Table 3-5 shows the number of days when the AQI reached or exceeded the “Unhealthy for Sensitive Groups” or “Unhealthy” threshold at any monitoring location in New Jersey in 2022. Table 3-6 lists the individual exceedance dates and shows the specific pollutants and their locations and concentrations.

Of all the criteria pollutants, ozone is predominantly responsible for AQI days above the moderate range in New Jersey. Exceedances are the result of weather conditions that favor the formation and transport of ozone. Ozone forms when emissions of nitrogen oxides and volatile organic compounds undergo chemical reactions in the presence of sunlight.

The PM<sub>2.5</sub> exceedance on June 20 is attributable to a fire in Wharton State Forest in the Pine Barrens, known as the Mullica River Fire.

**Table 3-5**  
**2022 Total Number of NAAQS Exceedance Days in New Jersey**

Pollutant	Total Exceedance Days	Unhealthy for Sensitive Groups Days	Unhealthy Days
Ozone	8	7	1
PM <sub>2.5</sub>	1	---	1
<b>TOTAL</b>	<b>9</b>	<b>7</b>	<b>2</b>

**Table 3-6**  
**AQI Days Over 100 in New Jersey in 2022**

Day No.	Date	Monitor Location	Pollutant	Concentration	Units	AQI Rating	AQI Value
1	6/4/22	Clarksboro	O <sub>3</sub>	0.071	ppm	USG	101
2*	6/20/22	Brigantine	PM <sub>2.5</sub>	103.6	µg/m <sup>3</sup>	U	176
3	6/30/22	Ancora	O <sub>3</sub>	0.073	ppm	USG	108
		Clarksboro	O <sub>3</sub>	0.078	ppm	USG	126
		Colliers Mills	O <sub>3</sub>	0.072	ppm	USG	105
4	7/11/22	Rider University	O <sub>3</sub>	0.071	ppm	USG	101
5	7/14/22	Monmouth University	O <sub>3</sub>	0.090	ppm	U	161
6	7/20/22	Bayonne	O <sub>3</sub>	0.075	ppm	USG	115
		Colliers Mills	O <sub>3</sub>	0.071	ppm	USG	101
		Rider University	O <sub>3</sub>	0.071	ppm	USG	101
7	7/22/22	Colliers Mills	O <sub>3</sub>	0.071	ppm	USG	101
8	7/23/22	Bayonne	O <sub>3</sub>	0.072	ppm	USG	105
		Clarksboro	O <sub>3</sub>	0.075	ppm	USG	115
		Monmouth University	O <sub>3</sub>	0.073	ppm	USG	108
9	7/27/22	Rider University	O <sub>3</sub>	0.076	ppm	USG	119
		Rutgers University	O <sub>3</sub>	0.071	ppm	USG	101

\*This exceedance is attributable to the Mullica River Fire in Wharton State Forest.

Rating

USG = Unhealthy for Sensitive Groups

U = Unhealthy

Pollutants

O<sub>3</sub> – Ozone

PM<sub>2.5</sub> – Fine particulate matter

Units

ppm – parts per million

µg/m<sup>3</sup> - micrograms per cubic meter

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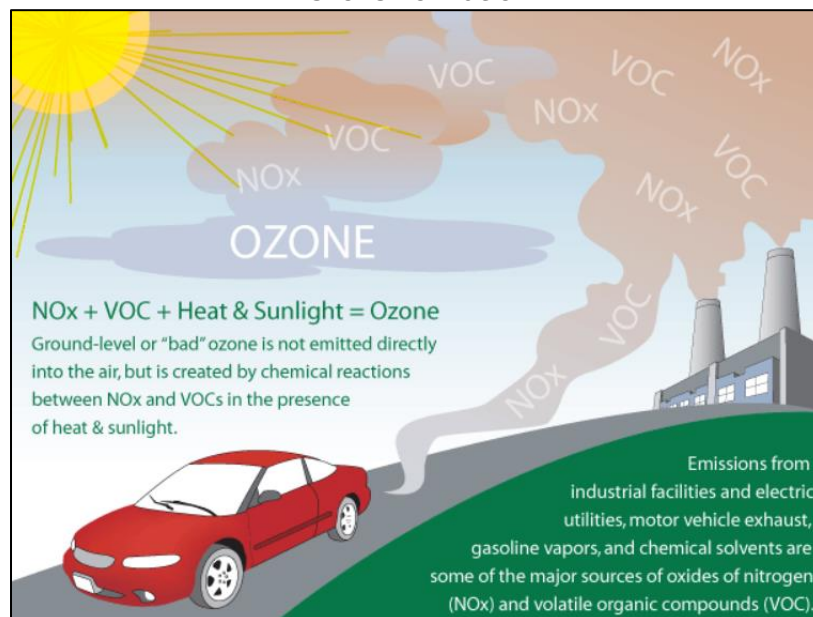
# 2022 Ozone Summary

New Jersey Department of Environmental Protection

## SOURCES

Ozone ( $O_3$ ) is a gas consisting of three oxygen atoms. It occurs naturally in the upper atmosphere (stratospheric ozone) where it protects us from harmful ultraviolet rays. However, at ground-level (tropospheric ozone), it is considered an air pollutant and can have serious adverse health effects. Ground-level ozone is created when nitrogen oxides ( $NO_x$ ) and volatile organic compounds (VOCs) react in the presence of sunlight (see Figure 4-1).  $NO_x$  is primarily emitted by motor vehicles, power plants, and other sources of combustion. VOCs are emitted from sources such as motor vehicles, chemical plants, factories, consumer and commercial products, and even natural sources such as trees. The pollutants that form ozone, referred to as “precursor” pollutants, and ozone itself can also be transported into an area from sources hundreds of miles upwind.

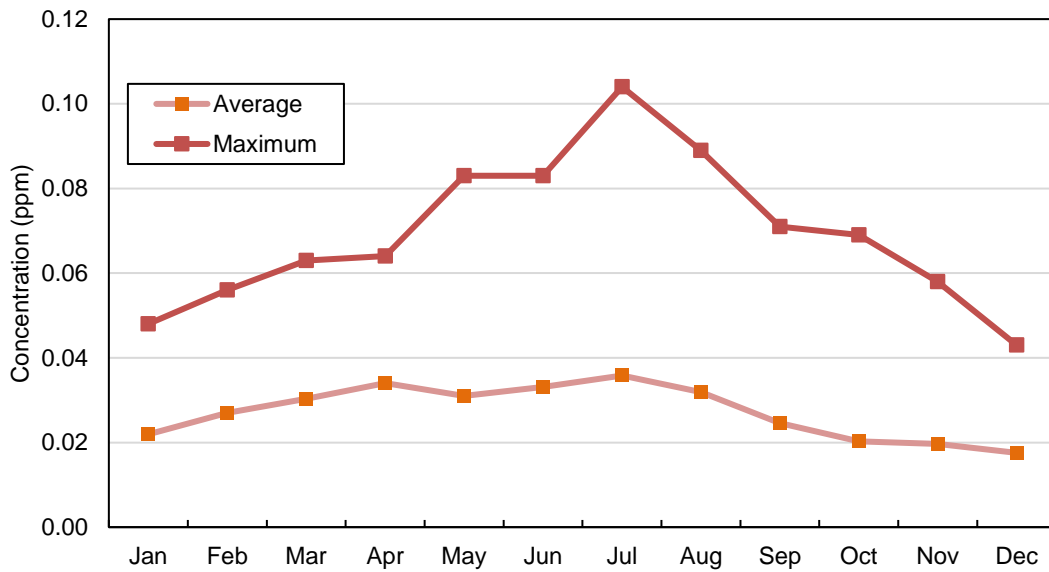
**Figure 4-1**  
**Ozone Formation**



<https://www.epa.gov/ground-level-ozone-pollution/ground-level-ozone-basics#wwh>

Since ground-level ozone needs sunlight to form, it is mainly a problem in the daytime during the summer months. The U.S. Environmental Protection Agency (USEPA) requires New Jersey to monitor ozone from March 1<sup>st</sup> to October 31<sup>st</sup>, the so-called “ozone season.” However, weather patterns have a significant effect on ozone formation, and hot dry summers will result in higher levels than cool wet ones. Figures 4-2 and 4-3 show the effect of sunlight on ambient ozone concentrations. In 2022, the highest monthly average and maximum hourly values occurred in July. The highest daily and maximum 1-hour averages occurred after 2 p.m.

**Figure 4-2**  
**2022 Ozone Concentrations in New Jersey**  
**Monthly Variation of Statewide Hourly Averages**  
 Parts per Million (ppm)



**Figure 4-3**  
**2022 Ozone Concentrations in New Jersey**  
**Daily Variation of Statewide Hourly Averages, March through October**  
 Parts per Million (ppm)

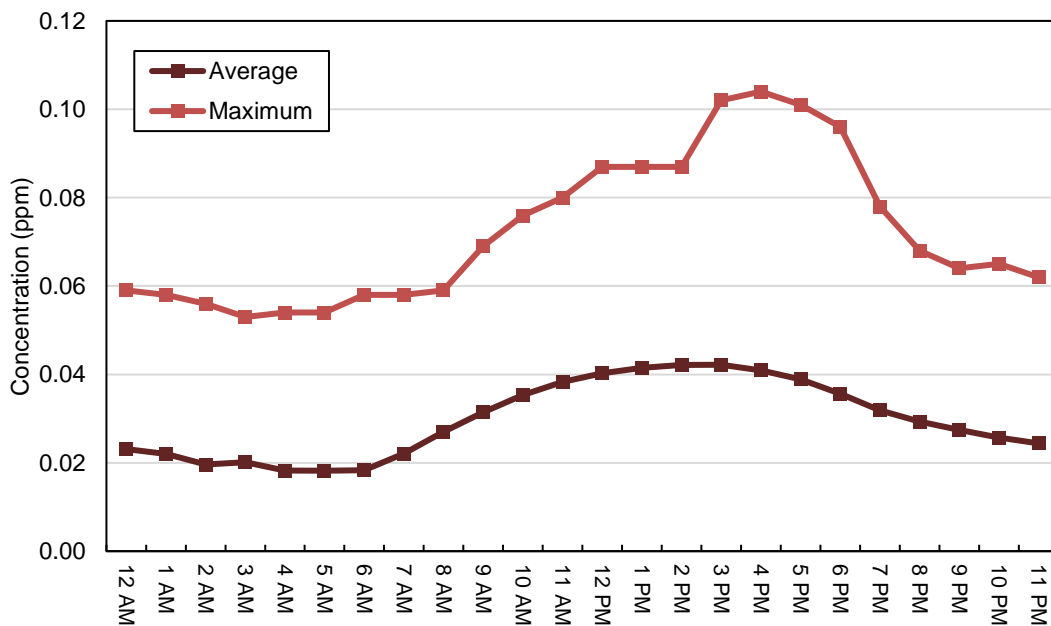
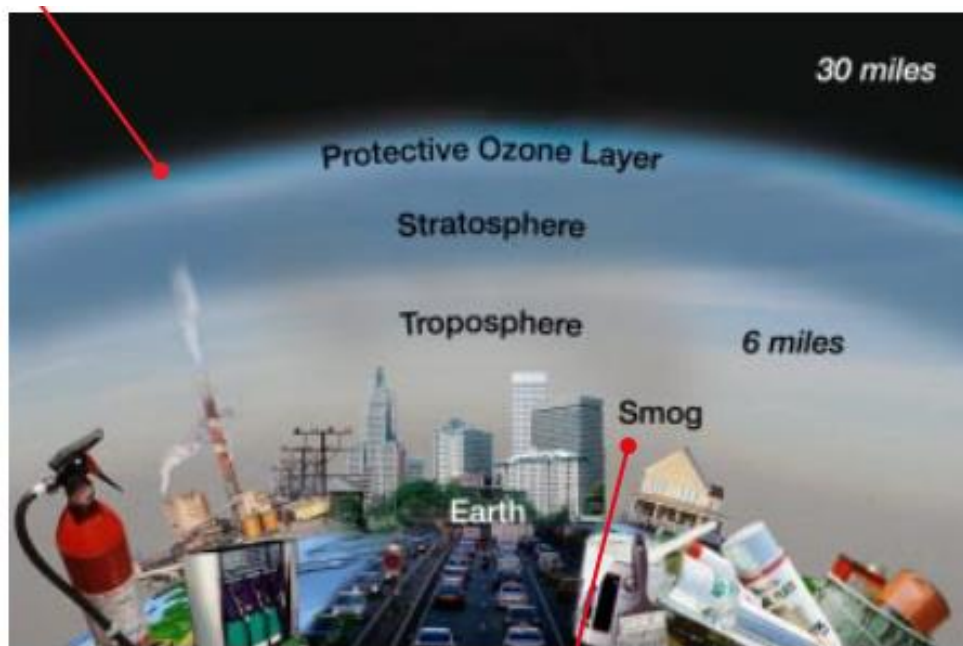


Figure 4-4 explains the difference between ozone in the upper and lower atmosphere. For more information, refer to the USEPA publication, “Good Up High, Bad Nearby – What is Ozone?”

**Figure 4-4. Good and Bad Ozone**

Ozone is good up here...Many popular consumer products like air conditioners and refrigerators involve CFCs or halons during either manufacturing or use. Over time, these chemicals damage the earth's protective ozone layer.



Ozone is bad down here...Cars, trucks, power plants and factories all emit air pollution that forms ground-level ozone.

<https://www.epa.gov/sites/production/files/documents/gooduphigh.pdf>

## HEALTH AND ENVIRONMENTAL EFFECTS

Ozone can irritate the entire respiratory tract. Repeated exposure to ozone pollution may cause permanent damage to the lungs. Even when ozone is present at low levels, inhaling it can trigger a variety of health problems including chest pains, coughing, nausea, throat irritation, and congestion. Ozone also can aggravate other medical conditions such as bronchitis, heart disease, emphysema, and asthma, and can reduce lung capacity. People with pre-existing respiratory ailments are especially prone to the effects of ozone. For example, asthmatics affected by ozone may have more frequent or severe attacks during periods when ozone levels are high. Children are at special risk for ozone-related problems. They breathe more air per pound of body weight than adults, and ozone can impact the development of their immature respiratory systems. They tend to be active outdoors during the summer when ozone levels are at their highest. Anyone who spends time outdoors in the summer can be affected, and studies have shown that even healthy adults can experience difficulty in breathing when exposed to ozone. Anyone engaged in strenuous outdoor activities, such as jogging, should limit activity to the early morning or late evening hours on days when ozone levels are expected to be high.

Ground-level ozone damages plant life, and a recent study (see below) estimated that it is responsible for about \$1 billion in reduced crop yield in the U.S. each year. It interferes with the ability of plants to produce and store food, making them more susceptible to harsh weather, disease, insects, and other pollutants. It damages the foliage of trees and other plants, sometimes marring the landscape of cities, national parks and forests, and recreation areas. For more information see: <https://coe.northeastern.edu/news/research-reveals-air-pollution-costs-us-estimated-1b-a-year-in-perennial-crop-yield/>.

## AMBIENT AIR QUALITY STANDARDS

National and state air quality standards for ground-level ozone were first promulgated in 1971. There are both primary standards, which are set to provide public health protection (including protecting the health of sensitive populations such as asthmatics, children, and the elderly), and secondary standards, which are based on welfare effects (such as damage to trees, crops and materials). For ground-level ozone, the primary and secondary National Ambient Air Quality Standards (NAAQS) are the same (see Table 4-1). The USEPA must periodically review the NAAQS to determine if they are sufficiently protective of public health based on the latest studies. Initially, the ozone NAAQS was an hourly average of 0.12 ppm, established in 1979. It has since been revoked by USEPA, although New Jersey retains it as a primary state standard. In 1997, the 0.08 parts per million (ppm) ozone NAAQS was promulgated, based on the maximum 8-hour average daily concentration. It was changed to 0.075 ppm in 2008. In October 2015 the 8-hour ozone NAAQS was lowered once again, to 0.070 ppm, effective in 2016.

Compliance with a NAAQS is based on meeting the design value, the actual statistic that determines whether the standard is being met. For ozone, calculating the design value is a two-step process using data from the most recent three years. The first step involves determining the fourth-highest daily maximum 8-hour average concentration for each monitoring site in the state for each of the three years. The values for each site are then used to calculate a three-year average. If this value exceeds the NAAQS at any site in the state, the state is determined to be in nonattainment.

**Table 4-1**  
**National and New Jersey Ambient Air Quality Standards for Ozone**  
Parts per Million (ppm)

Averaging Period	Type	National Level	New Jersey Level	Design Value
1-Hour	Primary	---	0.12 ppm	Annual 2 <sup>nd</sup> -highest daily maximum
8-Hours	Primary & secondary	0.070 ppm	---	3-year average of the annual 4 <sup>th</sup> -highest daily maximums

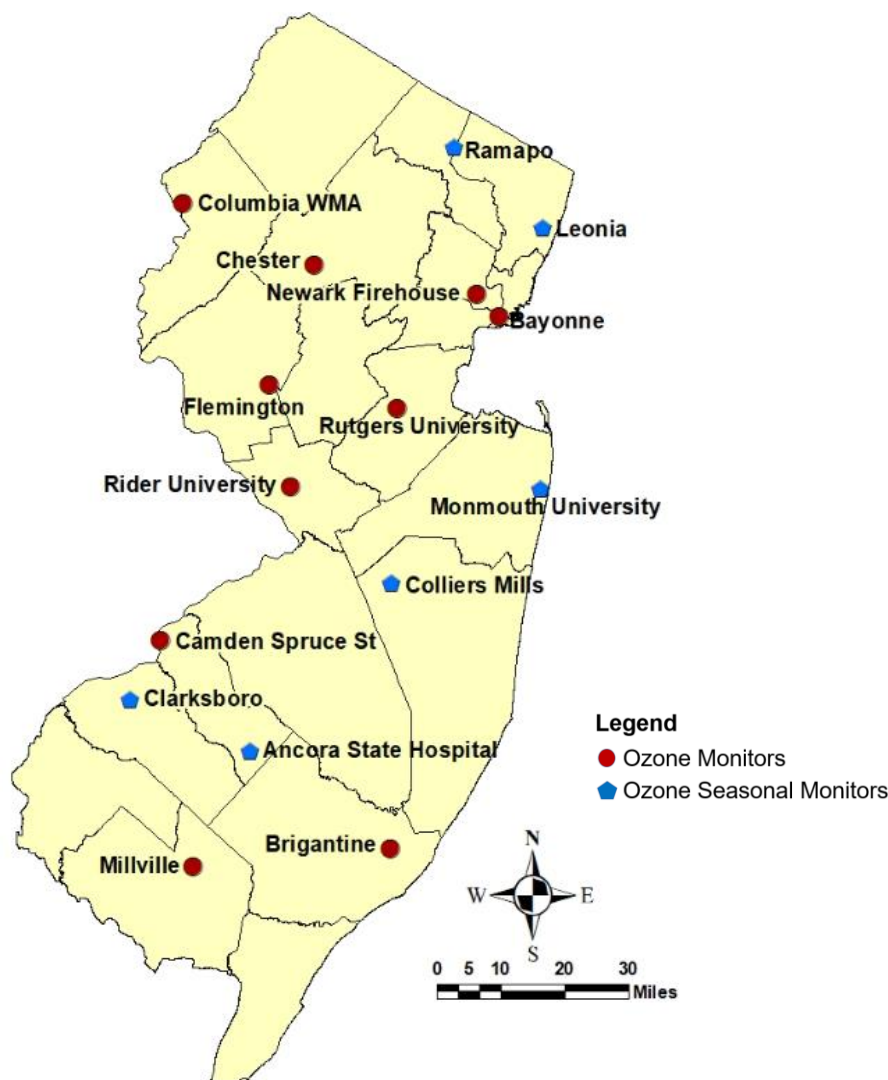


## OZONE MONITORING NETWORK

The New Jersey Department of Environmental Protection operated 16 monitoring stations in New Jersey during 2022 (see Figure 4-5). Of those 16 sites, ten operate year-round and six operate only during the ozone season, which is March 1<sup>st</sup> through October 31<sup>st</sup>. Bayonne, Brigantine, Camden Spruce Street, Chester, Columbia, Flemington, Millville, Newark Firehouse, Rider University and Rutgers University operate year-round. The Ancora, Clarksboro, Colliers Mills, Leonia, Monmouth University, and Ramapo sites operate only during the ozone season. Due to unforeseen construction, the Newark Firehouse monitoring station was shut down in September of 2022 and therefore did not have data for the complete ozone season. It is currently unknown when the station will become active again.

There is an ozone monitor at Washington Crossing State Park in Mercer County which is maintained and operated by USEPA. The site is included when determining New Jersey's NAAQS compliance status, although the data is not presented here. It can be obtained from USEPA at <https://www.epa.gov/outdoor-air-quality-data>.

**Figure 4-5**  
**2022 Ozone Monitoring Network**



## OZONE LEVELS IN 2022

The 2022 ozone season had eight days on which the NAAQS (8-hour daily maximum average concentration of 0.070 ppm) was exceeded, as shown in Table 4-2. Seven monitoring sites recorded levels above the standard at least once. Colliers Mills and Rider University had the most exceedances (three), followed by Bayonne, and Monmouth University (two), and Ancora and Rutgers University (one). Monmouth University was the only site to exceed the “Unhealthy” level in 2022. Brigantine, Camden, Chester, Columbia, Flemington, Leonia, Millville, Newark University, and Ramapo had no exceedances in 2022. For details, see the Air Quality Index section of this air quality report.

**Table 4-2**  
**2022 Exceedances of the O<sub>3</sub> NAAQS**

Day	Date	Site	8-Hour Maximum Average Concentration (ppm)	AQI Rating
1	6/4/2022	Clarksboro	0.071	USG
2	6/30/2022	Ancora	0.073	USG
		Clarksboro	0.078	USG
		Colliers Mills	0.072	USG
3	7/11/2022	Rider University	0.071	USG
4	7/14/2022	Monmouth University	0.090	U
5	7/20/2022	Bayonne	0.075	USG
		Colliers Mills	0.071	USG
		Rider University	0.071	USG
6	7/22/2022	Colliers Mills	0.071	USG
7	7/23/2022	Bayonne	0.072	USG
		Clarksboro	0.075	USG
		Monmouth University	0.073	USG
8	7/27/2022	Rider University	0.076	USG
		Rutgers University	0.071	USG

AQI = Air Quality Index

USG = Unhealthy for Sensitive Groups

U = Unhealthy

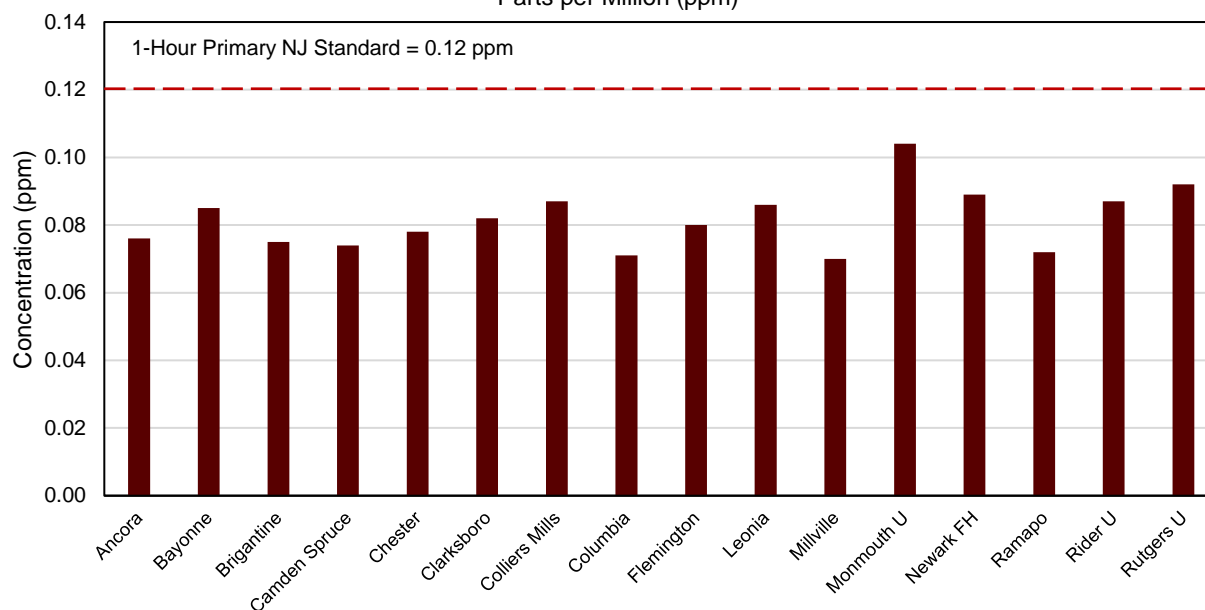
Table 4-3 presents the 2022 ozone data for the sixteen monitoring sites operated by NJDEP.

**Table 4-3**  
**2022 Ozone Concentrations in New Jersey**  
 Parts per Million (ppm)

Monitoring Site	1-Hour Daily Maximum	8-Hour Averages		
		Highest Daily Maximum	4th-Highest Daily Maximum	2019-2022 Average of 4th-Highest Daily Max.
Ancora	0.076	0.073	0.062	0.061
Bayonne	0.085	0.075	0.065	0.066
Brigantine	0.075	0.065	0.060	0.059
Camden Spruce St.	0.074	0.069	0.062	0.064
Chester	0.078	0.065	0.062	0.062
Clarksboro	0.082	0.078	0.069	0.066
Colliers Mills	0.087	0.072	0.069	0.066
Columbia	0.071	0.061	0.060	0.058
Flemington	0.080	0.068	0.063	0.062
Leonia	0.086	0.067	0.063	0.068
Millville	0.070	0.065	0.061	0.063
Monmouth University	0.104	0.090	0.069	0.067
Newark Firehouse	0.089	0.069	0.063	0.064
Ramapo	0.072	0.062	0.058	0.060
Rider University	0.087	0.076	0.068	0.069
Rutgers University	0.092	0.071	0.068	0.068

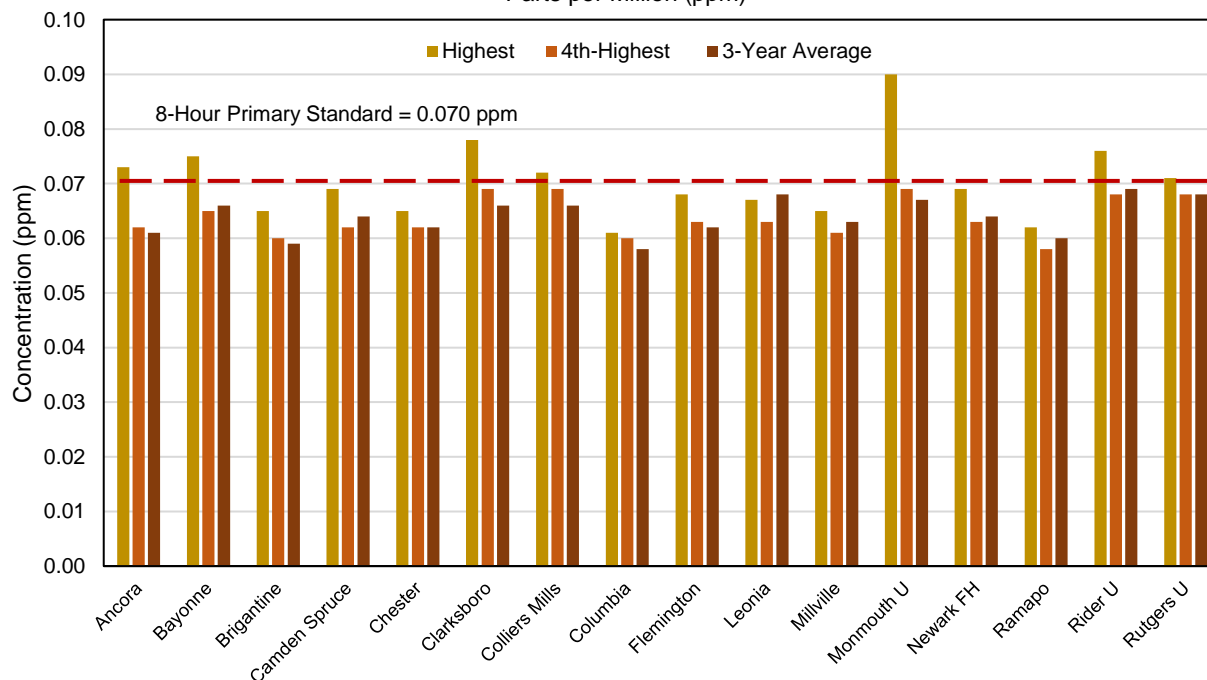
No site recorded levels above the New Jersey 1-hour standard of 0.12 ppm. The highest daily 1-hour concentration was 0.104 ppm, recorded at Monmouth University on July 14. The 1-hour standard was most recently exceeded in 2018. Figure 4-6 shows the maximum one-hour data for each site.

**Figure 4-6**  
**2022 Ozone Concentrations in New Jersey**  
**1-Hour Daily Maximums**  
 Parts per Million (ppm)



The highest daily maximum 8-hour average concentration was 0.090 ppm, recorded at Monmouth University on July 26. The 4<sup>th</sup>-highest 8-hour daily maximum value (0.069 ppm) was measured at Clarksboro, Colliers, and Monmouth University. None of the monitoring sites exceeded the design value (0.070) during the 2022 ozone season (3-year average of the 4<sup>th</sup>-highest 8-hour daily maximum). Figure 4-7 shows the 8-hour values for each site.

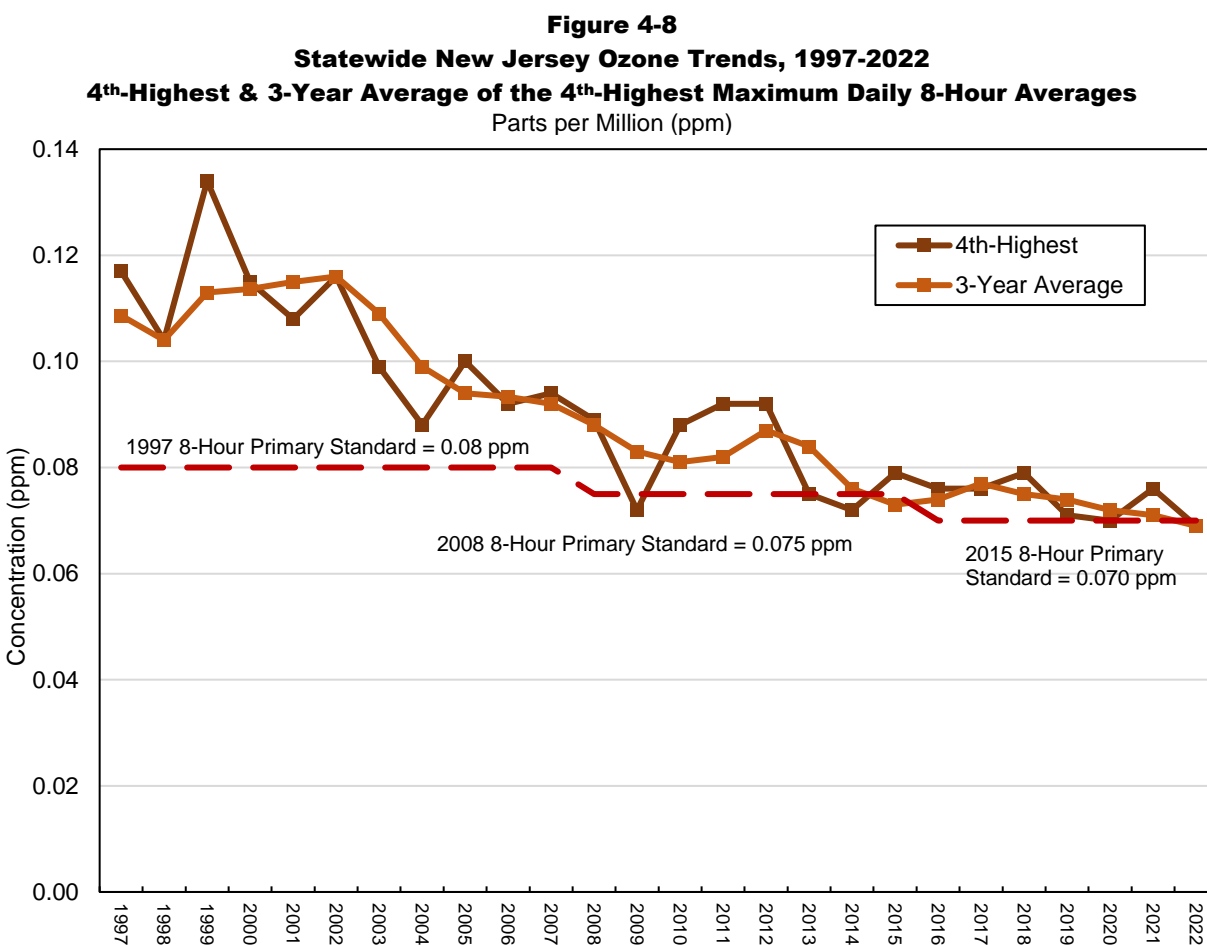
**Figure 4-7**  
**2022 Ozone Concentrations in New Jersey**  
**8-Hour Daily Maximums (Highest, 4<sup>th</sup>-Highest, and 3-Year Average of 4<sup>th</sup> Highest)**  
 Parts per Million (ppm)



## OZONE TRENDS

Studies have shown that to decrease ground-level ozone concentrations, emissions of VOCs and NO<sub>x</sub> must be reduced. Over the past couple of decades, emissions reductions have resulted in a relatively steady lowering of ozone levels in New Jersey. The chart in Figure 4-8 shows the fourth-highest 8-hour maximum daily average concentration and the ozone design value (which is a 3-year average of these values) recorded anywhere in the whole state each year since 1997. In 2022, the 4<sup>th</sup>-highest value of 0.069 ppm was measured at three sites, Clarksboro, Colliers Mills, and Monmouth University. Rider University had the highest design value of 0.069 ppm. This is lower than the NAAQS of 0.070 ppm. The yearly statewide design values can also be found in Table 4-4.

Figure 4-9 shows the total number of ozone exceedance days in New Jersey since 2000.

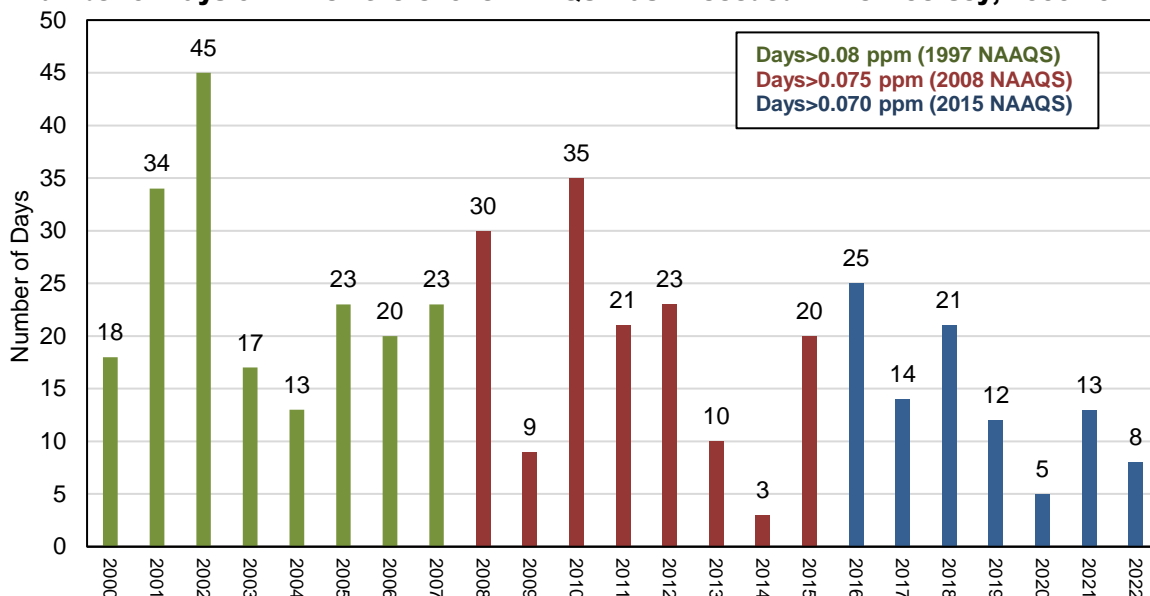


**Table 4-4**  
**Statewide New Jersey Ozone Trends, 1997-2022**  
**Maximum Daily 8-Hour Average Concentrations**  
Parts per Million (ppm)

Year	4th-Highest	3-Year Average of 4 <sup>th</sup> -Highest*
1997	0.117	0.109
1998	0.104	0.104
1999	0.134	0.113
2000	0.115	0.114
2001	0.108	0.115
2002	0.116	0.116
2003	0.099	0.109
2004	0.088	0.099
2005	0.100	0.094
2006	0.092	0.093
2007	0.094	0.092
2008	0.089	0.088
2009	0.072	0.083
2010	0.088	0.081
2011	0.092	0.082
2012	0.092	0.087
2013	0.075	0.084
2014	0.072	0.076
2015	0.079	0.073
2016	0.076	0.074
2017	0.076	0.077
2018	0.079	0.075
2019	0.071	0.074
2020	0.070	0.072
2021	0.076	0.071
2022	0.069	0.070

\*Design value

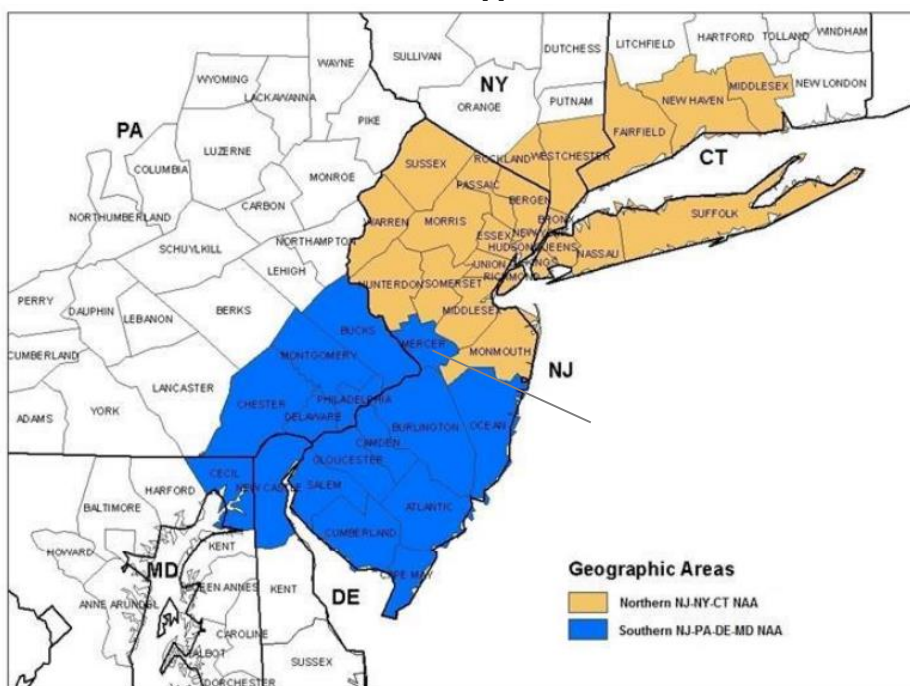
**Figure 4-9**  
**Number of Days on Which the Ozone NAAQS was Exceeded in New Jersey, 2000-2022**



## OZONE NONATTAINMENT AREAS IN NEW JERSEY

The Clean Air Act requires that all areas of the country be evaluated for attainment or nonattainment of each of the NAAQS. The entire state of New Jersey is designated as nonattainment for the ozone NAAQS, but in two separate sections that extend into neighboring states. New Jersey's nonattainment areas for the 2008 (0.075 ppm) and 2015 (0.070 ppm) 8-hour standards are shown in Figure 4-10.

**Figure 4-10**  
**New Jersey 8-Hour Ozone Nonattainment Areas (NAA)**  
**0.075 & 0.070 ppm NAAQS**



Source: NJDEP. State Implementation Plan (SIP) Revision for the Attainment and Maintenance of the Ozone NAAQS

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# 2022 Particulate Matter Summary

New Jersey Department of Environmental Protection

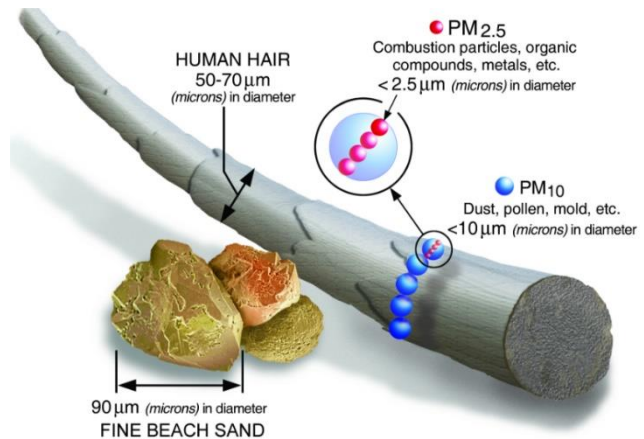
## SOURCES

Particulate air pollution is a complex mixture of organic and inorganic substances in the atmosphere, occurring as either liquids or solids. Particulates may be as large as 70 microns in diameter or smaller than 1 micron in diameter. Most particulates are small enough that individual particles are undetected by the human eye. Particulates may travel hundreds of miles from their original sources, suspended in the atmosphere, before falling to the ground.

Particulate pollution is categorized by size, measured in microns (one millionth of a meter, also known as a micrometer). Particulates with diameters of 2.5 microns or less are considered “fine particulate matter,” referred to as PM<sub>2.5</sub> (Figure 5-1). Particulates with diameters of 10 microns or less are “inhalable particulate matter,” and are referred to as PM<sub>10</sub>. “Total suspended particulate” (TSP) refers to all suspended particulates, including the largest ones.

Particulates can occur naturally or can be man-made. Examples of naturally-occurring particles are windblown dust and sea salt. Man-made particulates, which come from sources such as fossil fuel combustion and industrial processes, can be categorized as either primary particulates or secondary particulates. Primary particulates are directly emitted from their sources, while secondary particulates form in the atmosphere through reactions of gaseous emissions.

**Figure 5-1**  
**Size Comparisons for PM**



USEPA. [www.epa.gov/pm-pollution/particulate-matter-pm-basics#PM](http://www.epa.gov/pm-pollution/particulate-matter-pm-basics#PM)

## HEALTH AND ENVIRONMENTAL EFFECTS

The size of particles is directly linked to their potential for causing health problems. Fine particles (PM<sub>2.5</sub>) pose the greatest health risk.

They can get deep into the lungs and some may even get into the bloodstream. Exposure to these particles can affect a person's lungs and heart. They can lead to premature death in people with heart or lung disease, can cause heart attacks, decrease lung function, and aggravate asthma. PM<sub>10</sub> is of less concern, although it is inhalable and can irritate a person's eyes, nose, and throat.

Particulates of all sizes have an impact on the environment. PM is the major cause of reduced visibility in many parts of the United States. The national Regional Haze Program seeks to improve and protect visibility and air quality in specially designated areas such as national parks and wilderness areas. New Jersey has one Class I area listed under the program, the Brigantine wilderness area, located within the Edwin B. Forsythe National Wildlife Refuge in Atlantic County. A visibility camera ([www.hazecam.net](http://www.hazecam.net)) is located there. Airborne particles can also impact vegetation and aquatic ecosystems, and can cause damage to paints and building materials.

## AMBIENT AIR QUALITY STANDARDS

The U.S. Environmental Protection Agency (USEPA) first established National Ambient Air Quality Standards (NAAQS) for particulate matter in 1971. It set primary (health-based) and secondary (welfare-based) standards for total suspended particulate (TSP), which included PM up to about 25 to 45 micrometers. Over the years, new health data shifted the focus toward smaller and smaller particles. In 1987, USEPA replaced the TSP standards with standards for PM<sub>10</sub>. The 24-hour PM<sub>10</sub> primary and secondary standards were set at 150 µg/m<sup>3</sup>, and an annual standard was set at 50 µg/m<sup>3</sup> (it was revoked in 2010). In 1997, USEPA began regulating PM<sub>2.5</sub>. The annual PM<sub>2.5</sub> primary and secondary standards were set at 15.0 µg/m<sup>3</sup> until 2013, when the primary annual standard was lowered to 12.0 µg/m<sup>3</sup>. A 24-hour PM<sub>2.5</sub> standard of 65 µg/m<sup>3</sup> was promulgated in 1997, then lowered in 2006 to 35 µg/m<sup>3</sup>. Table 5-1 provides a summary of the current particulate matter standards.

Compliance with the standards is determined by calculating a statistic called the design value. For the annual PM<sub>2.5</sub> NAAQS, the design value is the highest statewide 3-year average of each site's annual average concentrations. For the 24-hour NAAQS, the 98th percentile of the 24-hour concentrations for each monitoring site must be averaged for the three most recent years. The highest site's value is the state's design value. For PM<sub>10</sub>, the design value is the second-highest 24-hour average concentration for a given year.

**Table 5-1**  
**National Ambient Air Quality Standards for Particulate Matter**  
Micrograms per Cubic Meter (µg/m<sup>3</sup>)

Pollutant	Averaging Period	Type	Level	Design Value
Fine Particulate (PM <sub>2.5</sub> )	Annual	Primary	12.0 µg/m <sup>3</sup>	3-year average of the annual means
	Annual	Secondary	15.0 µg/m <sup>3</sup>	3-year average of the annual means
	24-Hours	Primary & Secondary	35 µg/m <sup>3</sup>	3-year average of the annual 98 <sup>th</sup> percentile values
Inhalable Particulate (PM <sub>10</sub> )	24-Hours	Primary & Secondary	150 µg/m <sup>3</sup>	2 <sup>nd</sup> -highest annual average over 3 years

## PARTICULATE MONITORING NETWORK

Criteria pollutant monitors must meet strict USEPA requirements to determine compliance with the NAAQS. To measure ambient particulate matter, the New Jersey Department of Environmental Protection (NJDEP) uses two different approaches: a filter-based method, and continuous beta attenuation.

Filter-based samplers pull a predetermined amount of air through PM<sub>2.5</sub> or PM<sub>10</sub> size-selective inlets for a 24-hour period. The filters are weighed before and after sampling under controlled environmental conditions to determine the concentration of the captured particles. This filter-based method has for years been designated as the Federal Reference Method (FRM) for particulate matter compliance determination. It requires daily to weekly visits to pick up and replace filters.

New Jersey also uses Beta Attenuation Monitors (BAM), which measure the loss of intensity (attenuation) of beta particles due to absorption by PM<sub>2.5</sub> particles collected on a filter tape. These monitors are classified by USEPA as Federal Equivalent Methods (FEM) for PM<sub>2.5</sub>, and can also be used to determine compliance with the NAAQS. These monitors provide real-time hourly PM data, which is available to the public on the NJDEP air monitoring website (<https://nj.gov/dep/airmon/>).

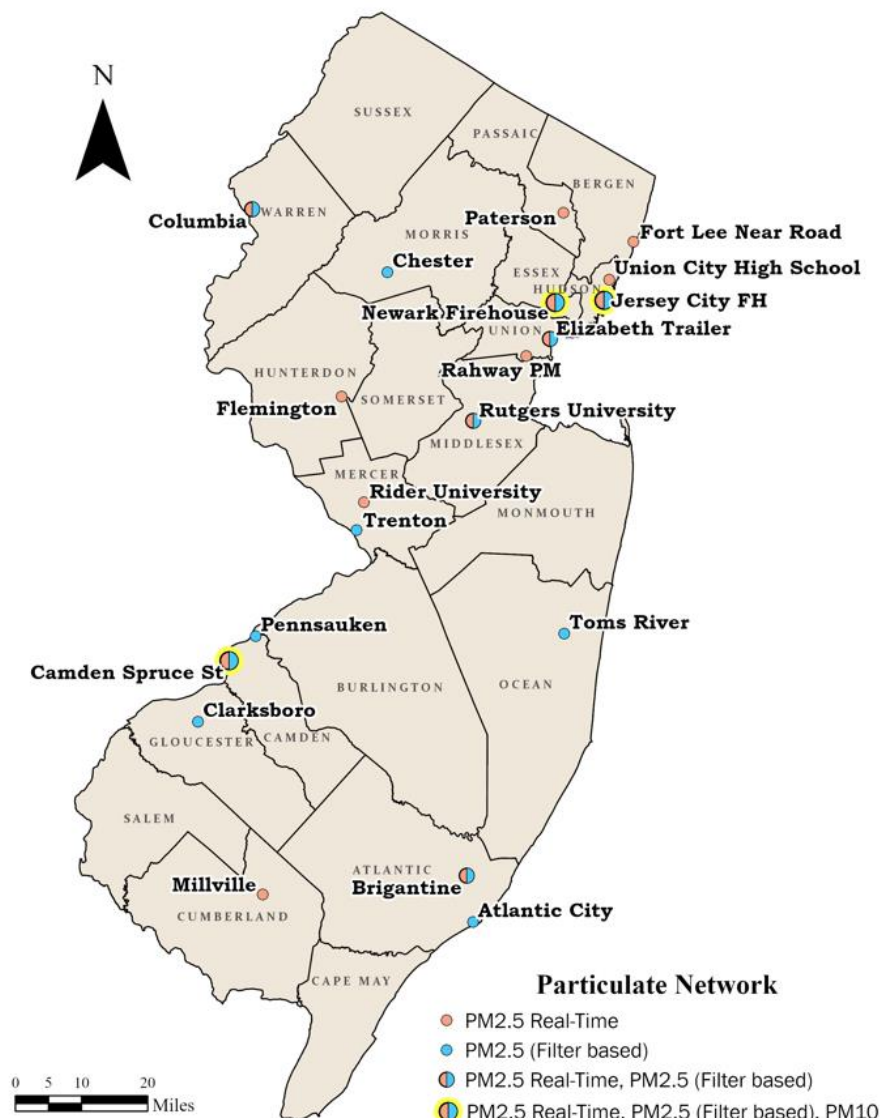
For 2022, NJDEP had twenty PM<sub>2.5</sub> monitoring sites around the state. There were ten filter-based monitors and 16 continuous monitors. Six sites have both types of particulate monitors, including the Newark Firehouse monitoring station. However, it was shut down in September because of construction at the site. A new monitoring location in Newark is currently being sought.

At one time, NJDEP had more than twenty inhalable particulate (PM<sub>10</sub>) sampling sites. After many years of low concentrations and the shift in emphasis to PM<sub>2.5</sub> monitoring, three sites remained for most of 2022, until the Newark station stopped operations in September. PM<sub>10</sub> samples are taken once every six days at Camden and Jersey City, and were taken every three days at Newark.

Five monitoring stations were part of the national Chemical Speciation Network (CSN), which analyzes ambient particles to determine their composition. CSN monitoring took place at the Camden Spruce Street, Chester, Elizabeth Lab, Newark Firehouse and Rutgers University monitoring stations. More information about New Jersey's 2022 CSN data, including a map of locations, can be found in Appendix B of the Air Quality Report.

Figure 5-2 shows the locations of all the particulate monitors in New Jersey.

**Figure 5-2**  
**2022 Particulate Monitoring Network**



## FINE PARTICLE (PM<sub>2.5</sub>) LEVELS IN 2022

### PM<sub>2.5</sub> LEVELS FOR FILTER-BASED MONITORS

In 2022, none of the filter-based FRM PM<sub>2.5</sub> monitoring sites exceeded either the annual (12.0 µg/m<sup>3</sup>) or the 24-hour (35 µg/m<sup>3</sup>) NAAQS.

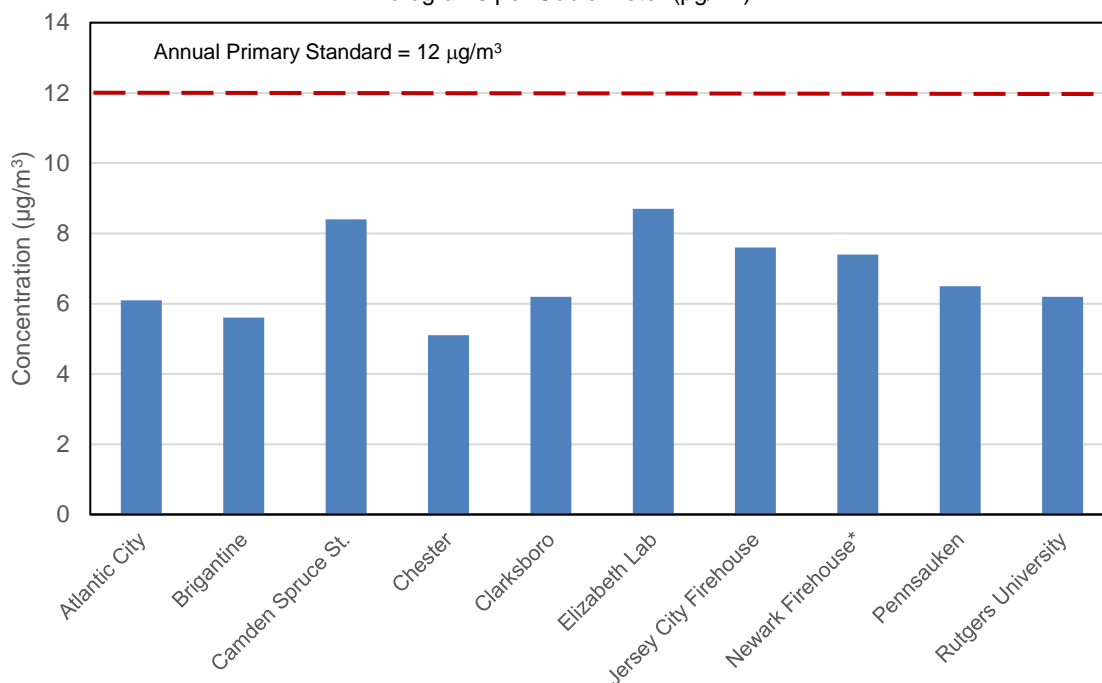
The annual mean concentrations of PM<sub>2.5</sub> measured at the ten filter-based samplers ranged from 5.1 µg/m<sup>3</sup> at the Chester monitoring site, to 8.7 µg/m<sup>3</sup> at the Elizabeth Lab monitoring station. The highest 24-hour concentrations ranged from 12.5 µg/m<sup>3</sup> at Brigantine to 28.6 µg/m<sup>3</sup> at the Jersey City Firehouse. Table 5-2 shows the annual mean, and highest and 98<sup>th</sup>-percentile 24-hour concentrations, as well as the number of valid samples collected. The data is also shown graphically in Figures 5-3 and 5-4. Two sites (Elizabeth Lab, Jersey City Firehouse) sample PM<sub>2.5</sub> every day. The other eight sites (Atlantic City, Brigantine, Camden Spruce Street, Chester, Clarksboro, Newark Firehouse, Pennsauken and Rutgers University) take a sample every third day, resulting in about 122 samples per year.

**Table 5-2**  
**2022 PM<sub>2.5</sub> Concentrations in New Jersey**  
**Annual and 24-Hour Averages for Filter-Based Monitors**  
Micrograms per Cubic Meter (µg/m<sup>3</sup>)

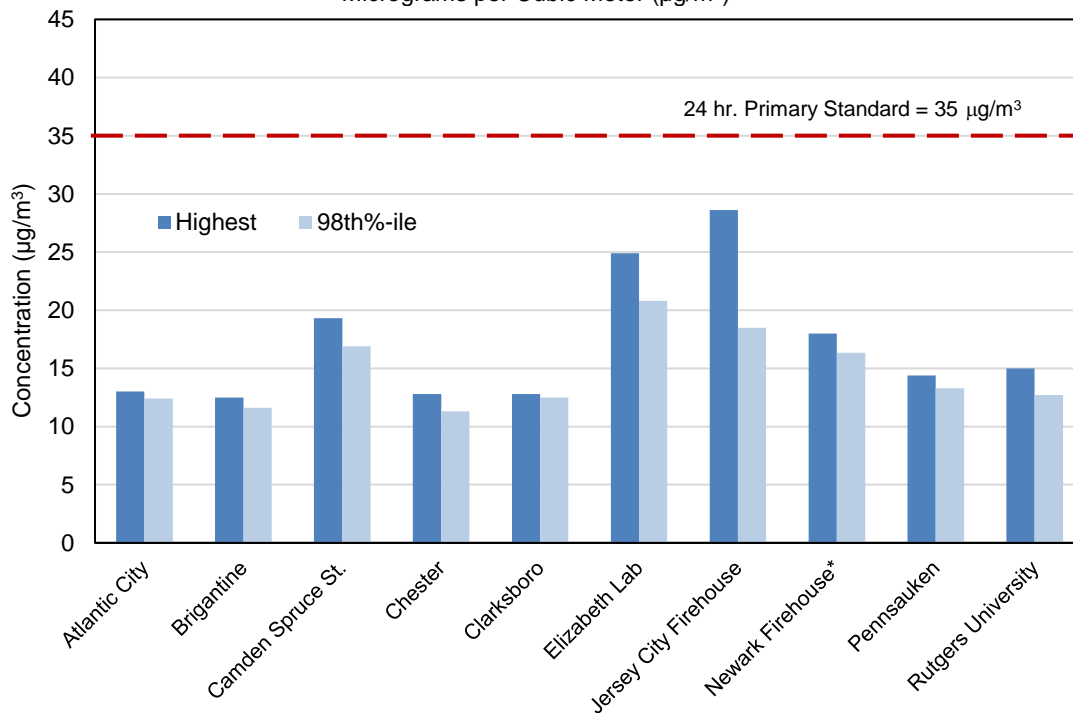
Monitoring Site	Number of Samples	Annual Average	24-Hour Average	
			Highest	98 <sup>th</sup> -ile
Atlantic City	114	6.1	13	12.4
Brigantine	111	5.6	12.5	11.6
Camden Spruce St.	113	8.4	19.3	16.9
Chester	109	5.1	12.8	11.3
Clarksboro	117	6.2	12.8	12.5
Elizabeth Lab	345	8.7	24.9	20.8
Jersey City Firehouse	323	7.6	28.6	18.5
Newark Firehouse*	85	7.4	18	16.3
Pennsauken	113	6.5	14.4	13.3
Rutgers University	121	6.2	15	12.7

\* Newark Firehouse ceased operations on 9/26/2022.

**Figure 5-3**  
**2022 PM<sub>2.5</sub> Concentrations in New Jersey**  
**Annual Averages for Filter-Based Monitors**  
 Micrograms per Cubic Meter ( $\mu\text{g}/\text{m}^3$ )



**Figure 5-4**  
**2022 PM<sub>2.5</sub> Concentrations in New Jersey**  
**24-Hour Averages for Filter-Based Monitors**  
 Micrograms per Cubic Meter ( $\mu\text{g}/\text{m}^3$ )



\*Newark Firehouse was shut down 9/26/2022.

## PM<sub>2.5</sub> LEVELS FOR CONTINUOUS MONITORS

Two more continuous particulate monitors were added to the New Jersey air monitoring network in 2022, for a total of sixteen, at Brigantine, Camden Spruce Street, Columbia, Elizabeth Lab, Flemington, Fort Lee Near Road, Jersey City Firehouse, Millville, Newark Firehouse, Paterson, Rahway, Rider University, Rutgers University, Toms River, Trenton and Union City. Near real-time PM<sub>2.5</sub> readings are transmitted to a central computer in Trenton, where they are averaged every hour and reported on the NJDEP website at <https://nj.gov/dep/airmon/>.

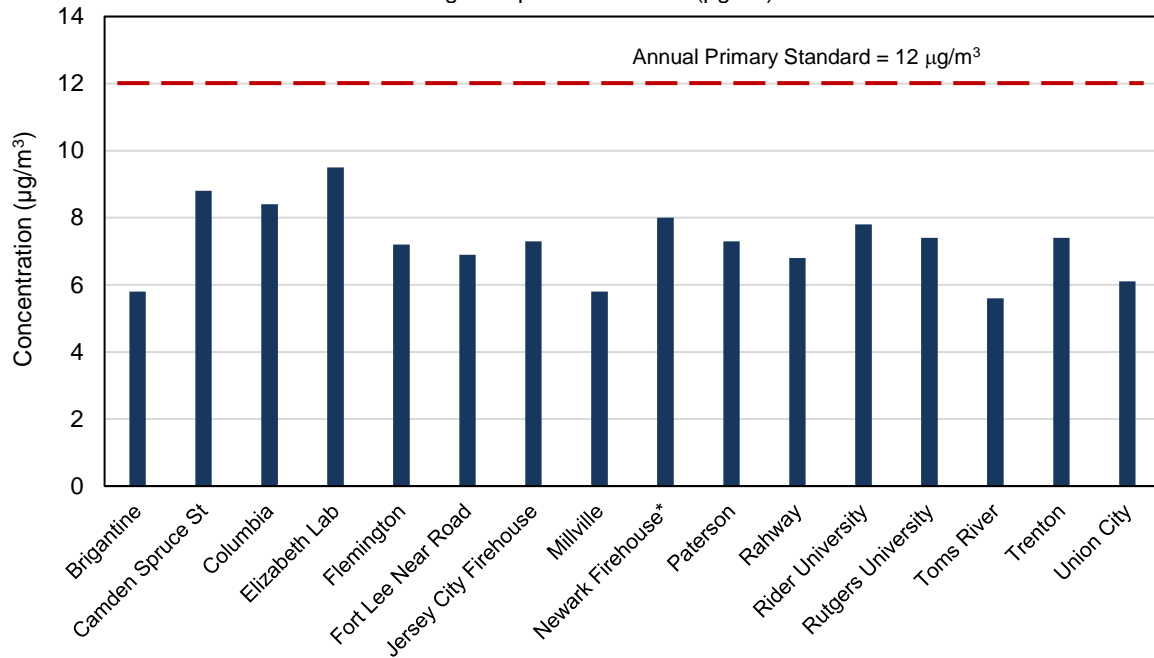
In 2022 there were no exceedances of the 12.0 µg/m<sup>3</sup> annual standard. However, there was one exceedance of the 24-hour standard at the Brigantine monitor on June 20, attributable to smoke from the Mullica River Fire in Wharton State Forest. Table 5-3 presents the annual mean, and the highest and 98<sup>th</sup> percentile 24-hour values from these sites for 2022. Figures 5-5 and 5-6 show the same data in graphs.

**Table 5-3**  
**2022 PM<sub>2.5</sub> Concentrations in New Jersey**  
**Annual and 24-Hour Averages for Continuous Monitors**  
Micrograms per Cubic Meter (µg/m<sup>3</sup>)

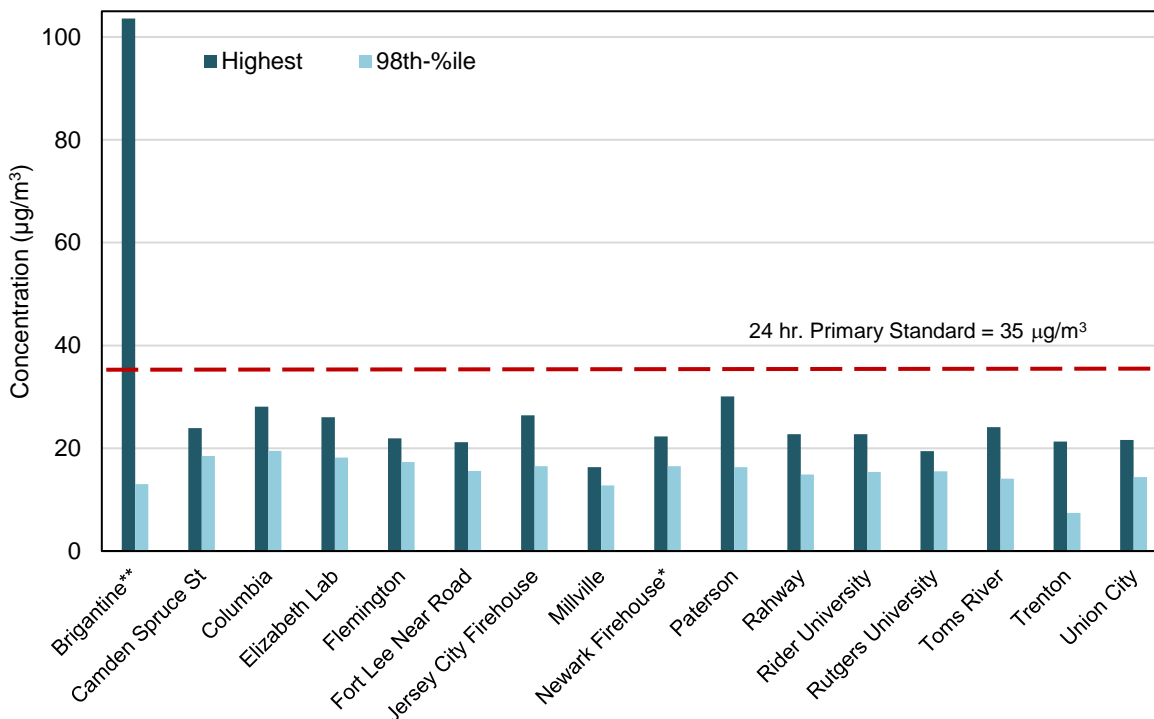
Monitoring Site	Annual Average	24-Hour Average	
		Highest	98 <sup>th</sup> -%ile
Brigantine	5.8	103.6	13.0
Camden Spruce Street	8.8	23.9	18.5
Columbia	8.4	28.1	19.5
Elizabeth Lab	9.5	26.0	18.2
Flemington	7.2	21.9	17.3
Fort Lee Near Road	6.9	21.2	15.6
Jersey City Firehouse	7.3	26.4	16.5
Millville	5.8	16.3	12.8
Newark Firehouse*	8.0	22.3	16.5
Paterson	7.3	30.1	16.3
Rahway	6.8	22.7	14.9
Rider University	7.8	22.7	15.4
Rutgers University	7.4	19.4	15.5
Toms River	5.6	24.1	14.1
Trenton	7.4	21.3	7.4
Union City	6.1	21.6	14.4

\*Newark Firehouse was shut down 9/26/2022.

**Figure 5-5**  
**2022 PM<sub>2.5</sub> Concentrations in New Jersey**  
**Annual Averages for Continuous Monitors**  
 Micrograms per Cubic Meter ( $\mu\text{g}/\text{m}^3$ )



**Figure 5-6**  
**2022 PM<sub>2.5</sub> Concentrations in New Jersey**  
**24-Hour Averages for Continuous Monitors**  
 Micrograms per Cubic Meter ( $\mu\text{g}/\text{m}^3$ )



\*Newark Firehouse was shut down 9/26/2022.

\*\*The maximum 24-hour concentration of 103.6  $\mu\text{g}/\text{m}^3$  was recorded at Brigantine on 6/20/2022 during the Mullica River Fire.

## 2022 PM<sub>2.5</sub> DESIGN VALUES

For PM<sub>2.5</sub> monitoring sites that have both a filter-based monitor and a continuous monitor, USEPA allows for the combination of data from both monitors to calculate annual design values, which require complete data sets for the three latest years.

USEPA determines the validity and completeness of a data set used for design values. Sites with incomplete data are determined to have “invalid” design values, and these are marked with an asterisk in the table and figures below. Due to limited access to many monitoring stations because of the COVID-19 pandemic in 2020, and some equipment problems in 2021, only eight out of New Jersey’s twenty PM<sub>2.5</sub> monitoring stations had the three years of complete data necessary to calculate a valid 2022 design value.

Table 5-4 and Figures 5-7 and 5-8 show USEPA’s calculated PM<sub>2.5</sub> 2022 design values for each of the New Jersey monitors. All of New Jersey’s PM<sub>2.5</sub> monitoring sites were below the annual and 24-hour design values in 2022.

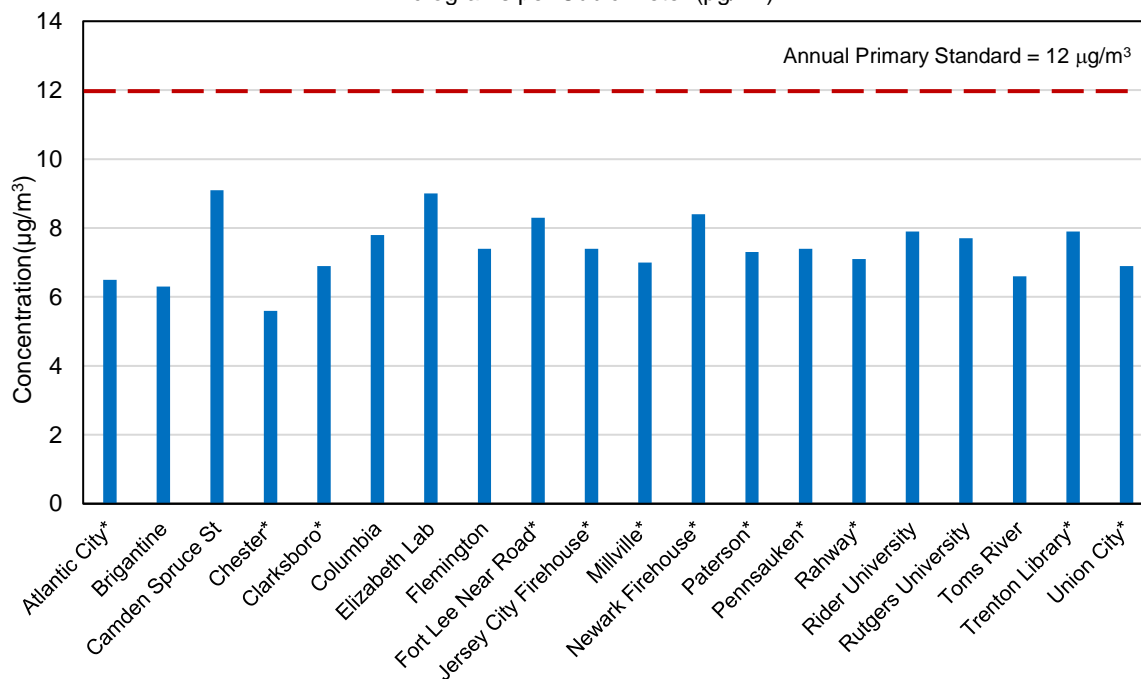
**Table 5-4**  
**New Jersey PM<sub>2.5</sub> Design Values for 2020-2022**  
**3-Year Average of the Annual Average Concentrations**  
**& 98<sup>th</sup> Percentile 24-Hour Average Concentrations**  
Micrograms per Cubic Meter (µg/m<sup>3</sup>)

Monitoring Site	3-Year (2020-2022) Average	
	Annual	98 <sup>th</sup> Percentile 24-Hour
Atlantic City*	6.5	15
Brigantine	6.3	15
Camden Spruce Street	9.1	20
Chester*	5.6	16
Clarksboro*	6.9	15
Columbia	7.8	20
Elizabeth Lab	9.0	21
Flemington	7.4	18
Fort Lee Near Road*	8.3	21
Jersey City Firehouse*	7.4	20
Millville*	7.0	16
Newark Firehouse*	8.4	20
Paterson*	7.3	16
Pennsauken*	7.4	18
Rahway*	7.1	18
Rider University	7.9	17
Rutgers University	7.7	19
Toms River	6.6	16
Trenton Library*	7.9	18
Union City*	6.9	16

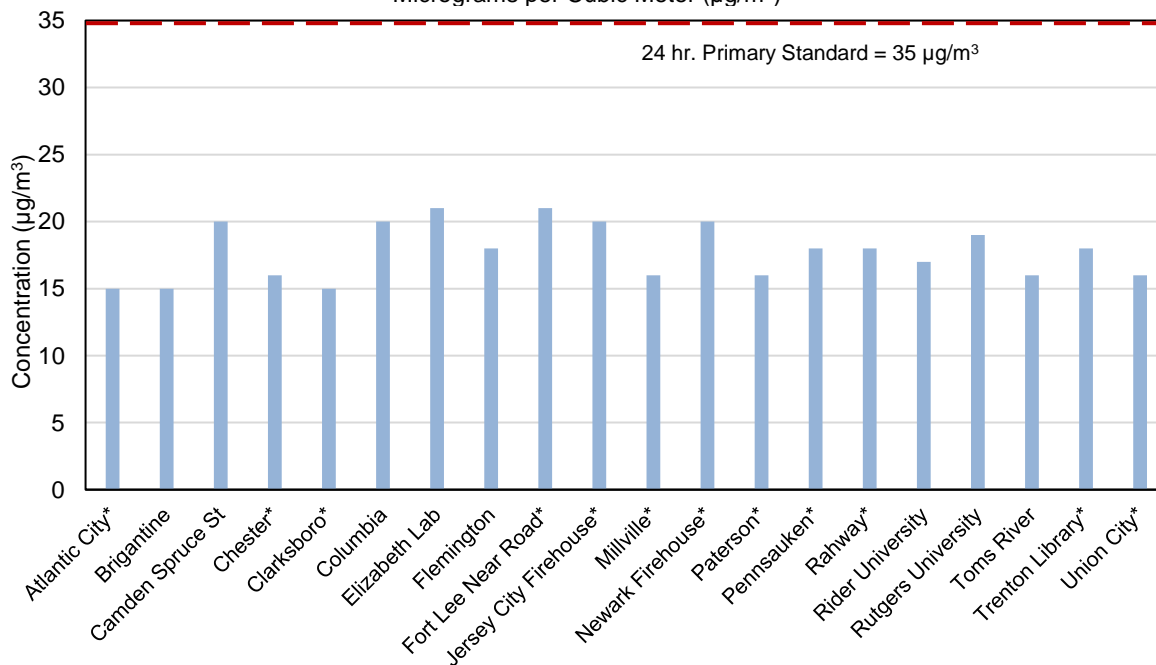
\*Design value deemed invalid by USEPA because of incomplete data set.



**Figure 5-7**  
**New Jersey PM<sub>2.5</sub> Design Values for 2020-2022**  
**3-Year Average of the Annual Average Concentrations**  
 Micrograms per Cubic Meter ( $\mu\text{g}/\text{m}^3$ )



**Figure 5-8**  
**New Jersey PM<sub>2.5</sub> Design Values for 2020-2022**  
**3-Year Average of the 98<sup>th</sup> Percentile 24-Hour Average Concentrations**  
 Micrograms per Cubic Meter ( $\mu\text{g}/\text{m}^3$ )



\*Design value deemed invalid by USEPA because of incomplete data set.

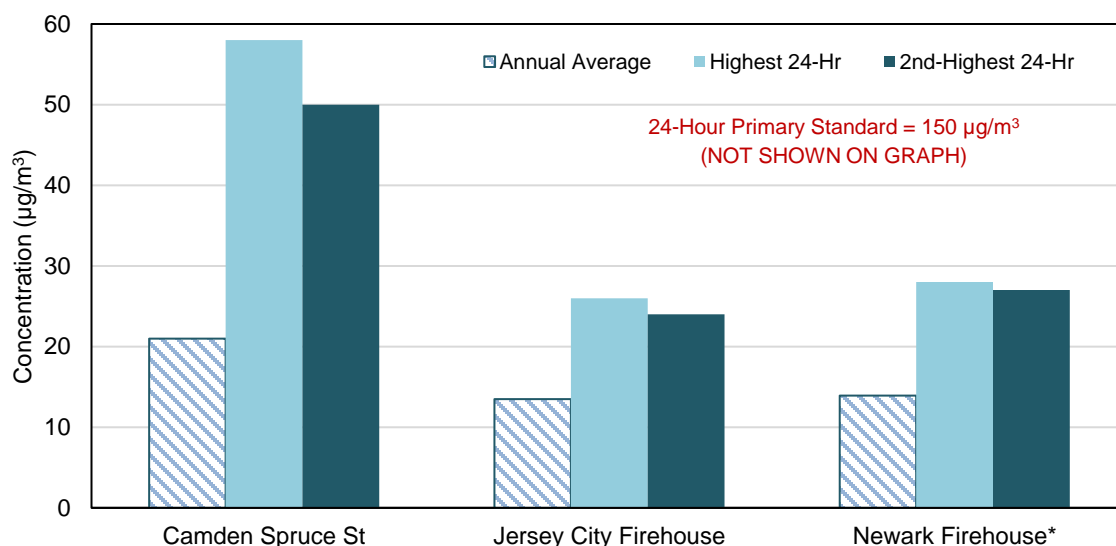
## INHALABLE PARTICULATE (PM<sub>10</sub>) LEVELS IN 2022

Table 5-5 presents 2022 data for each of the New Jersey PM<sub>10</sub> monitors. The highest and second-highest 24-hour concentrations are shown, as well as the annual averages. All areas of the state are in attainment for the 24-hour standard of 150 µg/m<sup>3</sup>, as can be seen in Figure 5-9. The standard is based on the second-highest 24-hour value.

**Table 5-5**  
**2022 PM<sub>10</sub> Concentrations in New Jersey**  
**Annual and 24-Hour Averages**  
Micrograms per Cubic Meter (µg/m<sup>3</sup>)

Monitoring Site	Number of Samples	Annual Average	24-Hour Average	
			Highest	Second-Highest
Camden Spruce Street	59	21.0	58	50
Jersey City Firehouse	59	13.5	26	24
Newark Firehouse*	85	13.9	28	27

**Figure 5-9**  
**2022 PM<sub>10</sub> Concentrations in New Jersey**  
**Annual and 24-Hour Averages**  
Micrograms per Cubic Meter (µg/m<sup>3</sup>)

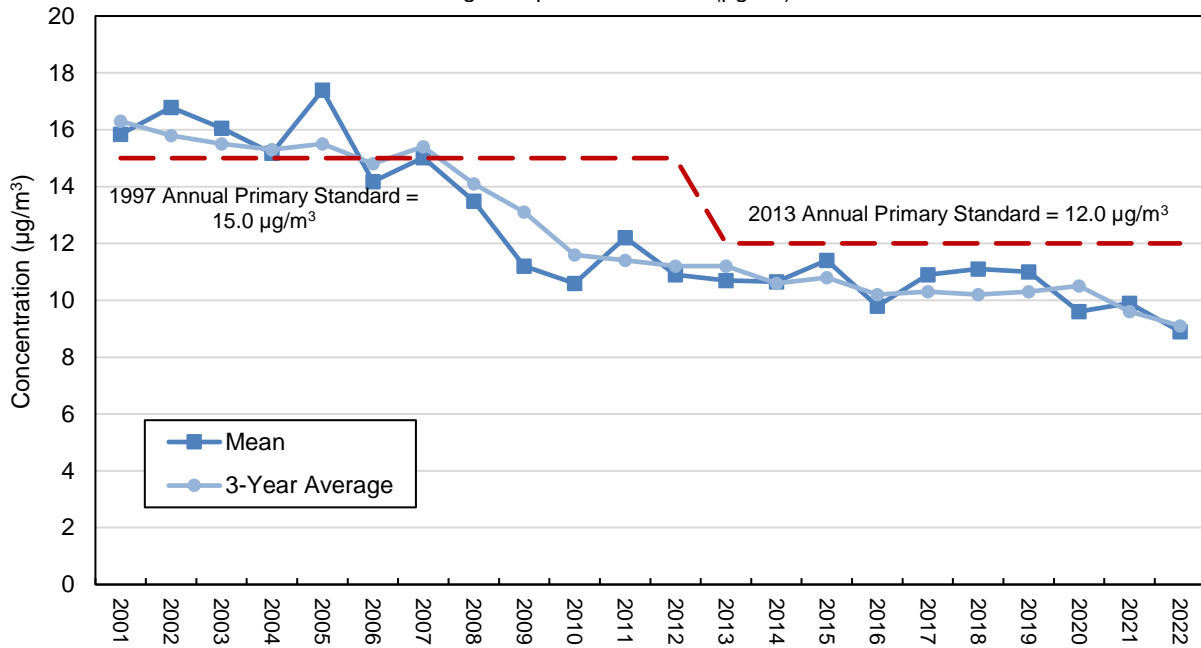


\*Newark Firehouse was shut down on 9/26/2022.

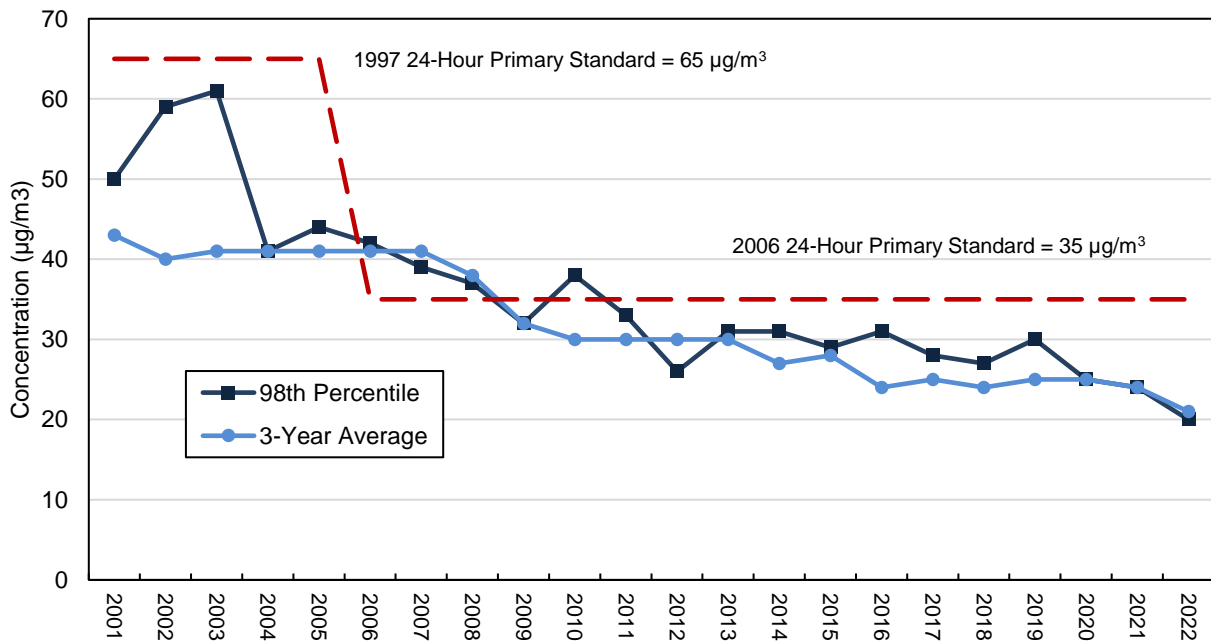
## PARTICULATE TRENDS

A PM<sub>2.5</sub> monitoring network was established in New Jersey in 1999. Using the maximum statewide annual values, Figures 5-10 and 5-11 show the highest New Jersey means and design values (3-year averages) since 2001, as well as changes to the NAAQS. The years of data show a noticeable decline in fine particulate concentrations.

**Figure 5-10**  
**Statewide New Jersey PM<sub>2.5</sub> Trends, 2001-2022**  
**Annual Mean & 3-Year Average of the Annual Mean Concentrations**  
 Micrograms per Cubic Meter ( $\mu\text{g}/\text{m}^3$ )

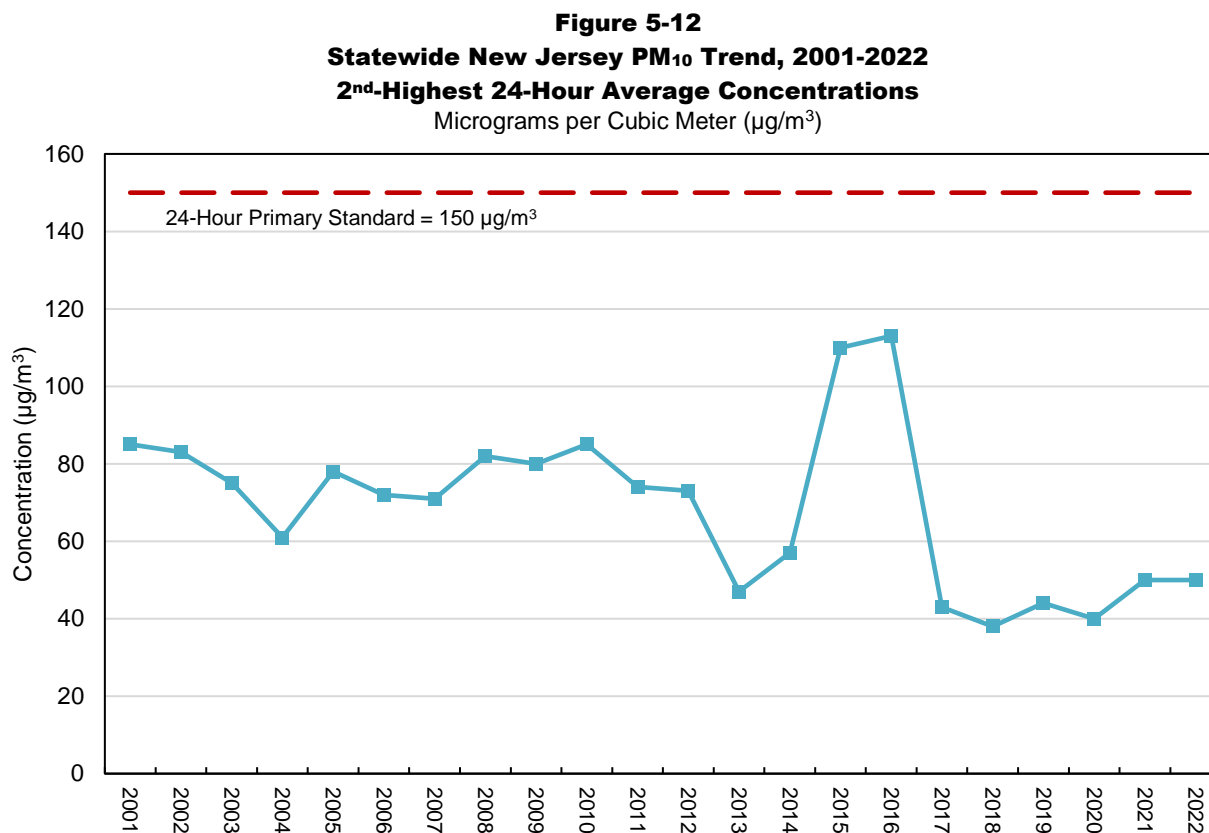


**Figure 5-11**  
**Statewide New Jersey PM<sub>2.5</sub> Trends, 2001-2022**  
**98<sup>th</sup> Percentile & 3-Year Average of the 98<sup>th</sup> Percentile 24-Hour Average Concentrations**  
 Micrograms per Cubic Meter ( $\mu\text{g}/\text{m}^3$ )



The PM<sub>10</sub> statewide design value trend is shown in Figure 5-12. The increase in concentration in 2015 and 2016 occurred at the Camden RRF monitor at 600 Morgan Street, during a period of major road reconstruction nearby. The Camden RRF site was shut down in early 2020, when a PM<sub>10</sub> monitor was placed at the Camden Spruce Street station.

Table 5-6 below presents the trend data displayed in Figures 5-10, 5-11 and 5-12.



**Table 5-6**  
**Statewide New Jersey Particulate Matter Trends, 2001-2022**  
**PM<sub>2.5</sub> & PM<sub>10</sub> Concentrations**  
Micrograms per Cubic Meter (µg/m<sup>3</sup>)

Year	PM <sub>2.5</sub>				PM <sub>10</sub>
	Annual		24-Hour		24-Hour
	Mean	3-Year Average*	98th Percentile	3-Year Average*	2nd-Highest*
2001	15.8	16.3	50	43	85
2002	16.8	15.8	59	40	83
2003	16.1	15.5	61	41	75
2004	15.2	15.3	41	41	61
2005	17.4	15.5	44	41	78
2006	14.2	14.8	42	41	72
2007	15.0	15.4	39	41	71
2008	13.5	14.1	37	38	82
2009	11.2	13.1	32	32	80
2010	10.6	11.6	38	30	85
2011	12.2	11.4	33	30	74
2012	10.9	11.2	26	30	73
2013	10.7	11.2	31	30	47
2014	10.6	10.6	31	27	57
2015	11.4	10.8	29	28	110
2016	9.8	10.2	31	24	113
2017	10.9	10.3	28	25	43
2018	11.1	10.2	27	24	38
2019	11.0	10.3	30	25	44
2020	9.6	10.5	25	25	40
2021	9.9	9.6	24	24	50
2022	8.9	9.1	20	21	50

\*Design value

## REFERENCES

Camnet. Realtime Air Pollution & Visibility Monitoring. [www.hazecam.net](http://www.hazecam.net). Accessed 7/31/23.

U.S. Environmental Protection Agency (USEPA). 2012 National Ambient air Quality Standards (NAAQS) for Particulate Matter (PM). [www.epa.gov/pm-pollution/2012-national-ambient-air-quality-standards-naaqs-particulate-matter-pm](http://www.epa.gov/pm-pollution/2012-national-ambient-air-quality-standards-naaqs-particulate-matter-pm). Accessed 7/31/23.

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USEPA. Visibility and Regional Haze. <https://www.epa.gov/visibility>. Accessed 7/31/23.



# 2022 Nitrogen Dioxide Summary

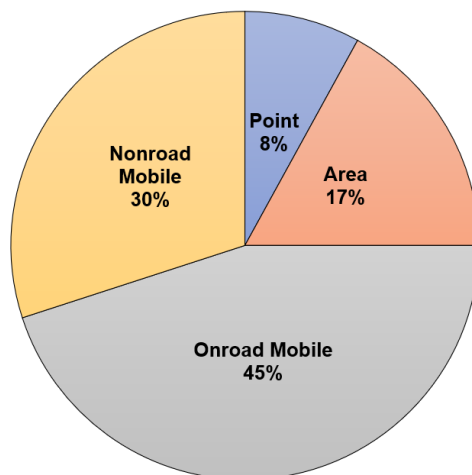
New Jersey Department of Environmental Protection

## SOURCES

Nitrogen dioxide ( $\text{NO}_2$ ) is a reddish-brown highly reactive gas that is formed in the air through the oxidation of nitric oxide ( $\text{NO}$ ).  $\text{NO}_2$  is used by regulatory agencies as the indicator for the group of gases known as nitrogen oxides ( $\text{NO}_x$ ). These gases are emitted from motor vehicle exhaust, combustion of coal, oil or natural gas, and industrial processes such as welding, electroplating, and dynamite blasting. Although most  $\text{NO}_x$  is emitted as  $\text{NO}$ , it is readily converted to  $\text{NO}_2$  in the atmosphere. In the home, gas stoves and heaters produce substantial amounts of nitrogen dioxide. When  $\text{NO}_2$  reacts with other chemicals it can form ozone, particulate matter, and other pollutant compounds. A pie chart summarizing the major sources of  $\text{NO}_x$  in New Jersey in 2017 is shown in Figure 6-1.

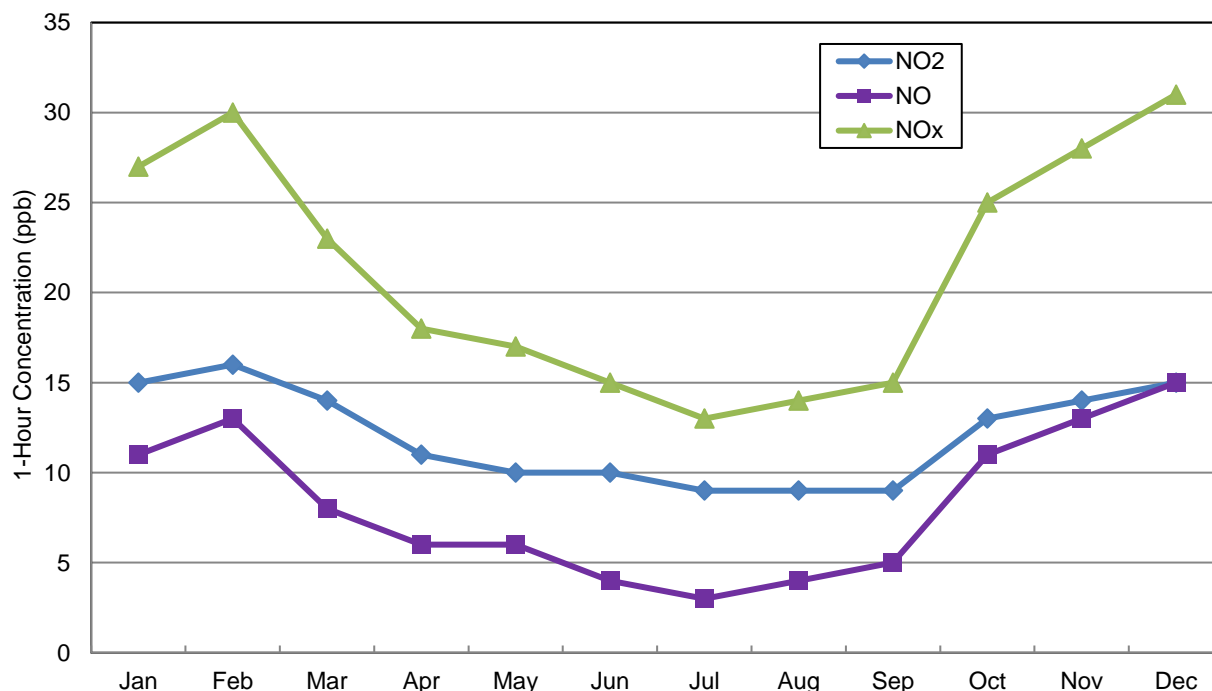
Figure 6-2 shows that  $\text{NO}_x$  concentrations tend to be higher in the winter than in the summer. This is due in part to heating of buildings, and to weather conditions that are more prevalent in the colder months of the year, such as lighter winds that result in poorer local dispersion conditions.

**Figure 6-1**  
**2017 New Jersey**  
 **$\text{NO}_x$  Annual Emissions**

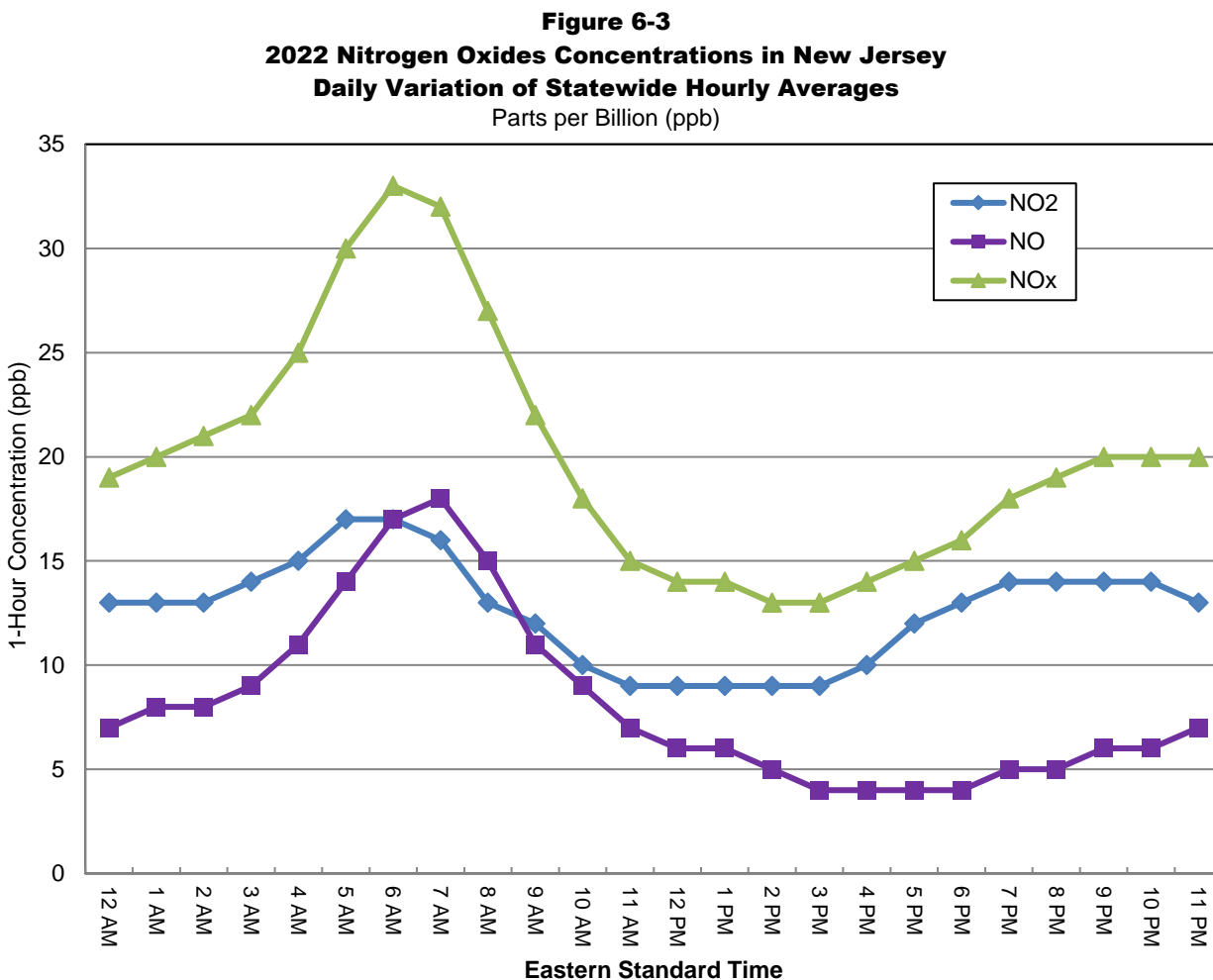


**Figure 6-2**  
**2022 Nitrogen Oxides Concentrations in New Jersey**  
**Monthly Variation of Statewide Hourly Averages**

Parts per Billion (ppb)



Because much of the NO<sub>x</sub> in the air is emitted by motor vehicles, concentrations tend to peak during the morning and afternoon rush hours. This is shown in Figure 6-3.



## HEALTH AND ENVIRONMENTAL EFFECTS

Short-term exposures to low levels of nitrogen dioxide may aggravate pre-existing respiratory illnesses and cause respiratory illnesses in children, people with asthma, and the elderly. Symptoms of low-level exposure to NO and NO<sub>2</sub> include irritation to eyes, nose, throat and lungs, coughing, shortness of breath, tiredness and nausea. Long-term exposures to NO<sub>2</sub> may increase susceptibility to respiratory infection and may cause permanent damage to the lung. Studies show a connection between breathing elevated short-term NO<sub>2</sub> concentrations and increases in hospital emergency room visits and hospital admissions for respiratory issues, especially asthma. Individuals who spend time on or near major roadways can experience elevated short-term NO<sub>2</sub> exposures.

Nitrogen oxides contribute to a wide range of environmental problems. Chemical reactions in the air form both ozone and particulate matter. Nitrate particles make the air hazy and impair visibility, and contribute to nutrient pollution in coastal waters, resulting in eutrophication. NO<sub>2</sub> also reacts with water and oxygen to form nitric acid, a component of acid rain, which causes acidification of freshwater bodies and harms sensitive ecosystems such as lakes and forests.



## AMBIENT AIR QUALITY STANDARDS

There are two types of National Ambient Air Quality Standards (NAAQS) established by the U.S. Environmental Protection Agency (USEPA), primary and secondary. Primary standards protect public health, including sensitive populations such as asthmatics, children, and the elderly. Secondary standards provide public welfare protection, including protection against decreased visibility and damage to animals, crops, vegetation, and buildings. A 1-hour primary standard for NO<sub>2</sub> of 100 parts per billion (ppb) was promulgated in 2010. The primary and secondary annual NAAQS for NO<sub>2</sub> is the same, a calendar year average concentration of 53 ppb. The annual New Jersey Ambient Air Quality Standards (NJAAQS) are identical to the NAAQS, except that micrograms per cubic meter (µg/m<sup>3</sup>) are the standard units (converted to ppm below) and the averaging time is any 12-month period (a running average) instead of a calendar year. Table 6-1 presents a summary of the NO<sub>2</sub> standards.

**Table 6-1**  
**National and New Jersey Ambient Air Quality Standards for Nitrogen Dioxide (NO<sub>2</sub>)**  
Parts per Billion (ppb)  
Parts per Million (ppm)

Averaging Period	Type	National Level	New Jersey Level	Design Value
1-Hour	Primary	100 ppb	---	3-year average of the annual 98th percentile daily maximums
Annual	Primary & secondary	53 ppb	---	Annual mean
12-Month	Primary & secondary	---	0.05 ppm	Highest 12-month running average

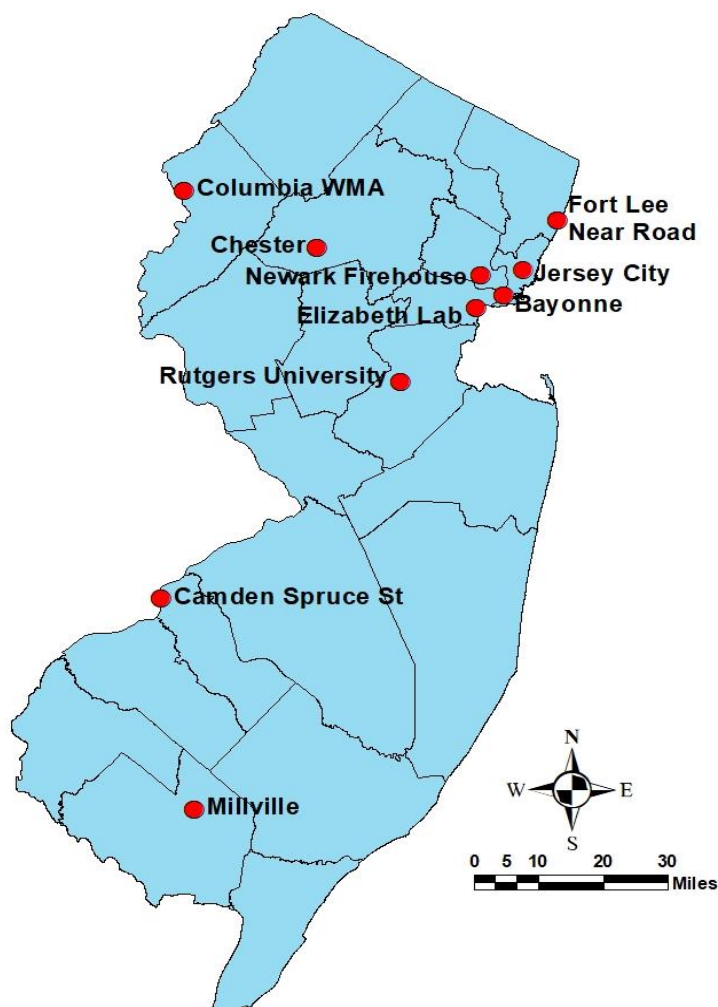
A state or other designated area is in compliance with a NAAQS when it meets the design value. For the annual standard, the annual average is the design value. However, for the 1-hour NO<sub>2</sub> standard, the NAAQS is met when the 3-year average of the 98<sup>th</sup>-percentile of the daily maximum 1-hour NO<sub>2</sub> concentrations is less than 100 ppb. This statistic is calculated by first obtaining the maximum 1-hour average NO<sub>2</sub> concentrations for each day at each monitor. Then the 98<sup>th</sup>-percentile value of the daily maximum NO<sub>2</sub> concentrations must be determined for the current year, and for each of the previous two years. Finally, the average of these three annual 98<sup>th</sup>-percentile values is the design value.

## NO<sub>2</sub> MONITORING NETWORK

NJDEP measured NO<sub>2</sub> levels at ten locations in 2022. The monitoring stations are Bayonne, Camden Spruce Street, Chester, Columbia, Elizabeth Lab, Fort Lee Near Road, Jersey City, Millville, Newark Firehouse, and Rutgers University. These sites are shown in Figure 6-4. These sites also measure NO and NO<sub>x</sub>, except for Rutgers, which measures NO and total reactive nitrogen (NO<sub>y</sub>) as required for the Photochemical Assessment Monitoring Station (PAMS) Program. NO<sub>y</sub> is also measured at Newark Firehouse, as required for an NCore Multipollutant Monitoring Network site.

However, the Newark Firehouse monitoring station was shut down on September 26<sup>th</sup> after we received one month's notice to vacate the property for construction.

**Figure 6-4**  
**2022 Nitrogen Dioxide Monitoring Network**



## **NO<sub>2</sub> LEVELS IN 2022**

There were no exceedances of any NO<sub>2</sub> NAAQS in 2022.

See Table 6-2 and Figure 6-5 for 1-hour values for all the monitoring sites. The maximum daily 1-hour average concentration was 97 ppb, recorded at Bayonne. Elizabeth Lab had the highest 1-hour 98th percentile value of 63 ppb, and the highest design value for 2020-2022, with 61 ppb.

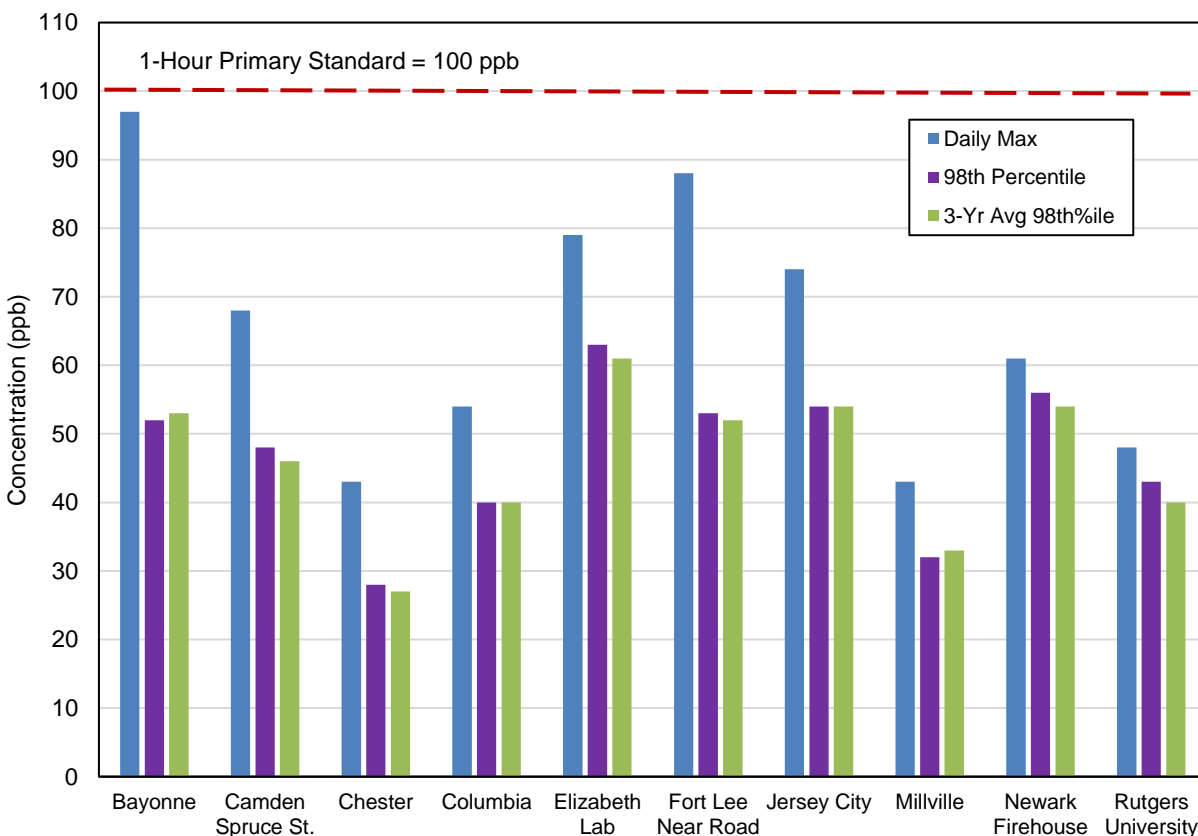
As noted, the Newark Firehouse site had no data after September 26<sup>th</sup> because the site had to be vacated.

**Table 6-2**  
**2022 Nitrogen Dioxide Concentrations in New Jersey**  
**Highest, 98<sup>th</sup>-Percentile, & 3-Year Average of 98<sup>th</sup>-Percentile Daily 1-Hour Averages**  
 Parts per Billion (ppb)

Monitoring Site	1-Hour Daily Average (ppb)		
	Maximum	98 <sup>th</sup> -Percentile	2020-2022 98 <sup>th</sup> -%ile 3-Year Avg
Bayonne	97	52	53
Camden Spruce St.	68	48	46
Chester	43	28	27
Columbia	54	40	40
Elizabeth Lab	79	63	61
Fort Lee Near Road	88	53	52
Jersey City	74	54	54
Millville	43	32	33
Newark Firehouse*	61	56	54
Rutgers University	48	43	40

\* Newark Firehouse was shut down 9/26/2022.

**Figure 6-5**  
**2022 Nitrogen Dioxide Concentrations in New Jersey**  
**Highest, 98<sup>th</sup>-Percentile, & 3-Year Average of 98<sup>th</sup>-Percentile Daily 1-Hour Averages**  
 Parts per Billion (ppb)



In order to meet the annual NAAQS for NO<sub>2</sub>, the calendar-year average (January 1 to December 31) must be less than or equal to 53 ppb, rounded to no more than one decimal place. The NJAAQS is also 53 ppb, but it is compared to the maximum running 12-month average (of any twelve consecutive months in the year). As shown in Table 6-3 and Figure 6-6, both the highest calendar-year average of 20 ppb and the highest 12-month running average of 20 ppb occurred at the Elizabeth Lab monitoring station, located off Exit 13 of the New Jersey Turnpike. Both these values are well below the standards.

**Table 6-3**  
**2022 Nitrogen Dioxide Concentrations in New Jersey**  
**Annual (12-Month) Averages**  
Parts per Billion (ppb)

Monitoring Site	12-Month Average (ppb)	
	Calendar Year	Maximum Running
Bayonne	15	15
Camden Spruce Street	11	11
Chester	3	3
Columbia	11	11
Elizabeth Lab	20	20
Fort Lee Near Road	16	16
Jersey City	17	18
Millville	6	6
Newark Firehouse*	14	15
Rutgers University	8	8

\* Newark Firehouse had no data after 9/26/2023.

**Figure 6-6**  
**2022 Nitrogen Dioxide Concentrations in New Jersey**  
**Annual (12-Month) Averages**  
Parts per Billion (ppb)

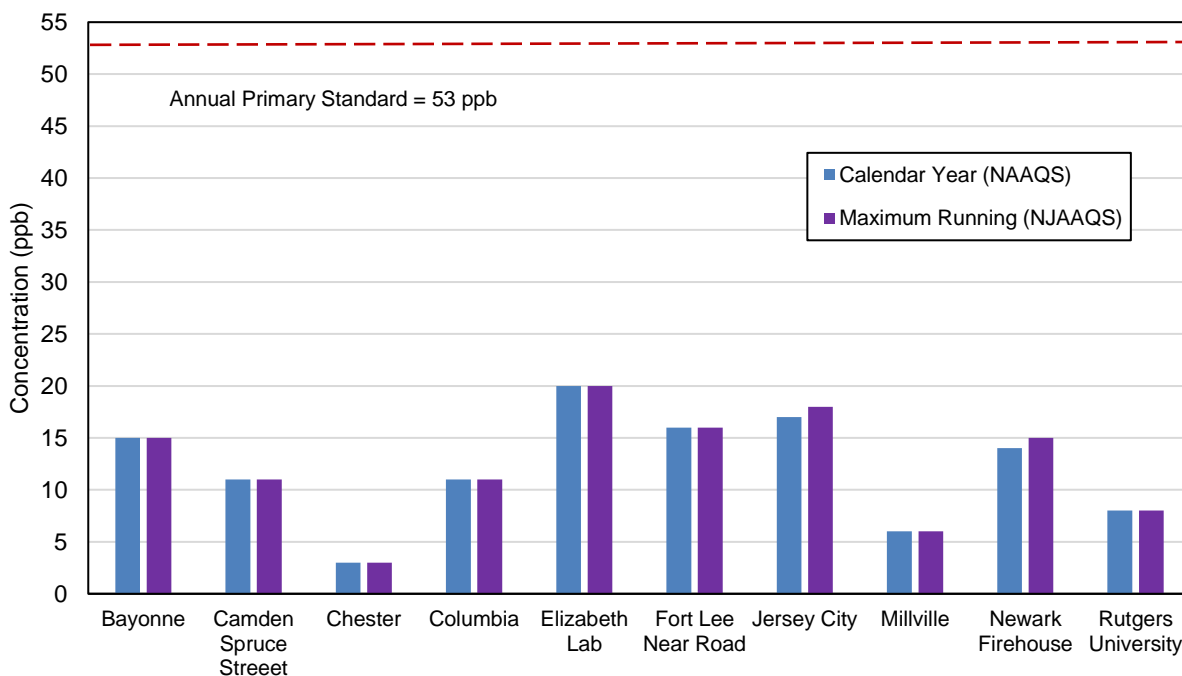
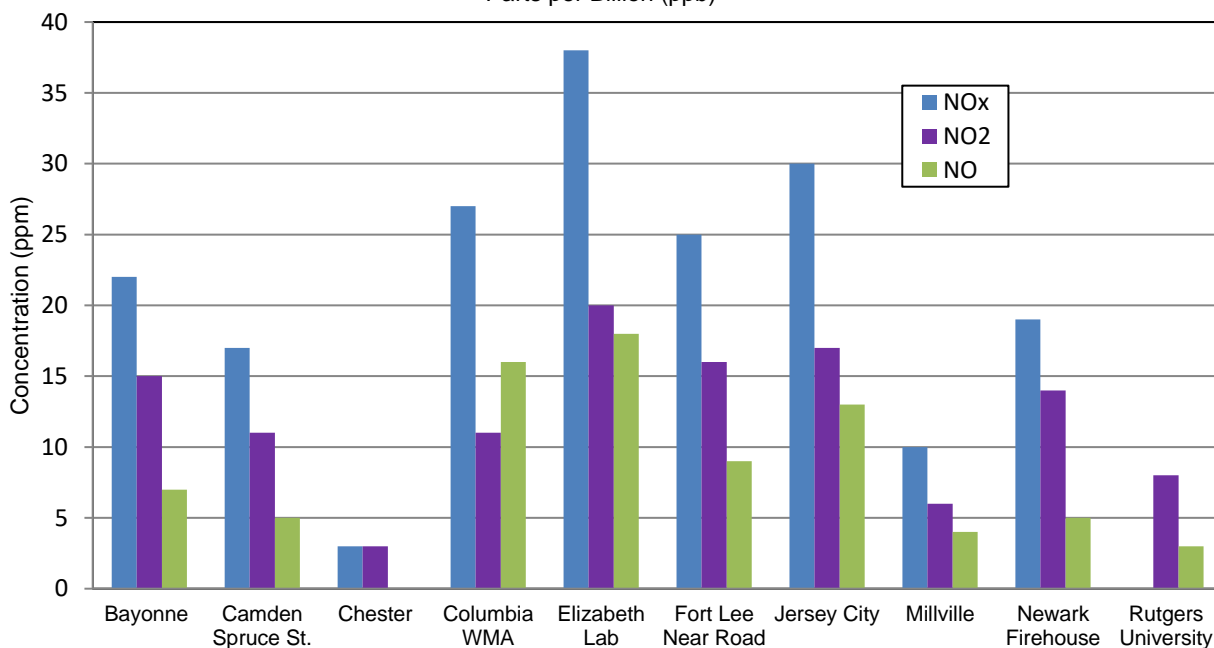


Figure 6-7 and Table 6-4 show the calendar-year annual average concentrations of NO<sub>2</sub>, NO and NO<sub>x</sub> at each New Jersey monitoring site. Even though there are no ambient air standards for NO and NO<sub>x</sub>, the stations that measure NO<sub>2</sub> concentrations also measure them (except for Rutgers, which measures NO<sub>y</sub> instead of NO<sub>x</sub>). NO<sub>x</sub> levels are approximately (not exactly) the sum of the NO<sub>2</sub> and NO concentrations. The concentration of NO tends to be lower than NO<sub>2</sub>, because it quickly reacts with other air pollutants (particularly ozone) after it is emitted from a source and converts to NO<sub>2</sub>. The Columbia monitoring site is an exception to this, with annual average levels of NO higher than NO<sub>2</sub>. The monitor is about 100 feet from Interstate Highway 80. The road is a significant source of NO emissions from vehicles, but the expected conversion of NO to NO<sub>2</sub> is probably hindered by the area's relatively low levels of other pollutants.

**Figure 6-7**  
**2022 Nitrogen Oxides Concentrations in New Jersey**  
**Annual Averages**  
 Parts per Billion (ppb)



**Table 6-4**  
**2022 Nitrogen Oxides Concentrations in New Jersey**  
**Annual Averages**  
 Parts per Billion (ppb)

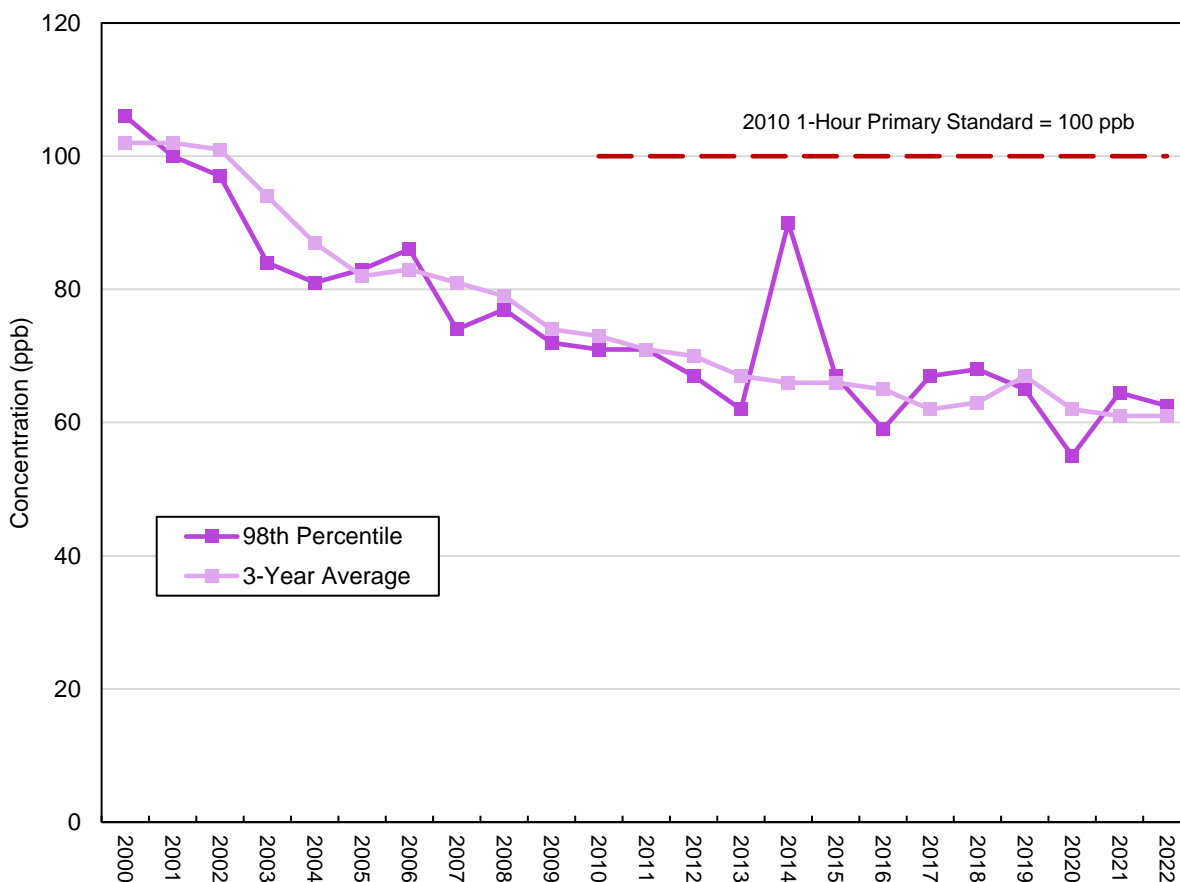
Site	NO <sub>2</sub>	NO	NO <sub>x</sub>
Bayonne	15	7	22
Camden Spruce Street	11	5	17
Chester	3	0	3
Columbia	11	16	27
Elizabeth Lab	20	18	38
Fort Lee Near Road	16	9	25
Jersey City	17	13	30
Millville	6	4	10
Newark Firehouse*	14	5	19
Rutgers University	8	3	

\*Newark Firehouse had no data after 9/26/2022.

## NO<sub>2</sub> TRENDS

New Jersey has not violated the 1-hour NAAQS since it was implemented in 2010. Figure 6-8 shows the highest statewide 98<sup>th</sup> percentile and design values for the 1-hour NAAQS for the years 2000-2022. The design values presented below are the highest 3-year averages of the 98<sup>th</sup> percentile values of the daily maximum one-hour concentrations at any New Jersey monitoring site.

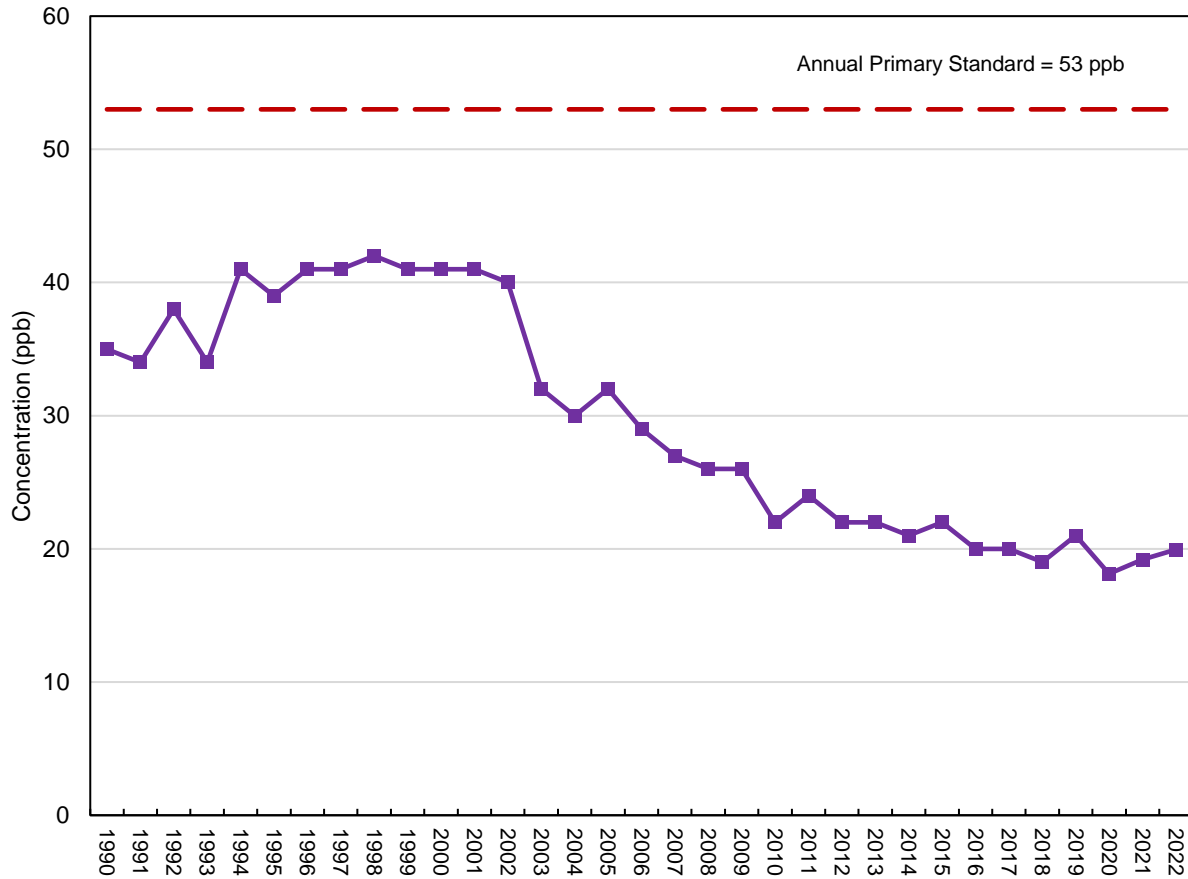
**Figure 6-8**  
**Statewide New Jersey Nitrogen Dioxide Trends, 1990-2022**  
**98<sup>th</sup> Percentile & 3-Year Average 98<sup>th</sup> Percentile Daily Maximum 1-Hour Concentrations**  
Parts per Billion (ppb)



Routine monitoring for NO<sub>2</sub> in New Jersey began in 1966. The last year in which the annual average NO<sub>2</sub> concentration exceeded the NAAQS was 1974. The graph of NO<sub>2</sub> levels in Figure 6-9 shows the highest statewide annual average concentrations recorded from 1990 to 2022. Although NO<sub>2</sub> concentrations are well within the NAAQS, there is still a great deal of concern about the role of nitrogen oxides in the formation of other pollutants, most notably ozone and fine particles. Both of those pollutants still occasionally reach problematic levels in the northeastern United States. Efforts to reduce levels of ozone and fine particles are likely to require continued reductions in NO<sub>x</sub> emissions.

The statewide trend data is also presented in Table 6-5.

**Figure 6-9**  
**Statewide New Jersey Nitrogen Dioxide Trend, 1990-2022**  
**Highest Annual (Calendar Year) Average Concentrations**  
 Parts per Billion (ppb)



**Table 6-5**  
**Statewide New Jersey Nitrogen Dioxide Trends, 1990-2022**  
**1-Hour Daily & 3-Year Average of 98<sup>th</sup> Percentile Concentrations**  
**Maximum Annual Average Concentrations**  
Parts per Billion (ppb)

Year	1-Hour Daily		Annual Average*
	98 <sup>th</sup> Percentile	3-Year Average of 98 <sup>th</sup> Percentile*	
1990	112	---	35
1991	103	---	34
1992	120	112	38
1993	96	106	34
1994	116	111	41
1995	92	101	39
1996	104	104	41
1997	106	101	41
1998	101	104	42
1999	100	102	41
2000	106	102	41
2001	100	102	41
2002	97	101	40
2003	84	94	32
2004	81	87	30
2005	83	82	32
2006	86	83	29
2007	74	81	27
2008	77	79	26
2009	72	74	26
2010	71	73	22
2011	71	71	24
2012	67	70	22
2013	62	67	22
2014	90	66	21
2015	67	66	22
2016	59	65	20
2017	67	62	20
2018	68	63	19
2019	65	67	21
2020	55	62	18
2021	65	61	19
2022	63	61	20

\*Design value



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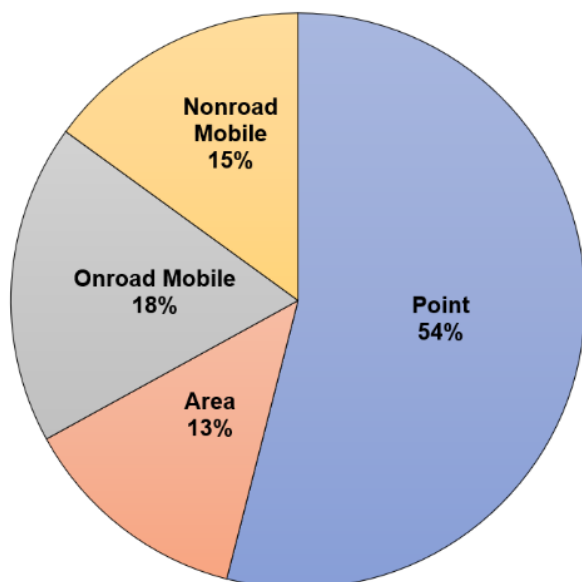
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# 2022 Sulfur Dioxide Summary

New Jersey Department of Environmental Protection

**Figure 7-1**  
**2017 New Jersey**  
**SO<sub>2</sub> Annual Emissions**



## SOURCES

Sulfur dioxide (SO<sub>2</sub>) is a heavy, colorless gas with a suffocating odor, that easily dissolves in water to form sulfuric acid. SO<sub>2</sub> gases are formed when fuels containing sulfur (coal, oil, and gasoline) are burned, or when gasoline is extracted from oil. Most of the sulfur dioxide released into the air comes from fuel combustion in electric utilities, especially those that burn coal with a high sulfur content. Sulfur is found in raw materials such as crude oil, coal, and ores that contain metals. Industrial facilities that derive their products from these materials may also release SO<sub>2</sub>. The pie chart in Figure 7-1 summarizes the primary sources of SO<sub>2</sub> in New Jersey in 2017, the most recent data available.

## HEALTH AND ENVIRONMENTAL EFFECTS

Sulfur dioxide causes irritation of the mucous membranes. This is probably the result of sulfurous acid forming when the highly soluble SO<sub>2</sub> gas dissolves at the surface of the membranes. Groups that are especially susceptible to the harmful health effects of SO<sub>2</sub> include children, the elderly, and people with heart or lung disorders such as asthma. When SO<sub>2</sub> concentrations in the air become elevated, people in these sensitive groups and those who are active outdoors may have trouble breathing.

Sulfur dioxide reacts with other gases and particles in the air to form sulfates, which also can be harmful to people and the environment. Sulfate particles are the major cause of reduced visibility in the eastern United States. SO<sub>2</sub> forms acids that fall to the earth in rain and snow. Better known as acid rain, this acidic precipitation can damage forests and crops, can make lakes and streams too acidic for fish, and can speed up the decay of building materials and paints.

## AMBIENT AIR QUALITY STANDARDS

The current National Ambient Air Quality Standards (NAAQS) for SO<sub>2</sub> are shown in Table 7-1. Primary standards are set to provide public health protection, including protecting the health of sensitive populations such as asthmatics, children, and the elderly. Secondary standards provide public welfare protection, including protection against decreased visibility and damage to animals, crops, vegetation, and buildings. In June 2010 the United States Environmental Protection Agency (USEPA) established a new primary 1-hour NAAQS for SO<sub>2</sub> at a level of 75 parts per billion (ppb). At the same time, the old 24-hour and annual average NAAQS were revoked, and the 3-hour secondary NAAQS was retained. Compliance with the 1-hour standard is determined by calculating the 99th percentile of 1-hour daily maximum concentrations for each monitoring site in the state each year, and then averaging each site's values for the three most recent years. This statistic is called the design value. Compliance with the secondary standard is based on the second-highest 3-hour average concentration for a given year.

Table 7-1 also shows New Jersey's ambient air quality standards (NJAAQS) for SO<sub>2</sub>, which are based on the older NAAQS. NJAAQS for SO<sub>2</sub> are calculated using running averages (consecutive 3-hour, 24-hour and 12 month averages) rather than calendar year or non-overlapping block averages. The secondary 3-hour New Jersey standard is the same as the NAAQS, except that New Jersey uses a running average. Also, the NJAAQS use ppm units instead of ppb.

**Table 7-1**  
**National and New Jersey Ambient Air Quality Standards for Sulfur Dioxide (SO<sub>2</sub>)**

Parts per Billion (ppb)  
Parts per Million (ppm)

Averaging Period	Type	National Level	New Jersey Level <sup>a</sup>	Design Value
1-hour	Primary	75 ppb	---	3-year average of the annual 99th percentile daily maximums
3-hours	Secondary	0.5 ppm <sup>b</sup>	0.5 ppm	Annual 2 <sup>nd</sup> -highest
24-hours	Primary	---	0.14 ppm	Annual 2 <sup>nd</sup> -highest
24-hours	Secondary	---	0.1 ppm	Annual 2 <sup>nd</sup> -highest
12-months	Primary	---	0.03 ppm	Not to be exceeded
12-months	Secondary	---	0.02 ppm	Not to be exceeded

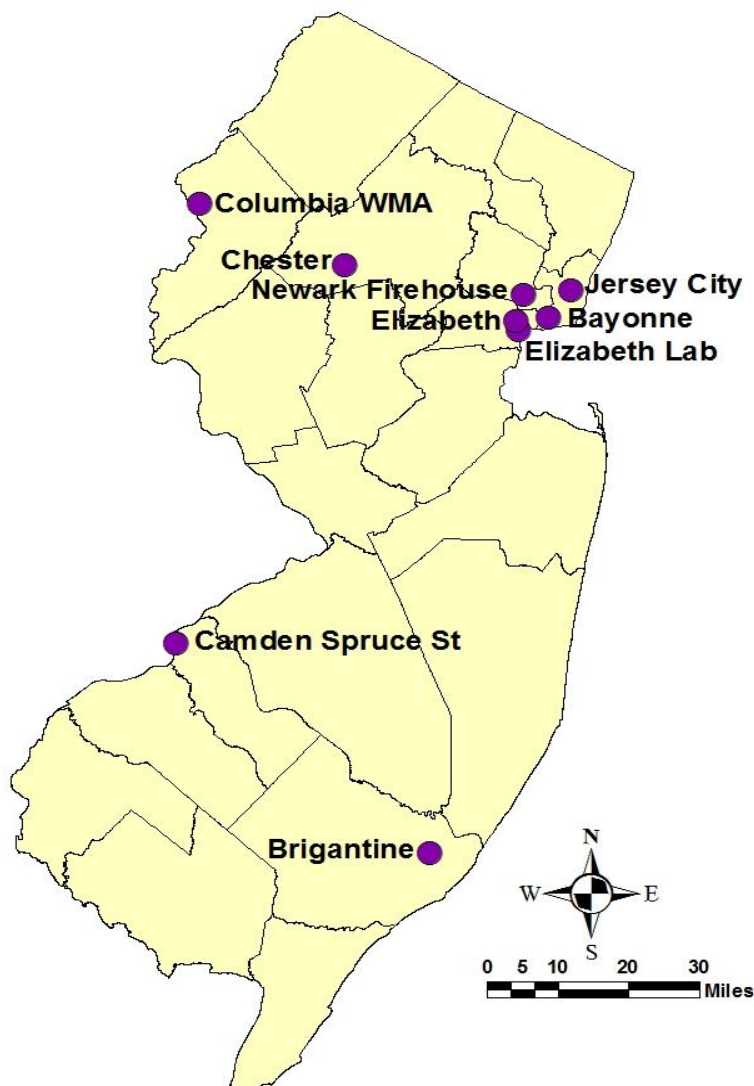
<sup>a</sup> Based on running averages, over any 12 consecutive months.

<sup>b</sup> Based on successive non-overlapping blocks, beginning at midnight each day.

## SO<sub>2</sub> MONITORING NETWORK

The New Jersey Department of Environmental Protection (NJDEP) monitored SO<sub>2</sub> levels at nine sites in 2022. The monitoring stations are Bayonne, Brigantine, Camden Spruce Street, Chester, Columbia, Elizabeth, Elizabeth Lab, Jersey City, and Newark Firehouse. However, the Newark Firehouse monitoring station was shutdown unexpectedly in September 2022. The locations are shown in Figure 7-2.

**Figure 7-2**  
**2022 Sulfur Dioxide Monitoring Network**



## SO<sub>2</sub> LEVELS IN 2022

For 2022, there were no exceedances of any AAQS. 1-hour daily data is presented in Table 7-2 and Figure 7-3. Elizabeth Lab had the highest 1-hour value at 8.8 ppb, and Camden had the second-highest at 5.9 ppb. Elizabeth Lab had the highest 99<sup>th</sup> percentile value at 5.2 ppb. The highest design value, which is the 3-year average of the 99<sup>th</sup> percentile of the daily maximum 1-hour SO<sub>2</sub> concentrations, was at Jersey City with a value of 8 ppb.

Three-hour averages for all sites were well below the national and New Jersey 3-hour secondary standards of 0.5 ppm for the second-highest value. The NAAQS is based on successive non-overlapping 3-hour blocks, while the NJAAQS uses running 3-hour averages (although the second-highest value can't overlap the highest value). The maximum and second-highest values were all measured at Elizabeth Lab. The block averages were 0.0052 and 0.0049 ppm respectively, and the running averages were 0.0053 and 0.0049 ppm. Results are shown in Table 7-3 and Figure 7-4.

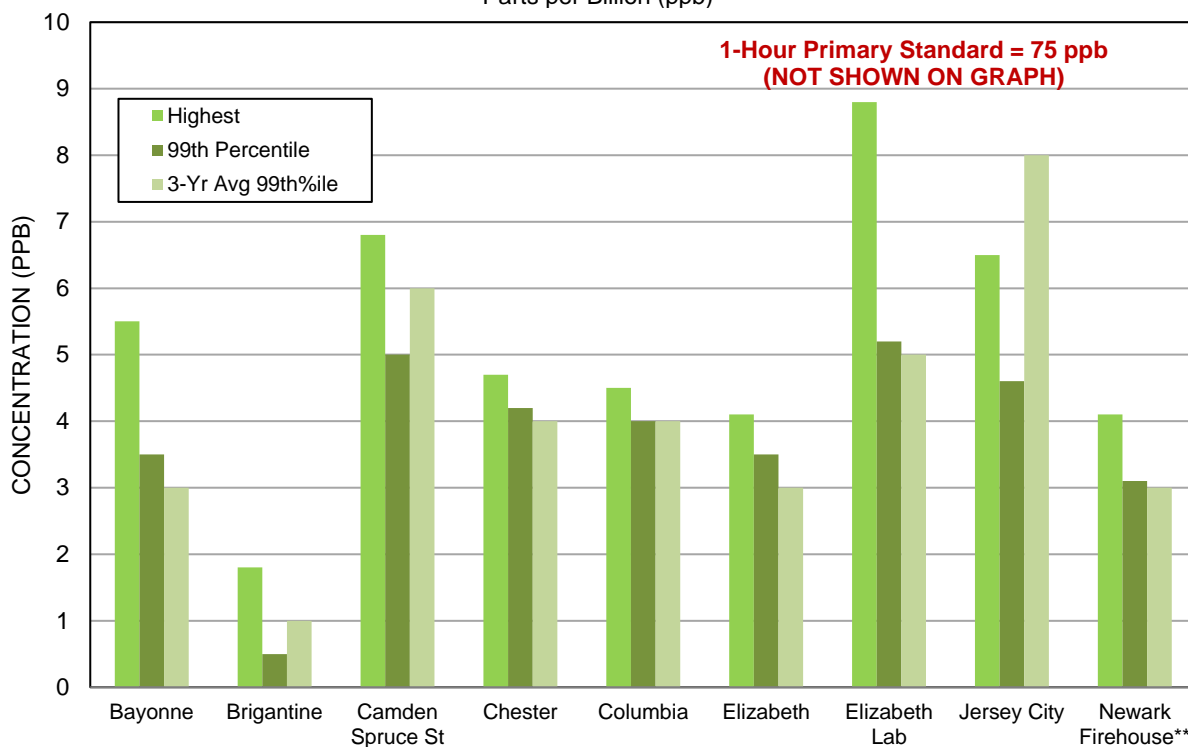
The New Jersey 24-hour AAQS is 0.14 ppm, and the 12-month standard is 0.03 ppm. In 2022, the highest and second-highest 24-hour average concentrations were 0.0038 ppm and 0.0033 ppm, both at the Jersey City site. The highest 12-month running average concentration of was 0.0009 ppm at Elizabeth Lab. See Tables 7-4 and 7-5, and Figures 7-5 and 7-6, for data for the other monitoring sites.

**Table 7-2**  
**2022 Sulfur Dioxide Concentrations in New Jersey**  
**Highest, 99<sup>th</sup> Percentile, and 3-Year Average of the 99<sup>th</sup> Percentile 1-Hour Daily Averages**  
 Parts per Billion (ppb)

Monitoring Site	1-Hour Daily Average (ppb)		
	Highest	99 <sup>th</sup> Percentile	2020-2022 99 <sup>th</sup> -%ile 3-Yr Avg
Bayonne	5.5	3.5	3
Brigantine	1.8	0.5	1
Camden Spruce Street	6.8	5.0	6
Chester	4.7	4.2	4
Columbia	4.5	4.0	4
Elizabeth	4.1	3.5	3
Elizabeth Lab	8.8	5.2	5
Jersey City	6.5	4.6	8
Newark Firehouse*	4.1	3.1	3

\*The Newark Firehouse monitoring station was shut down on 9/26/2022.

**Figure 7-3**  
**2022 Sulfur Dioxide Concentrations in New Jersey**  
**Highest, 99<sup>th</sup> Percentile, and 3-Year Average of the 99<sup>th</sup> Percentile 1-Hour Daily Averages**  
 Parts per Billion (ppb)



**Table 7-3**  
**2022 Sulfur Dioxide Concentrations in New Jersey**  
**3-Hour Averages**  
 Parts per Million (ppm)

Monitoring Site	3-Hour Average Concentrations			
	Block <sup>a</sup>		Running <sup>b</sup>	
	Maximum	2nd-Highest	Maximum	2nd-Highest*
Bayonne	0.0047	0.0031	0.0047	0.0033
Brigantine	0.0008	0.0007	0.0009	0.0009
Camden Spruce Street	0.0045	0.0037	0.0045	0.0037
Chester	0.0033	0.0030	0.0035	0.0033
Columbia	0.0031	0.0025	0.0031	0.0031
Elizabeth	0.0037	0.0034	0.0037	0.0035
Elizabeth Lab	0.0052	0.0049	0.0053	0.0049
Jersey City	0.0045	0.0044	0.0046	0.0046
Newark Firehouse**	0.0031	0.0029	0.0031	0.0029

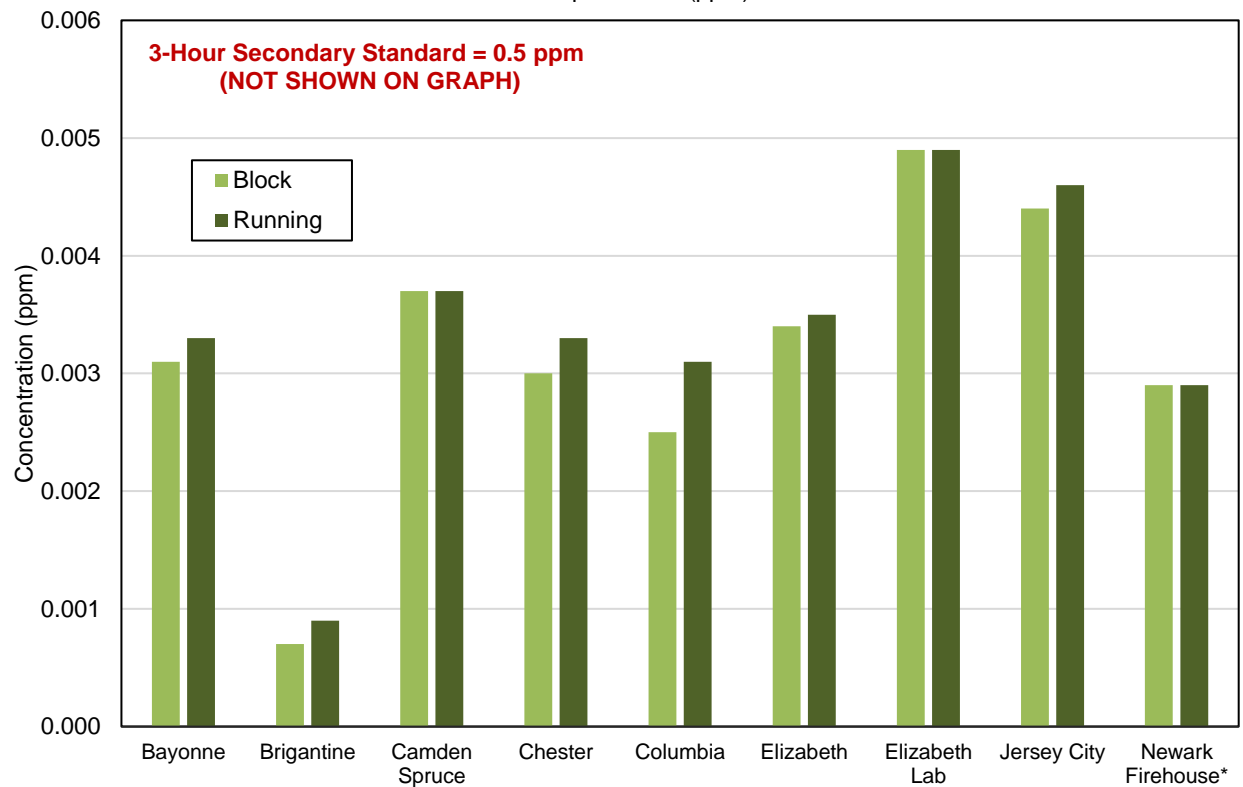
<sup>a</sup> NAAQS

<sup>b</sup> NJAQS

\*Non-overlapping

\*\*Newark Firehouse shut down on 9/26/22.

**Figure 7-4**  
**2022 Sulfur Dioxide Concentrations in New Jersey**  
**2<sup>nd</sup> Highest 3-Hour Averages**  
 Parts per Million (ppm)

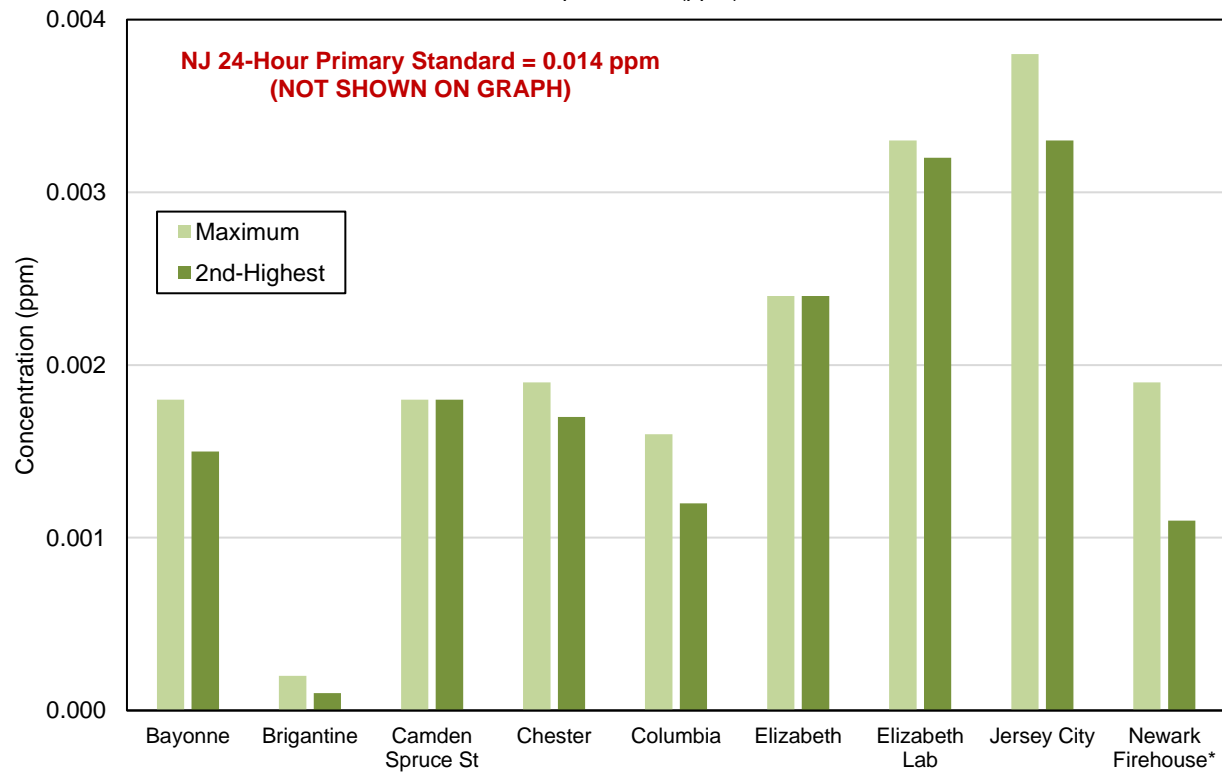


**Table 7-4**  
**2022 Sulfur Dioxide Concentrations in New Jersey**  
**24-Hour Running Averages**  
 Parts per Million (ppm)

Monitoring Site	24-Hour Running Average	
	Maximum	2 <sup>nd</sup> Highest (Non-overlapping)
Bayonne	0.0018	0.0015
Brigantine	0.0002	0.0001
Camden Spruce Street	0.0018	0.0018
Chester	0.0019	0.0017
Columbia	0.0016	0.0012
Elizabeth	0.0024	0.0024
Elizabeth Lab	0.0033	0.0032
Jersey City	0.0038	0.0033
Newark Firehouse*	0.0019	0.0011

\*Newark Firehouse shut down on 9/26/22.

**Figure 7-5**  
**2022 Sulfur Dioxide Concentrations in New Jersey**  
**24-Hour Running Averages**  
 Parts per Million (ppm)



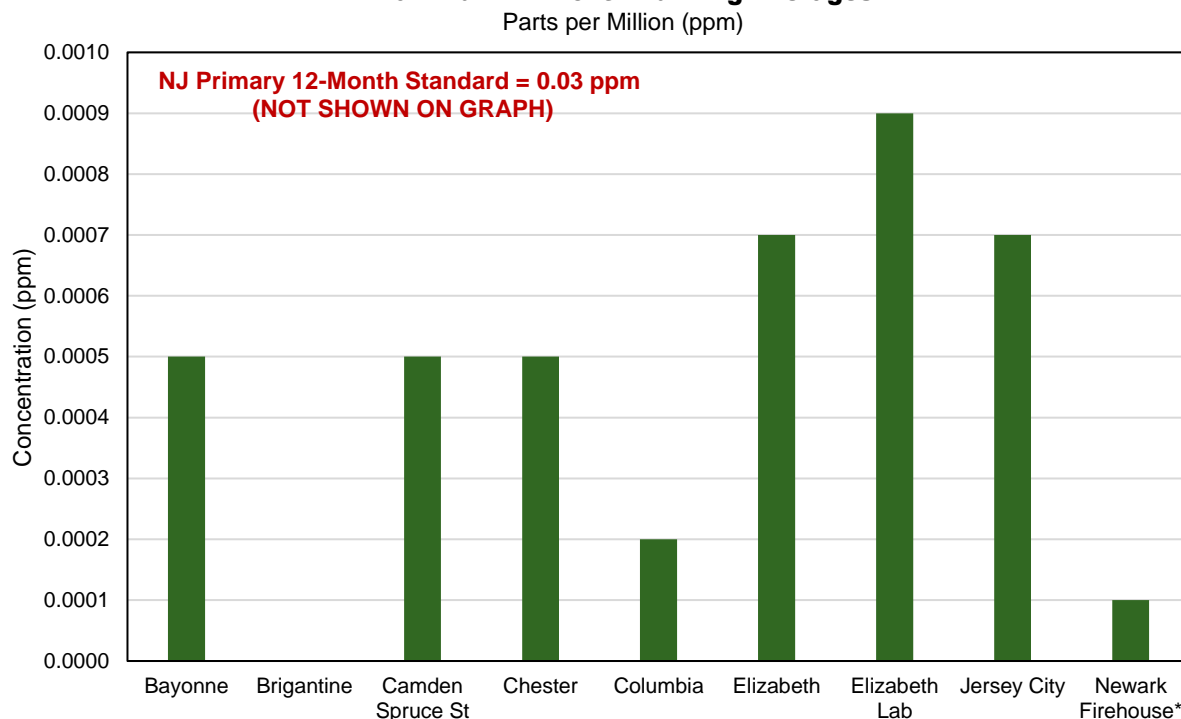
**Table 7-5**  
**2022 Sulfur Dioxide Concentrations in New Jersey**  
**Maximum 12-Month Running Averages**  
 Parts per Million (ppm)

Monitoring Site	Maximum 12-Month Running Average
Bayonne	0.0005
Brigantine	0.0000
Camden Spruce Street	0.0005
Chester	0.0005
Columbia	0.0002
Elizabeth	0.0007
Elizabeth Lab	0.0009
Jersey City	0.0007
Newark Firehouse*	0.0001

\*Newark Firehouse shut down on 9/26/22.



**Figure 7-6**  
**2022 Sulfur Dioxide Concentrations in New Jersey**  
**Maximum 12-Month Running Averages**



\*Newark Firehouse shut down on 9/26/22.

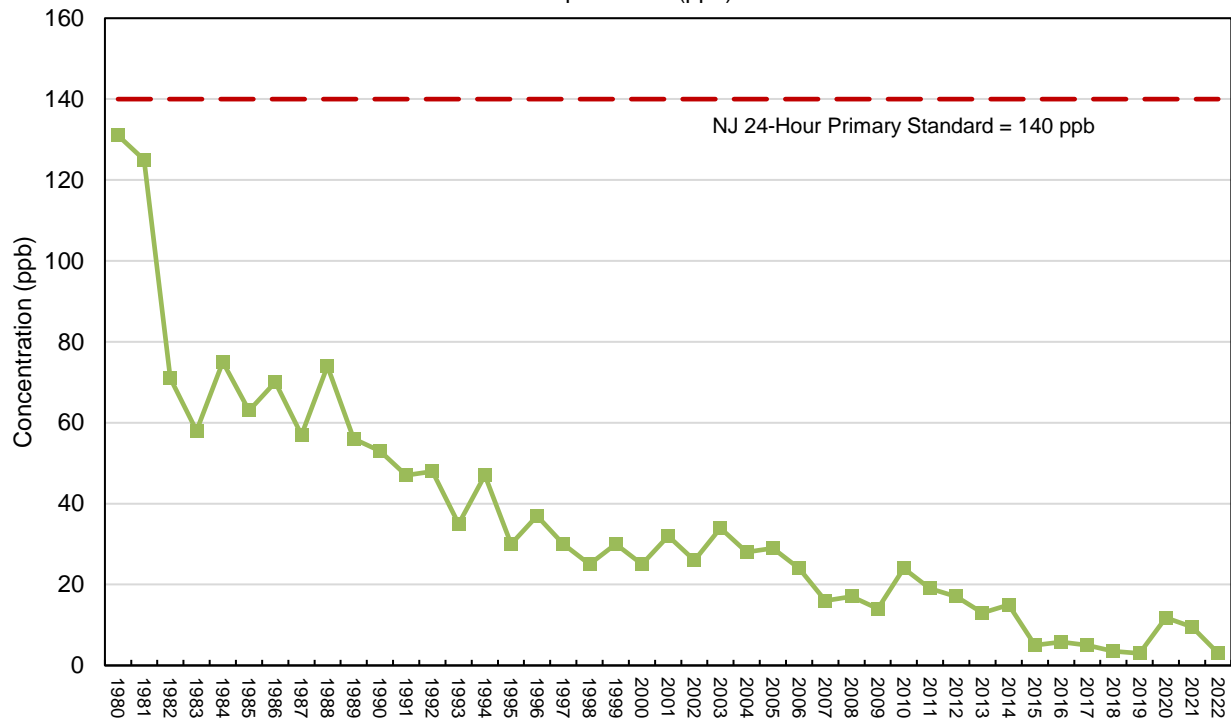
## SO<sub>2</sub> TRENDS

Sulfur dioxide concentrations across the country have decreased significantly since the first NAAQS were set in 1971. Figure 7-7 shows the second-highest daily average concentrations of SO<sub>2</sub> recorded in New Jersey each year since 1980 (also see Table 7-6). Nationwide efforts to reduce ambient sulfur levels have focused on sulfur in fuels. Regulations passed in 2000 reduced the sulfur content of gasoline by up to 90 percent, and enabled the use of new emission control technologies in cars, sport utility vehicles (SUVs), minivans, vans and pick-up trucks (beginning with model year 2004). Even more stringent gasoline and emissions controls for sulfur went into effect in 2017. And in New Jersey, limits on sulfur in commercial fuel oil were implemented beginning in 2014.

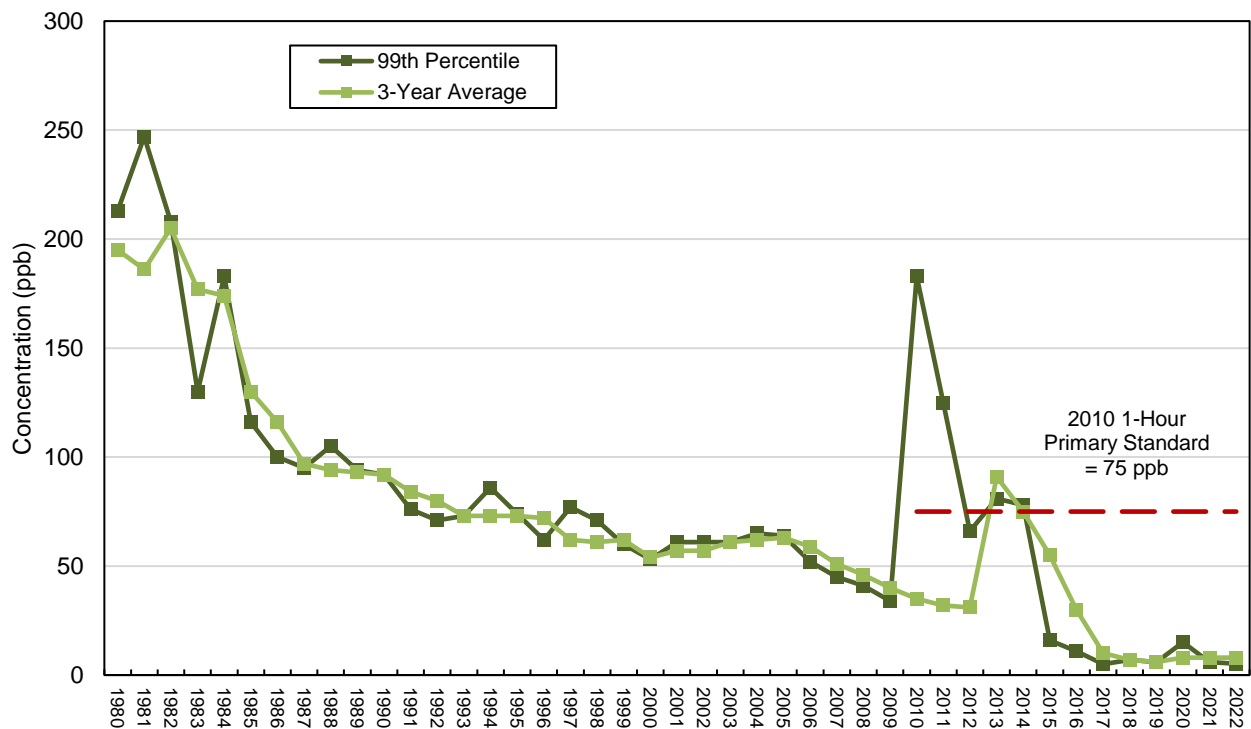
A coal-burning power plant across the Delaware River in Pennsylvania had for many years been suspected of causing high SO<sub>2</sub> levels in New Jersey. Air dispersion modeling carried out by NJDEP showed that the facility was causing likely violations of the SO<sub>2</sub> NAAQS. New Jersey petitioned the USEPA under Section 126 of the Clean Air Act to take action against the Portland Power Plant. In support of the petition, NJDEP established an SO<sub>2</sub> monitoring station at the Columbia Wildlife Management Area in Knowlton Township, Warren County, in September 2010. The dramatic increase in the monitored 99<sup>th</sup> percentile 1-hour SO<sub>2</sub> concentration in 2010 (shown in Figure 7-8) is attributable to measurements taken at the Columbia site. In October 2011, USEPA finalized a rule to grant New Jersey's petition. This final rule required the Portland Power Plant to reduce its SO<sub>2</sub> emissions such that the plant's contribution to predicted air quality standard violations would be lowered within one year, and completely eliminated within three years. The power plant stopped operating in mid-2014. Recent monitoring data has shown that Warren County and its vicinity are now able to meet the 1-hour SO<sub>2</sub> NAAQS.

Figure 7-8 also shows the trend in the design value, the value that determines compliance with the NAAQS. The design value for the 1-hour NAAQS is the 3-year average of the 99<sup>th</sup> percentile of the daily maximum 1-hour concentrations of SO<sub>2</sub> at each site. The values presented are the highest statewide for a given year.

**Figure 7-7**  
**Statewide New Jersey Sulfur Dioxide Trend, 1980-2022**  
**2nd-Highest 24-Hour Average Concentrations**  
 Parts per Billion (ppb)



**Figure 7-8**  
**Statewide New Jersey Sulfur Dioxide Trends, 1980-2022**  
**99<sup>th</sup> Percentile & 3-Year Average of the 99<sup>th</sup> Percentile Daily Maximum 1-Hour Concentrations**  
 Parts per Billion (ppb)



**Table 7-6**  
**Statewide New Jersey Sulfur Dioxide Trends**  
Parts per Billion (ppb)

Year	24-Hour Average	1-Hour Daily Maximum	
	2nd-Highest	99th Percentile	3-Year Average*
1980	131	213	195
1981	125	247	186
1982	71	208	205
1983	58	130	177
1984	75	183	174
1985	63	116	130
1986	70	100	116
1987	57	95	97
1988	74	105	94
1989	56	94	93
1990	53	92	92
1991	47	76	84
1992	48	71	80
1993	35	73	73
1994	47	86	73
1995	30	74	73
1996	37	62	72
1997	30	77	62
1998	25	71	61
1999	30	60	62
2000	25	53	54
2001	32	61	57
2002	26	61	57
2003	34	61	61
2004	28	65	62
2005	29	64	63
2006	24	52	59
2007	16	45	51
2008	17	41	46
2009	14	34	40
2010	24	183	35
2011	19	125	32
2012	17	66	31
2013	13	81	91
2014	15	78	75
2015	5	16	55
2016	5.8	11	30
2017	5	5	10
2018	3.5	7	7
2019	3	6	6
2020	11.8	15	8
2021	9.5	6	8
2022	3	5.2	8

\*Design value

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# 2022 Carbon Monoxide Summary

New Jersey Department of Environmental Protection

## SOURCES

Carbon monoxide (CO) is a colorless, odorless gas formed when carbon in fuels is not burned completely. The main source of outdoor CO is exhaust from internal combustion engines, primarily on-road vehicles, as well as non-road vehicles, generators, construction equipment, boats, and other types of mobile sources. Fifty percent of all CO emissions nationwide are attributable to mobile sources, and over 90% in New Jersey. Significant amounts of CO are also emitted from fuel combustion in boilers and incinerators, natural sources such as forest fires, and various industrial processes. A pie chart estimating the contribution of different source categories of CO in New Jersey in 2017 (latest estimate available) is shown in Figure 8-1.

## HEALTH EFFECTS

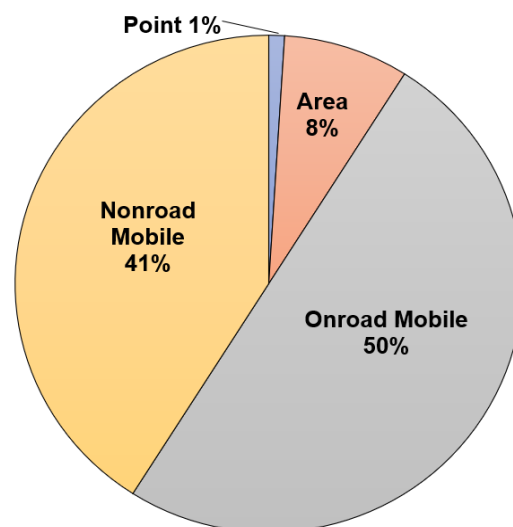
Carbon monoxide reduces the oxygen-carrying capacity of blood, therefore reducing the distribution of oxygen to organs like the heart and brain. The most common symptoms of exposure to high concentrations of carbon monoxide are headaches and nausea. Exposure to extremely high concentrations, usually resulting from combustion exhaust accumulating in enclosed indoor spaces, can be life-threatening. Such high levels of CO are not likely to occur outdoors. The health threat from exposure to outdoor CO is most serious for those who suffer from cardiovascular disease. For a person with heart disease, a single exposure to CO at low levels may reduce that individual's ability to exercise and may cause chest pain (also known as angina).

## AMBIENT AIR QUALITY STANDARDS

National Ambient Air Quality Standards (NAAQS) for CO are summarized in Table 8-1. Primary standards are set to protect the health of the public, including sensitive populations such as asthmatics, children, and the elderly. For carbon monoxide, there are currently two primary health-based, NAAQS: a 1-hour standard of 35 parts per million (ppm), and an 8-hour standard of 9 ppm. These levels are not to be exceeded more than once in any calendar year, so the design values, or the actual statistical values that determine compliance with the NAAQS, are the second-highest 1-hour and 8-hour values in a given year. Even though New Jersey's primary standards are the same as the NAAQS, the 8-hour state standard is based on a running average, not to be exceeded more than once in a 12-month period, rather than a calendar year.

Secondary standards provide public welfare protection from decreased visibility and damage to animals, crops, vegetation, and buildings. Although there are no national secondary standards for CO at this time, New Jersey has set secondary standards for CO equal to the primary standards.

**Figure 8-1**  
**2017 New Jersey**  
**CO Annual Emissions**



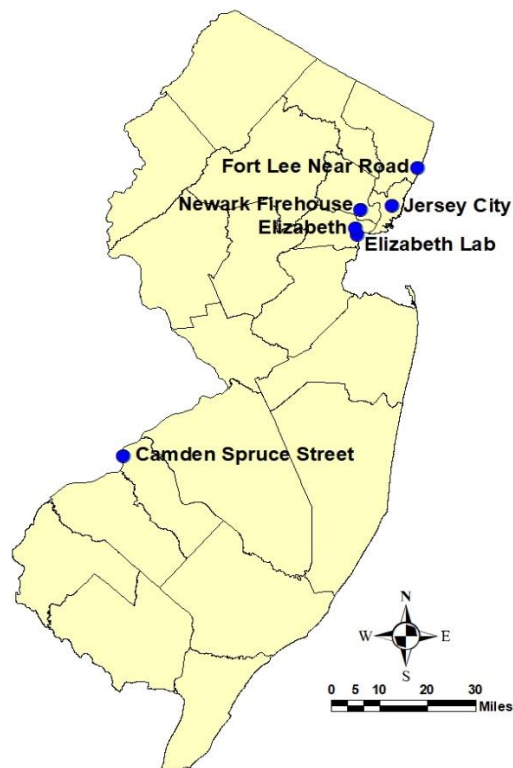
**Table 8-1**  
**National and New Jersey Ambient Air Quality Standards**  
**for Carbon Monoxide**  
Parts per Million (ppm)

Averaging Period	Type	National Level	New Jersey Level	Design Value
1-Hour	Primary	35 ppm	35 ppm	Annual 2 <sup>nd</sup> -highest
1-Hour	Secondary	----	35 ppm	2 <sup>nd</sup> -highest 12-month value
8-Hours	Primary	9 ppm	9 ppm	Annual 2 <sup>nd</sup> -highest
8-Hours	Secondary	----	9 ppm	2 <sup>nd</sup> -highest 12-month value

## CO MONITORING NETWORK

The New Jersey Department of Environmental Protection (NJDEP) had six CO monitors around the state in 2022, as shown on the map in Figure 8-2. They are located at the Camden Spruce Street, Elizabeth, Fort Lee Near Road, Jersey City, and Newark Firehouse stations. Unfortunately, the Newark Firehouse station had to be shut down in September of 2022. The Newark Firehouse station was part of the U.S. Environmental Protection Agency's (USEPA) National Core Multipollutant Monitoring Network (NCore). NJDEP is hoping to have a replacement station set up by the end of 2023.

**Figure 8-2**  
**2022 Carbon Monoxide Monitoring Network**



## CO LEVELS IN 2022

There were no exceedances of any CO standards at any of the New Jersey monitoring sites during 2022. The maximum and 2<sup>nd</sup>-highest 1-hour average CO concentrations, both recorded at the Jersey City site, were 3.8 ppm and 3.7 ppm. The highest and second-highest 8-hour average CO concentrations were 3.4 and 2.0 ppm, both also at Jersey City. Because Newark Firehouse is an NCore site, it measures and reports CO concentrations at trace levels, down to a thousandth of a ppm (0.000 ppm). However, data collection there stopped in late September. The 2022 data is summarized in Table 8-2, and Figures 8-3 and 8-4.

**Table 8-2**  
**2022 Carbon Monoxide Concentrations in New Jersey**

Parts per Million (ppm)

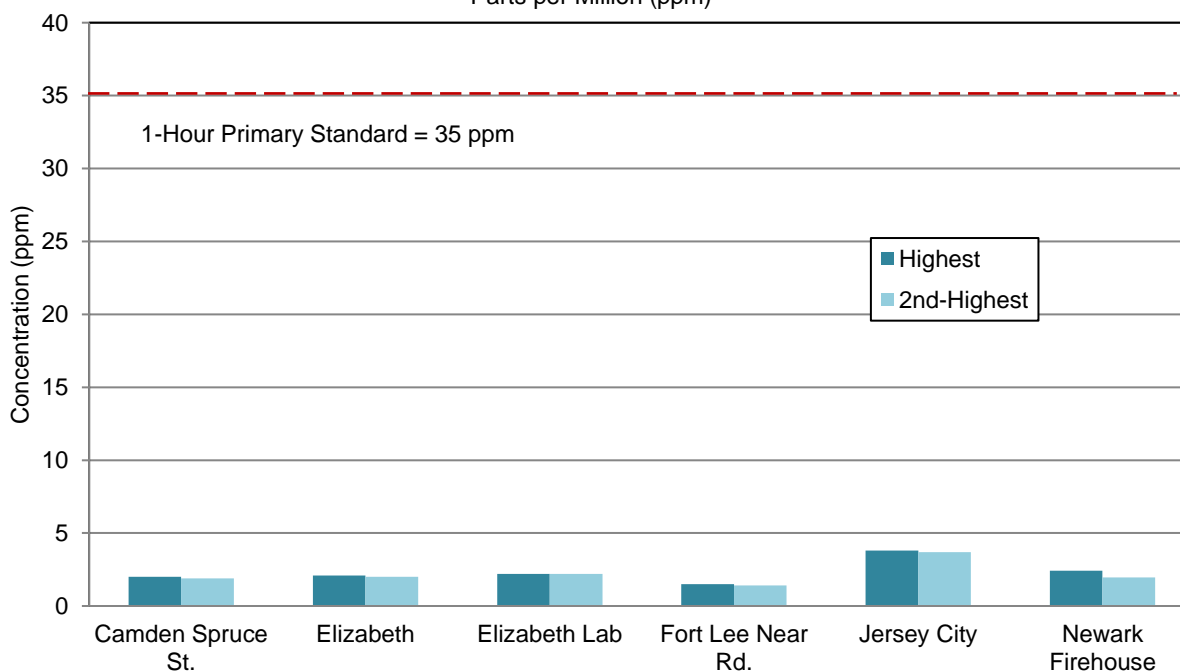
Monitoring Site	1-Hour Average Concentrations		8-Hour Average Concentrations	
	Highest	2nd-Highest	Highest	2nd-Highest*
Camden Spruce Street	2.0	1.9	1.5	1.4
Elizabeth	2.1	2.0	1.7	1.5
Elizabeth Lab	2.2	2.2	1.5	1.4
Fort Lee Near Road	1.5	1.4	1.2	1.2
Jersey City	3.8	3.7	3.4	2.0
Newark Firehouse**	2.424	1.951	1.4	1.3

\*Non-overlapping 8-hour periods

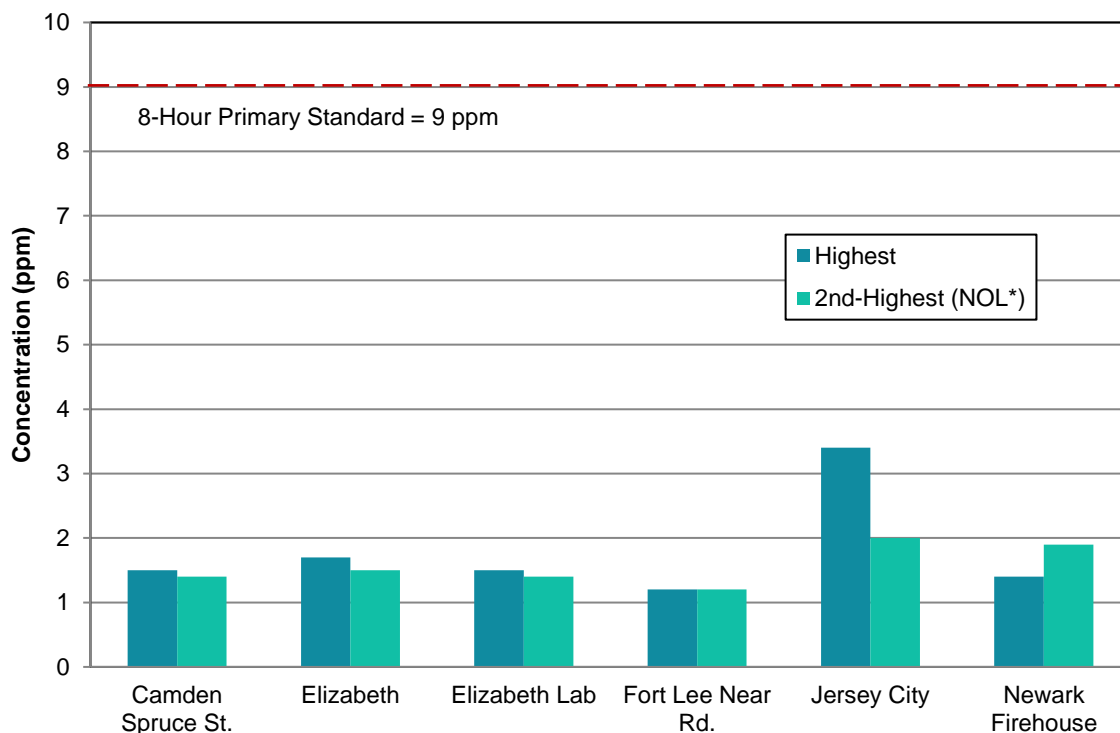
\*\*Data collection ended 9/26/2022.

**Figure 8-3**  
**2022 Carbon Monoxide Concentrations in New Jersey**

**1-Hour Averages**  
Parts per Million (ppm)



**Figure 8-4**  
**2022 Carbon Monoxide Concentrations in New Jersey**  
**8-Hour Averages**  
 Parts per Million (ppm)



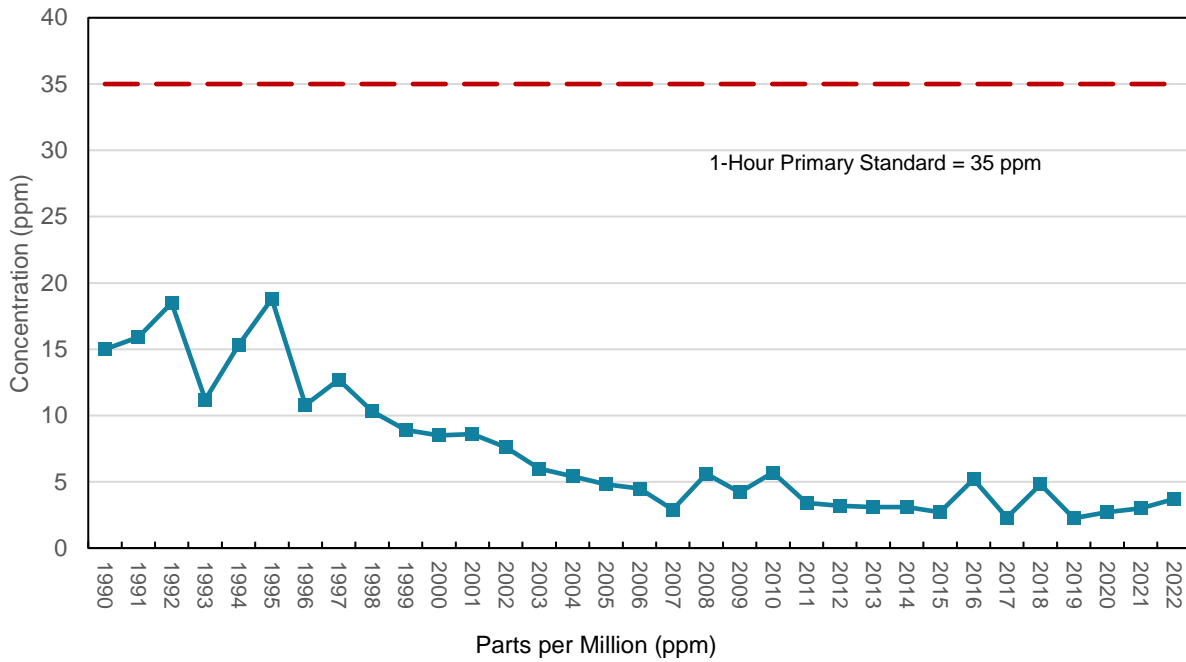
\*Non-overlapping 8-hour periods

## CO TRENDS

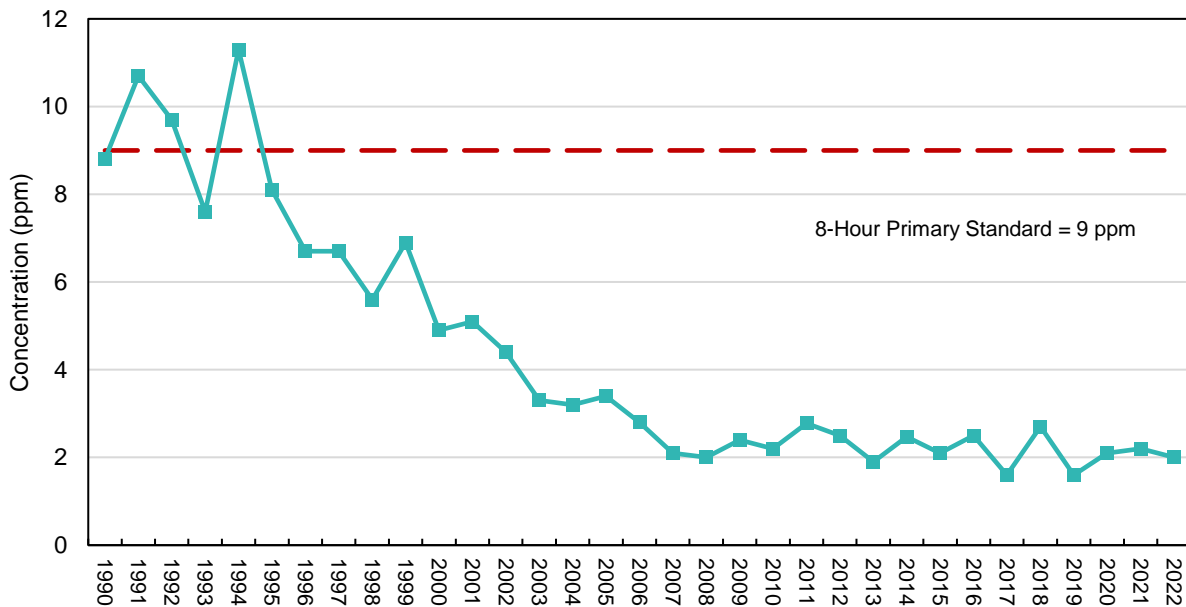
Carbon monoxide levels in outdoor air have improved dramatically over the past two-and-a-half decades. Figures 8-5 and 8-6 and Table 8-3 present the trends in CO levels since 1990. The graphs and table actually show the second-highest 1-hour and 8-hour values recorded, because those are the design values that determine if the NAAQS are being met (one exceedance per site is allowed each year). The entire state was officially declared to have attained the CO standards as of August 23, 2002. Years ago, unhealthy levels of CO were recorded on a regular basis. The reduction in CO levels is due primarily to cleaner-running cars and other vehicles, which are by far the largest source of this pollutant outdoors. The last violation of the 8-hour NAAQS was in 1994.



**Figure 8-5**  
**Carbon Monoxide Design Value Trend in New Jersey, 1990-2022**  
**2nd-Highest 1-Hour Average Concentration**  
 Parts per Million (ppm)



**Figure 8-6**  
**Carbon Monoxide Design Value Trend in New Jersey, 1990-2022**  
**2nd-Highest 8-Hour Average Concentration**  
 Parts per Million (ppm)



**Table 8-3**  
**Statewide New Jersey Carbon Monoxide Trends, 1990-2022**  
**1-Hour and 8-Hour Average Concentrations**  
Parts per Million (ppm)

Year	2nd-Highest Average Concentrations*	
	1-Hour	8-Hour
1990	15	8.8
1991	15.9	10.7
1992	18.5	9.7
1993	11.2	7.6
1994	15.3	11.3
1995	18.8	8.1
1996	10.8	6.7
1997	12.7	6.7
1998	10.3	5.6
1999	8.9	6.9
2000	8.5	4.9
2001	8.6	5.1
2002	7.6	4.4
2003	6	3.3
2004	5.4	3.2
2005	4.8	3.4
2006	4.5	2.8
2007	2.9	2.1
2008	5.6	2
2009	4.2	2.4
2010	5.7	2.2
2011	3.4	2.8
2012	3.2	2.5
2013	3.1	1.9
2014	3.1	2.5
2015	2.7	2.1
2016	5.2	2.5
2017	2.3	1.6
2018	4.8	2.7
2019	2.3	1.6
2020	2.7	2.1
2021	3.0	2.2
2022	3.7	2.0

\*Design values

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# 2022 Lead Summary

New Jersey Department of Environmental Protection

## SOURCES

Lead is a criteria pollutant as well as a Hazardous Air Pollutant listed under the 1990 Clean Air Act. It is one of the first known and most widely studied environmental and occupational toxins.

Lead was once commonly used in paint and gasoline, and is still used in batteries, solder, pipes, pottery, roofing materials and some cosmetics. Since 1980, there has been a 99% decrease in the average lead air concentration nationwide. A phase-out of lead additive in gasoline began in the mid-1970s, although it is still used in aviation fuel in some smaller aircraft. The U.S. Environmental Protection Agency (USEPA) National Emissions Inventory estimates that 4.36 tons of lead were emitted in New Jersey in 2017, mostly from aircraft. New Jersey no longer has any significant industrial sources of lead.

## HEALTH EFFECTS

Lead that is emitted into the air can be inhaled, or ingested after it settles (ingestion is actually the main route of human exposure to airborne lead). There is no level of lead exposure that is considered safe. The main target for lead toxicity is the nervous system, both in adults and children. However, children's developing brains are the most vulnerable to the effects of lead, leading to lifelong effects, even after exposure ceases. The brain damage caused by lead exposure can result in learning disabilities and delinquent behavior, impacting IQ and academic achievement. Lead can also damage red blood cells and weaken the immune system. Other effects in adults include increased blood pressure, cardiovascular disease, and decreased kidney function. In addition, lead is classified as a "probable human carcinogen."

## AMBIENT AIR QUALITY STANDARDS

A NAAQS for lead was first promulgated in 1978. A value of  $1.5 \mu\text{g}/\text{m}^3$  was established as both the primary and secondary standard. It was based on an average for each calendar quarter, and was not to be exceeded. The New Jersey AAQS was based on a rolling three-month average. Thirty years later, in 2008, the NAAQS was lowered tenfold to  $0.15 \mu\text{g}/\text{m}^3$ , also averaged over a rolling three-month period, and not to be exceeded.

A rolling three-month average considers each of the 12 three-month periods associated with a given year, not just the four calendar quarters within that year. The old NAAQS required lead to be sampled as total suspended particulate (TSP). In New Jersey, lead is now measured as  $\text{PM}_{10}$ .

**Table 9-1**  
**National Ambient Air Quality Standards for Lead**  
Micrograms Per Cubic Meter ( $\mu\text{g}/\text{m}^3$ )

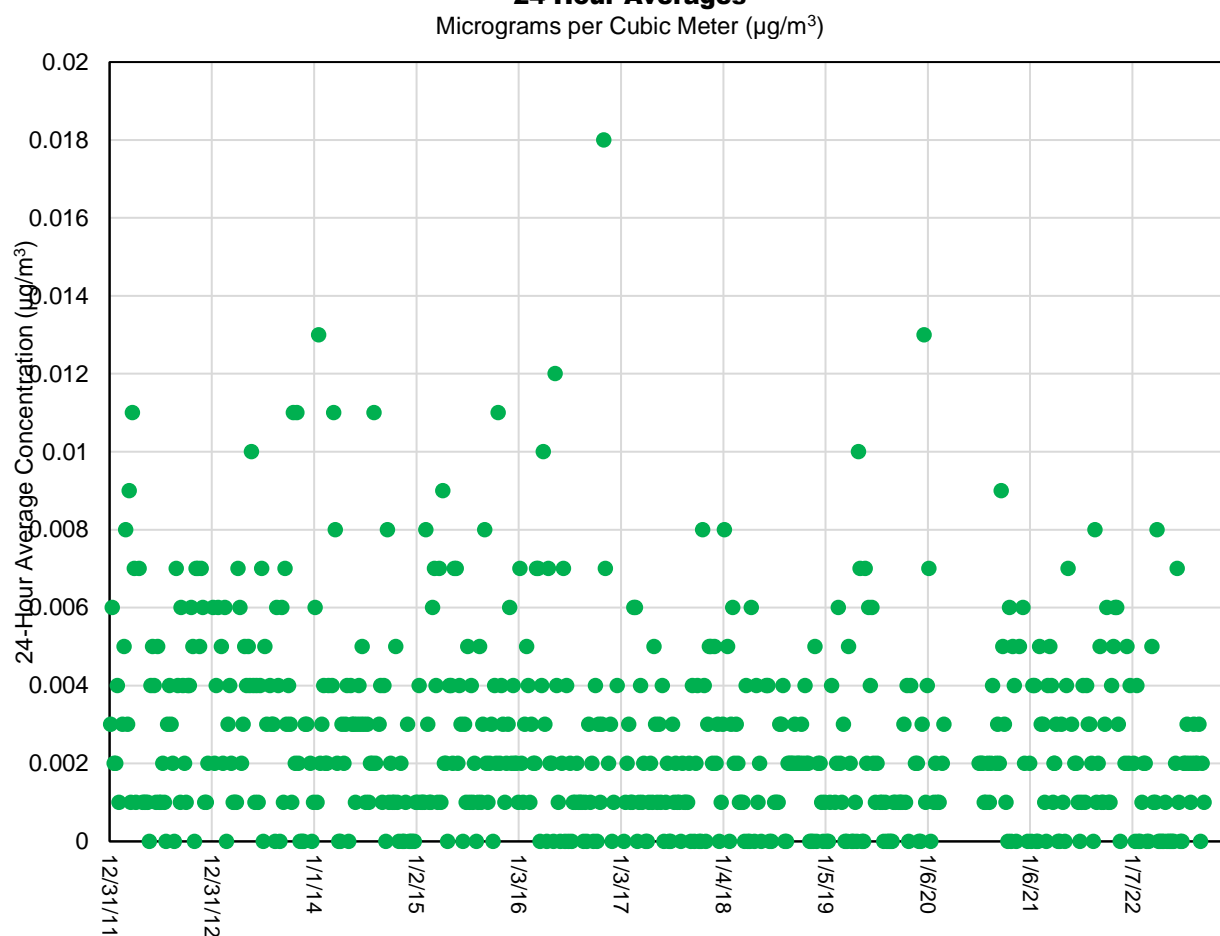
Averaging Period	Type	Level	Design Value
3 Months (Rolling)	Primary & Secondary	$0.15 \mu\text{g}/\text{m}^3$	Not to be exceeded

## LEAD AIR LEVELS IN 2022

In the 1980s NJDEP had more than 20 lead monitors around the state, including a few specifically located near lead-emitting facilities, such as a battery manufacturer in New Brunswick and a paint factory in Newark. By 2008, after years of decreasing concentrations, all of New Jersey's lead monitors were shut down. In March 2012, a lead monitor was installed at the Newark Firehouse monitoring station in accordance with new NAAQS requirements. Figure 9-1 presents all of the data from the Newark site since it started operating. Table 9-1 shows the rolling three-month averages for 2022.

Unfortunately, the Newark Firehouse monitoring station had to be shut down in late September. NJDEP expects to have a site reestablished somewhere in Newark by the end of 2023, and lead monitoring will resume then.

**Figure 9-1**  
**Lead Concentrations at Newark Firehouse in New Jersey, 2012-2022**  
**24-Hour Averages**



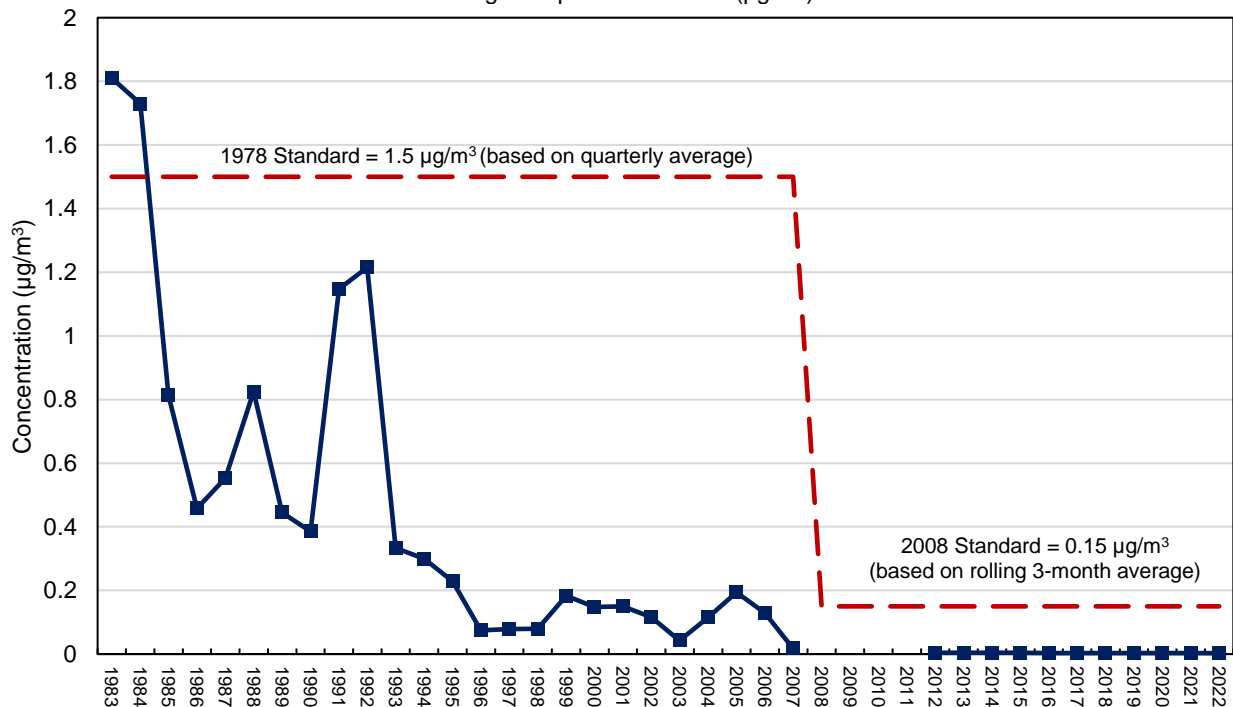
**Table 9-1**  
**2022 Lead Concentrations in New Jersey**  
**3-Month Rolling Averages**  
 Micrograms per Cubic Meter ( $\mu\text{g}/\text{m}^3$ )

3-Month Period	3-Month Average
November-January	0.003
December-February	0.002
January-March	0.001
February-April	0.001
March-May	0.001
April-June	0.001
May-July	0.001
June-August	0.002
July-September	0.002
August-October	No Data
September-November	No Data
October-December	No Data

## LEAD AIR TREND

The last exceedances of the NAAQS were in 1983 and 1984 (as shown in Figure 9-2), and the last exceedance of the NJAAQS was in 1992 (based on a rolling 3-month average; not shown in the graph). Since then, air concentrations of lead in New Jersey have dropped considerably. The highest annual 3-month rolling average concentrations at Newark Firehouse since 2012 have ranged from 0.003 to 0.004  $\mu\text{g}/\text{m}^3$ .

**Figure 9-2**  
**Statewide New Jersey Lead Trend, 1983-2022**  
**Highest 3-Month Averages**  
 Micrograms per Cubic Meter ( $\mu\text{g}/\text{m}^3$ )



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# 2022 Air Toxics Summary

New Jersey Department of Environmental Protection

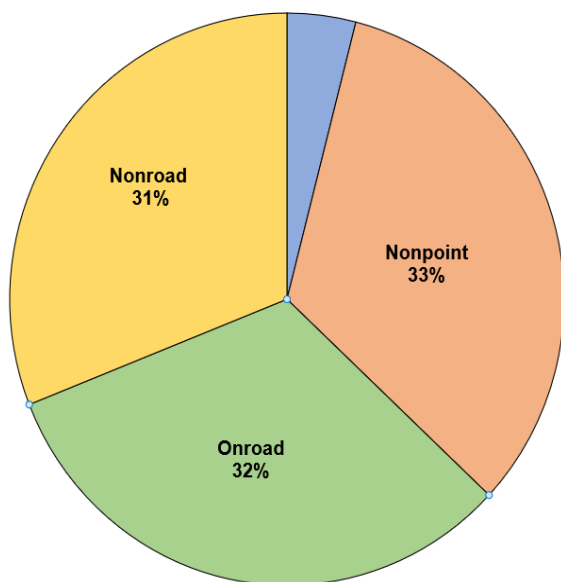
## INTRODUCTION

Air pollutants can be generally divided into two categories: criteria pollutants (ozone, sulfur dioxide, carbon monoxide, nitrogen dioxide, particulate matter, and lead); and air toxics. The criteria pollutants have been regulated at the national level since the 1970s. The United States Environmental Protection Agency (USEPA) has set National Ambient Air Quality Standards (NAAQS) for them, and states and local or tribal jurisdictions are required to plan and implement a process to bring and keep levels below the NAAQS, using monitoring, reporting, and control measures. Each of these pollutants is discussed in its own section (Sections 4 through 9) of this New Jersey Department of Environmental Protection (NJDEP) 2019 Air Quality Report.

Air toxics are all the other chemicals released into the air that have the potential to cause adverse health effects in humans. These effects cover a wide range of conditions, from lung irritation to birth defects to cancer. There are no NAAQS for these pollutants, but in 1990 the U.S. Congress directed the USEPA to begin addressing a list of 187 air toxics by developing control technology standards for specific types of sources that emit them. These air toxics are known as the Clean Air Act Hazardous Air Pollutants (HAPs). You can get more information about HAPs at the USEPA Air Toxics web site at <https://www.epa.gov/haps>. NJDEP also has several web pages dedicated to air toxics. They can be accessed at [www.nj.gov/dep/airtoxics](http://www.nj.gov/dep/airtoxics).

## SOURCES OF AIR TOXICS

**Figure 10-1**  
**2017 Air Toxics Emissions Source**  
**Estimates for New Jersey**



USEPA compiles a National Emissions Inventory (NEI) every three years. In addition to criteria pollutants and criteria precursors, it also collects information on emissions of hazardous air pollutants. The pie chart in Figure 10-1, taken from the most recent available NEI (for 2017), shows that mobile sources are the largest contributors of air toxics emissions in New Jersey.

In New Jersey, on-road mobile sources (cars and trucks) account for 32% of the air toxics emissions, and non-road mobile sources (airplanes, trains, construction equipment, lawnmowers, boats, dirt bikes, etc.) contribute an additional 31%. Nonpoint sources (residential, commercial, and small industrial sources) represent 33% of the inventory and point sources (such as factories and power plants) account for the remaining 4%.

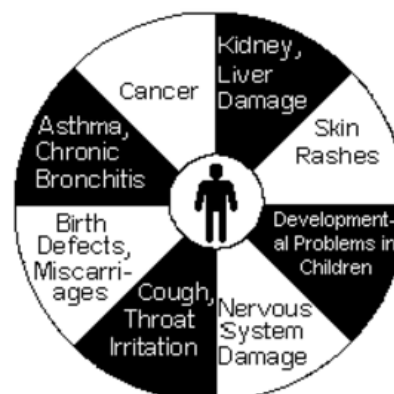
Source:  
<https://dep.nj.gov/airplanning/airtoxics/emissions2017/>



## HEALTH EFFECTS

People exposed to air toxics in significant amounts or for significant periods may have an increased chance of developing cancer or experiencing other serious health effects. The noncancer health effects can range from respiratory, neurological, reproductive, developmental, or immune system damage, to irritation and effects on specific organs (see Figure 10-2). In addition to inhalation exposure, there can be risks from the deposition of toxic pollutants onto soil or surface water. There, they can be taken up by humans directly, or by consuming exposed plants and animals.

**Figure 10-2**  
**Potential Effects of Air Toxics**

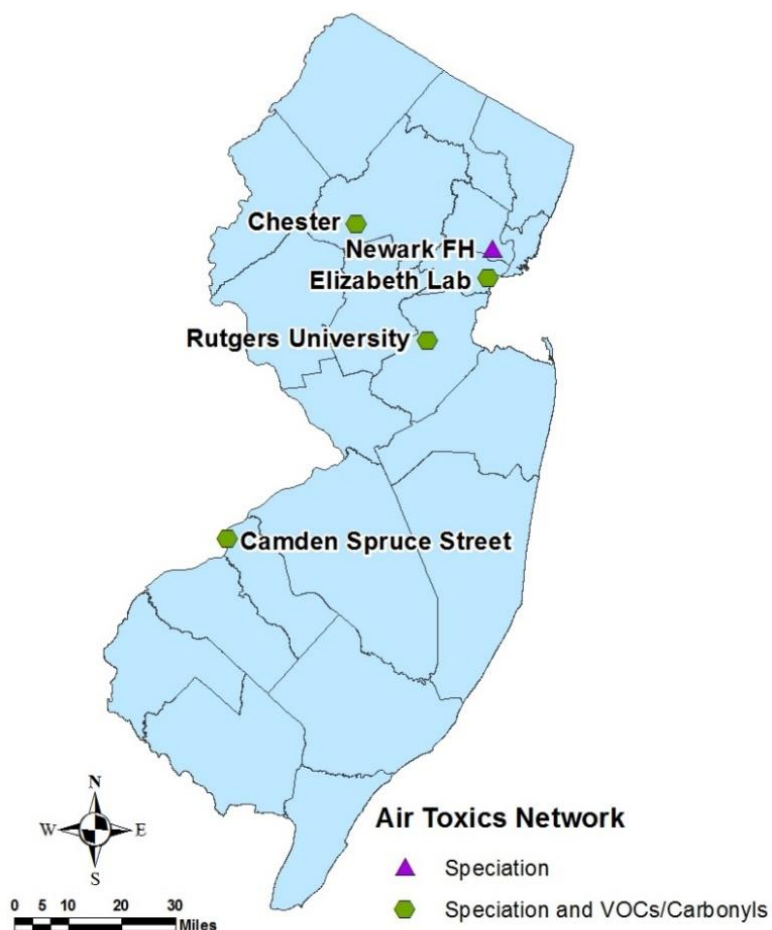


[www3.epa.gov/ttn/atw/3\\_90\\_024.html](http://www3.epa.gov/ttn/atw/3_90_024.html)

## AIR TOXICS MONITORING NETWORK

NJDEP has four air toxics monitoring sites that measure volatile organic compounds (VOCs) and carbonyls (a subset of VOCs that includes formaldehyde, acetaldehyde and other related compounds).

**Figure 10-3**  
**2022 Air Toxics Monitoring Network**



As shown in Figure 10-3, the monitors are located at Camden Spruce Street, Chester, Elizabeth Lab, and Rutgers University in East Brunswick. Toxic metals data are collected at the same four monitoring stations, plus Newark Firehouse (which unfortunately had to be shut down in September of 2022).

The Chester monitoring site is in rural Morris County, away from known sources, and serves as kind of a “background” monitor. The Rutgers University monitoring station is situated on Rutgers University agricultural lands in East Brunswick. The Elizabeth Lab monitoring station sits next to the Exit 13 tollbooths on the New Jersey Turnpike. The Camden Spruce Street monitoring station is located in an industrial urban setting. The Newark Firehouse monitoring station is in an urban residential area. More information about the air monitoring sites can be found in the Air Monitoring Network section and Appendix A of this annual Air Quality Report.

New Jersey’s VOC monitors are part of the Urban Air Toxics Monitoring Program (UATMP), sponsored by the USEPA. A 24-hour integrated air sample is collected in a canister every six days, and then sent to the USEPA contract laboratory (ERG, located in North Carolina) to be analyzed for VOCs and carbonyls. A previous monitoring site in Camden (officially called the Camden Lab site) had been measuring toxic VOCs for the UATMP since 1989, but was shut down in 2008 when NJDEP lost access to the location. A new monitoring station, the Camden Spruce Street monitoring site, became operational in 2013. The Elizabeth Lab site began measuring VOCs in 2000, and in July 2001 toxics monitoring began at the Chester and New Brunswick monitoring stations. In 2016 the New Brunswick VOC monitor was moved to the Rutgers University monitoring site, less than a mile away.

The map in Figure 10-3 also shows the monitoring sites that are part of USEPA’s Chemical Speciation Network (CSN). The CSN was established to analyze fine particulate matter (PM<sub>2.5</sub>) for toxic metals, elements, ions and carbon constituents. Filters are collected every three or six days and sent to a national lab for analysis. Sampling began in 2001 at Camden, Chester, Elizabeth Lab and New Brunswick. The Newark Firehouse site was added in 2010 (and shut down in 2022), and the New Brunswick CSN monitor was moved to Rutgers University in 2016. See Appendix B for more information.

## **AIR TOXICS LEVELS IN 2022**

Annual average concentrations of VOCs and carbonyls for the four New Jersey monitoring sites are shown in Table 10-1. All values are in micrograms per cubic meter (µg/m<sup>3</sup>). Values in parts per billion by volume (ppbv), as well as other statistics and risk estimates, can be found in Tables 10-4 through 10-7. The ppbv units are more common in air monitoring, while µg/m<sup>3</sup> units are generally used in air dispersion modeling and health studies.

Detection limit information and the health benchmarks used in the analysis can be found in Table 10-8. Some compounds are not always detected in the samples analyzed by the lab; however, this does not mean they are not present in the air below the detection limit level. For chemicals detected in less than 50% of the samples, there is significant uncertainty in the calculated averages. Median values (the value of the middle sample when the results are ranked) are reported in Tables 10-4 through 10-7 along with the mean (average) concentrations, because for some compounds only a single value or a few very high values were recorded. These high values could skew the mean concentrations, but would have less effect on the median value. In such cases, the median value may be a better indicator of long-term exposure concentrations.

**Table 10-1**  
**2022 Summary of Toxic Volatile Organic Compounds Monitored in New Jersey**  
**Annual Average Concentrations**  
Micrograms per Cubic Meter ( $\mu\text{g}/\text{m}^3$ )

	Pollutant	Synonym	HAP	CAS No.	Camden	Chester	Elizabeth	Rutgers
1	Acetaldehyde		*	75-07-0	2.470	0.964	1.383	1.277
2	Acetone			67-64-1	2.307	2.159	2.285	2.182
3	Acetonitrile		*	75-05-8	0.398	2.083	0.794	0.605
4	Acetylene			74-86-2	0.726	0.347	1.726	0.551
5	Acrolein		*	107-02-8	0.851	0.836	0.868	1.450
6	Acrylonitrile		*	107-13-1	0.007	0.002	0.004	0.005
7	tert-Amyl Methyl Ether			994-05-8	0.001	0.002	0.002	0.002
8	Benzaldehyde			100-52-7	0.204	0.096	0.084	0.063
9	Benzene		*	71-43-2	0.655	0.305	0.704	0.389
10	Bromochloromethane			74-97-5	0.002	0.004	0.002	0.004
11	Bromodichloromethane			75-27-4	0.007	0.006	0.007	0.009
12	Bromoform		*	75-25-2	0.020	0.023	0.025	0.023
13	Bromomethane	Methyl bromide	*	74-83-9	0.677	0.038	0.038	0.036
14	1,3-Butadiene		*	106-99-0	0.045	0.014	0.057	0.025
15	Butyraldehyde			123-72-8	0.241	0.079	0.100	0.116
16	Carbon Disulfide		*	75-15-0	0.065	0.061	0.110	0.061
17	Carbon Tetrachloride		*	56-23-5	0.442	0.469	0.438	0.433
18	Chlorobenzene		*	108-90-7	0.002	0.004	0.003	0.003
19	Chloroethane	Ethyl chloride	*	75-00-3	0.033	0.031	0.031	0.045
20	Chloroform		*	67-66-3	0.114	0.096	0.132	0.122
21	Chloromethane	Methyl chloride	*	74-87-3	1.004	0.990	1.016	0.994
22	Chloroprene	2-Chloro-1,3-butadiene	*	126-99-8	0.001	0.002	0.001	0.002
23	Crotonaldehyde			123-73-9	0.173	0.017	0.013	0.014
24	Dibromochloromethane	Chlorodibromomethane		124-48-1	0.007	0.008	0.008	0.009
25	1,2-Dibromoethane	Ethylene dibromide	*	106-93-4	0.003	0.005	0.004	0.005
26	m-Dichlorobenzene	1,3-Dichlorobenzene		541-73-1	0.003	0.005	0.004	0.006
27	o-Dichlorobenzene	1,2-Dichlorobenzene		95-50-1	0.004	0.005	0.006	0.006
28	p-Dichlorobenzene	1,4-Dichlorobenzene	*	106-46-7	0.041	0.014	0.035	0.023
29	Dichlorodifluoromethane			75-71-8	2.553	2.477	2.496	2.491
30	1,1-Dichloroethane	Ethylidene dichloride	*	75-34-3	0.001	0.003	0.002	0.003
31	1,2-Dichloroethane	Ethylene dichloride	*	107-06-2	0.077	0.057	0.058	0.059
32	1,1-Dichloroethylene	Vinylidene chloride	*	75-35-4	0.004	0.008	0.005	0.006
33	cis-1,2-Dichloroethylene	cis-1,2-Dichloroethene		156-59-2	0.001	0.002	0.001	0.002
34	trans-1,2-Dichloroethylene	trans-1,2-Dichloroethene		156-60-5	0.017	0.027	0.012	0.009
35	Dichloromethane	Methylene chloride	*	75-09-2	0.590	0.475	0.684	0.570
36	1,2-Dichloropropane	Propylene dichloride	*	78-87-5	0.002	0.003	0.002	0.002

Continued

- Values in ppbv can be found in Tables 10-4 through 10-7.
- Values in *italics* indicate that fewer than 50% of samples had detectable levels.
- **Zero** indicates that there were no samples with reportable levels.
- HAP = Hazardous air pollutant as listed in the Clean Air Act.

**Table 10-1 (continued)**  
**2022 Summary of Toxic Volatile Organic Compounds Monitored in New Jersey**  
**Annual Average Concentrations**  
Micrograms per Cubic Meter ( $\mu\text{g}/\text{m}^3$ )

	Pollutant	Synonym	HAP	CAS No.	Camden	Chester	Elizabeth	Rutgers
37	cis-1,3-Dichloropropylene	cis-1,3-Dichloropropene	*	10061-01-5	0.001	0.003	0.001	0.002
38	trans-1,3-Dichloropropylene	trans-1,3-Dichloropropene	*	10061-02-6	0.001	0.002	0.001	0.001
39	Dichlorotetrafluoroethane	Freon 114		76-14-2	0.123	0.126	0.106	0.126
40	Ethyl Acrylate		*	140-88-5	0.001	0	0.001	0.001
41	Ethylbenzene		*	100-41-4	0.246	0.065	0.259	0.108
42	Ethyl tert-Butyl Ether	tert-Butyl ethyl ether		637-92-3	0.003	0.007	0.001	0.010
43	Formaldehyde		*	50-00-0	3.450	1.874	2.346	2.010
44	Hexachlorobutadiene	Hexachloro-1,3-butadiene	*	87-68-3	0.006	0.010	0.008	0.010
45	Hexaldehyde	Hexanaldehyde		66-25-1	0.213	0.236	0.276	0.183
46	Methyl Ethyl Ketone	MEK, 2-Butanone		78-93-3	0.296	0.258	0.291	0.296
47	Methyl Isobutyl Ketone	MIBK	*	108-10-1	0.141	0.086	0.130	0.094
48	Methyl Methacrylate		*	80-62-6	0.030	0.001	0.005	0.001
49	Methyl tert-Butyl Ether	MTBE	*	1634-04-4	0.001	0.002	0.001	0.002
50	n-Octane			111-65-9	0.231	0.088	0.304	0.108
51	Propionaldehyde		*	123-38-6	0.405	0.244	0.258	0.273
52	Propylene			115-07-1	3.220	0.713	4.119	1.016
53	Styrene		*	100-42-5	0.301	0.051	0.113	0.051
54	1,1,2,2-Tetrachloroethane		*	79-34-5	0.003	0.004	0.003	0.004
55	Tetrachloroethylene	Perchloroethylene	*	127-18-4	0.150	0.052	0.101	0.061
56	Toluene		*	108-88-3	1.795	0.376	1.470	0.629
57	1,2,4-Trichlorobenzene		*	120-82-1	0.012	0.012	0.011	0.015
58	1,1,1-Trichloroethane	Methyl chloroform	*	71-55-6	0.015	0.013	0.015	0.014
59	1,1,2-Trichloroethane		*	79-00-5	0.002	0.004	0.011	0.004
60	Trichloroethylene		*	79-01-6	0.027	0.028	0.033	0.030
61	Trichlorofluoromethane			75-69-4	1.809	1.344	1.364	1.357
62	Trichlorotrifluoroethane	1,1,2-Trichloro-1,2,2-trifluoroethane		76-13-1	0.584	0.581	0.585	0.586
63	1,2,4-Trimethylbenzene			95-63-6	0.314	0.034	0.250	0.068
64	1,3,5-Trimethylbenzene			108-67-8	0.090	0.016	0.066	0.026
65	Valeraldehyde			110-62-3	0.179	0.087	0.094	0.099
66	Vinyl chloride		*	75-01-4	0.009	0.003	0.003	0.004
67	m,p-Xylene		*	108-38-3 106-42-3	0.723	0.131	0.708	0.255
68	o-Xylene		*	95-47-6	0.303	0.058	0.294	0.106

- Values in ppbv can be found in Tables 10-4 through 10-7.
- Values in *italics* indicate that fewer than 50% of samples had detectable levels.
- **Zero** indicates that there were no samples with reportable levels.
- HAP = Hazardous air pollutant as listed in the Clean Air Act.

## ESTIMATING HEALTH RISK

The effects on human health resulting from exposure to specific air toxics can be estimated by using chemical-specific **health benchmarks**. These are based on toxicity values developed by the USEPA and other agencies, using animal or human health studies. For carcinogens, which are chemicals suspected of causing cancer, the health benchmark is the concentration of the pollutant that corresponds to a one-in-one-million increase in the risk of getting cancer if a person was to breathe that concentration over his or her entire lifetime. The health benchmark for a noncarcinogen is the air concentration at which no adverse health effect is expected to occur, even if a person is exposed to that concentration on a daily basis for a lifetime (this is also known as a reference concentration). Because of a lack of toxicity studies, not all air toxics have health benchmarks. The health benchmarks used to evaluate the VOCs and carbonyls monitored in New Jersey are listed in Table 10-8. Health benchmarks for specific toxic metals and elements are shown in Table 10-3. These are all based on long-term exposure.

A **risk ratio** can be used to quantify risk from exposure to a specific chemical. This is calculated by dividing the annual average air concentration of a chemical by its long-term health benchmark. If the risk ratio is less than one, the air concentration should not pose a health risk. If it is greater than one, it may be of concern. The risk ratio also indicates how much higher or lower the estimated air concentration is compared to the health benchmark. Identifying problematic chemicals helps regulatory agencies focus their efforts to reduce emissions and exposure.

Air toxics with risk ratios greater than one for at least one monitoring site are summarized in Table 10-2. Acrolein and formaldehyde showed the highest risk statewide. Other pollutants above health benchmarks at all four sites were acetaldehyde, benzene, carbon tetrachloride, chloroform, chloromethane (methyl chloride), 1,2-dibromoethane (ethylene dibromide), and 1,2-dichloroethane (ethylene dichloride). 1,3-Butadiene had a risk ratio slightly greater than one at all sites except Chester and Rutgers.

**Table 10-2**  
**Monitored Air Toxics with Risk Ratios Greater Than One in 2022**

	Pollutant	CAS No.	Annual Average Risk Ratio			
			Camden	Chester	Elizabeth	Rutgers
1	Acetaldehyde	75-07-0	5.5	2.1	3.1	2.8
2	Acrolein	107-02-8	<i>42.6</i>	<i>41.8</i>	<i>43.4</i>	<i>72.5</i>
3	Benzene	71-43-2	5.0	2.3	5.4	3.0
4	1,3-Butadiene	106-99-0	1.4	0.4	1.7	0.8
5	Carbon Tetrachloride	56-23-5	2.6	2.8	2.6	2.5
6	Chloroform	67-66-3	2.7	2.2	3.1	2.8
7	Chloromethane	74-87-3	1.8	1.8	1.8	1.8
8	1,2-Dibromoethane	<i>106-93-4</i>	<b>1.8</b>	<b>3.2</b>	<b>2.5</b>	<b>3.1</b>
9	1,2-Dichloroethane	107-06-2	2.0	1.5	1.5	1.6
10	Formaldehyde	50-00-0	44.8	24.3	30.5	26.1

- Risk ratio = annual average air concentration/health benchmark
- Health benchmarks in italics have a noncancer endpoint. See section on “Estimating Health Risk” for more information.
- Health benchmarks in bold are based on less than 50% of samples detected and are highly uncertain.

Table 10-3 presents annual average concentrations and health benchmarks for certain toxic metals and elements that can be found in fine particles. This fine particulate matter is analyzed through USEPA's Chemical Speciation Network (CSN). No risk ratios were calculated, because most of the chemicals were below the detection limit and so the resulting average concentrations are highly uncertain. Additional data from the CSN monitors can be found in Appendix B (Fine Particulate Speciation Summary) of the annual Air Quality Report.

**Table 10-3**  
**2022 Summary of Toxic Metals and Elements Monitored in New Jersey**  
**Annual Average Concentrations & Health Benchmarks**  
Micrograms per Cubic Meter ( $\mu\text{g}/\text{m}^3$ )

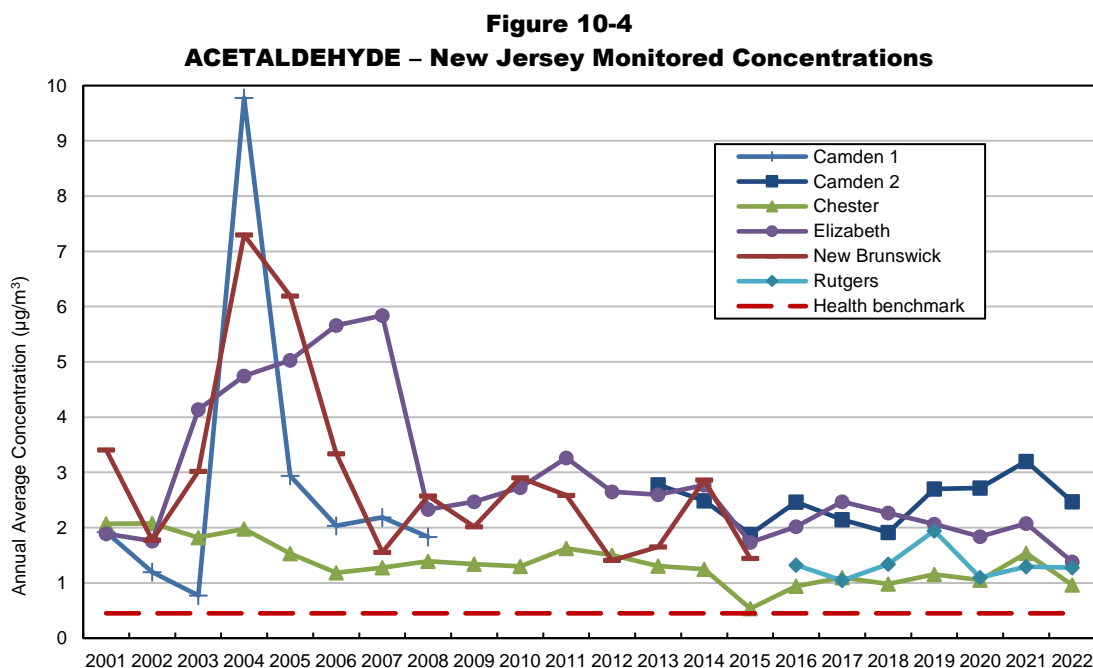
Pollutant	HAP	Camden	Chester	Elizabeth	Newark	Rutgers	Health Benchmark
Antimony	*	<i>0.001</i>	0.003	0.002	0.002	0.001	<i>0.2</i>
Arsenic	*	<i>0</i>	<i>0</i>	<i>0</i>	<i>0</i>	<i>0</i>	0.00023
Cadmium	*	<i>0</i>	0.014	0.001	0.002	0.0002	0.00024
Chlorine	*	0.133	<i>0.005</i>	0.019	0.031	0.019	<i>0.2</i>
Chromium <sup>a</sup>	*	0.002	0.001	0.002	0.001	0.002	0.000083
Cobalt	*	<i>0</i>	<i>0.0001</i>	<i>0</i>	<i>0</i>	<i>0</i>	0.00014
Lead	*	0.006	0.001	0.002	0.003	0.003	0.083
Manganese	*	0.006	0.001	0.002	0.001	0.003	<i>0.05</i>
Nickel <sup>b</sup>	*	0.001	0.0003	0.001	0.001	0.001	0.0021
Phosphorus	*	0.0004	0.0001	0.001	0.0005	0.001	<i>0.07</i>
Selenium	*	<i>0.0003</i>	<i>0.0002</i>	<i>0.0001</i>	<i>0.0003</i>	0.0002	<i>20</i>
Silicon		0.074	0.025	0.052	0.052	0.043	3
Vanadium		<i>0.0001</i>	<i>0.0001</i>	<i>0.0001</i>	<i>0.0002</i>	<i>0.0001</i>	<i>0.1</i>

- Annual average values in italics had fewer than 50% of samples detectable, so the averages are highly uncertain.
- HAP = Hazardous air pollutant listed in the Clean Air Act.
- Health benchmarks in italics have a noncancer endpoint. See section on “Estimating Health Risk” for more information.
- a) Chromium’s health benchmark is based on carcinogenicity of hexavalent chromium (Cr+6). It is not known how much of the chromium measured by the monitor is hexavalent.
- b) Nickel’s health benchmark is based on specific nickel compounds. It is not known how much of the nickel measured by the monitor is in that form.

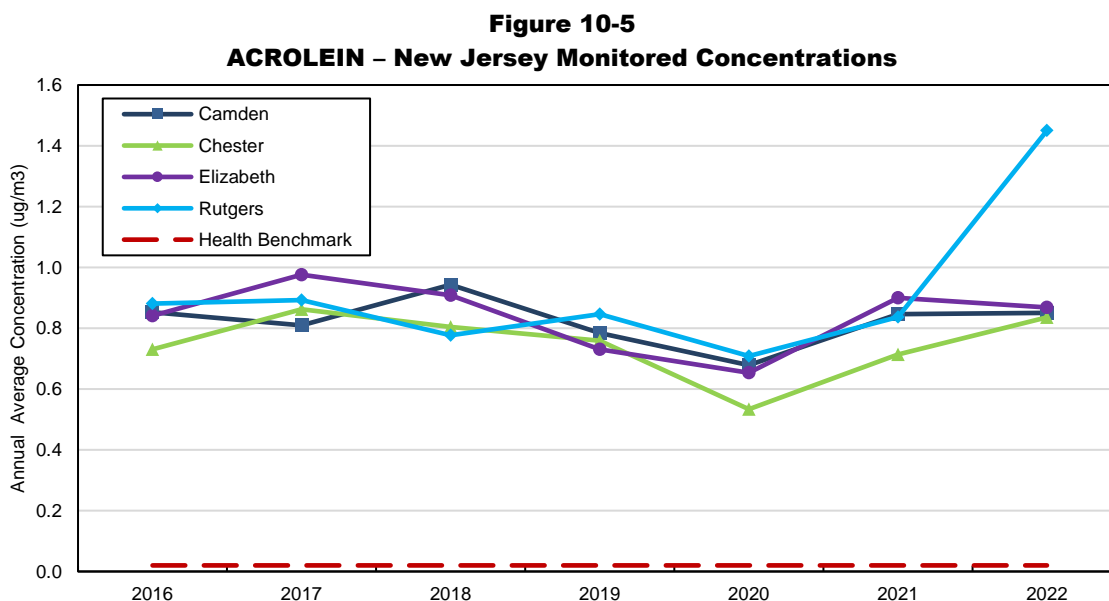
## TRENDS AND COMPARISONS

Monitoring of air toxics in New Jersey has been going on since a UATMP site was established in Camden in 1989. Sampling and analysis methods continue to evolve, most notably with improvements in the ability to detect chemicals at lower concentrations. Figures 10-4 through 10-11 present data for some of the VOCs that have been measured for a number of years and are at levels of concern (above their health benchmarks). As mentioned previously, the first toxics monitoring site in Camden (Camden Lab) was shut down in 2008. It is identified in Figures 10-4 through 10-11 as “Camden 1.” The new Camden site (Camden Spruce Street), located about two miles from the old site, is designated “Camden 2” in the trend graphs. The New Brunswick monitoring station was shut down in 2016, and the monitors were moved less than a mile to the Rutgers University site.

According to USEPA's 2014 National Air Toxics Assessment (NATA), **acetaldehyde** concentrations in New Jersey (Figure 10-4) are primarily influenced by secondary formation, a process in which chemicals in the air react with each other and are transformed into other chemicals. Mobile sources also contribute to ambient levels. In 2003, no data was collected in Camden after September, which probably had an influence on the low annual average for that year. In 2004, high levels of acetaldehyde were measured over a number of weeks at both Camden and New Brunswick.

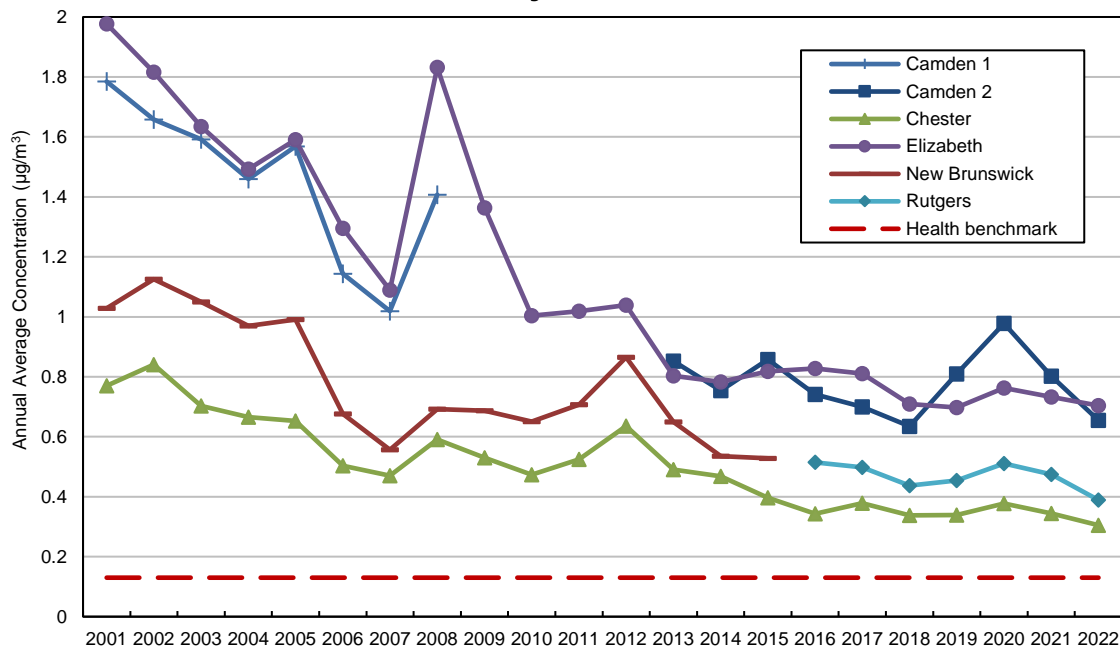


**Acrolein** is sometimes used as a pesticide and to make other chemicals, but by far most of it is formed in the air from burning fossil fuels (gasoline, oil) and organic matter (including cigarettes). It is not known if it causes cancer, but it can have detrimental effects on the respiratory system. Prior to 2016, there were concerns that the laboratory methods used to measure acrolein were inadequate. The analysis methods have since been improved, and the recent data is presented in Figure 10-5. The increase in the Rutgers average is the result of an exceptionally high value on 8/27/2022.

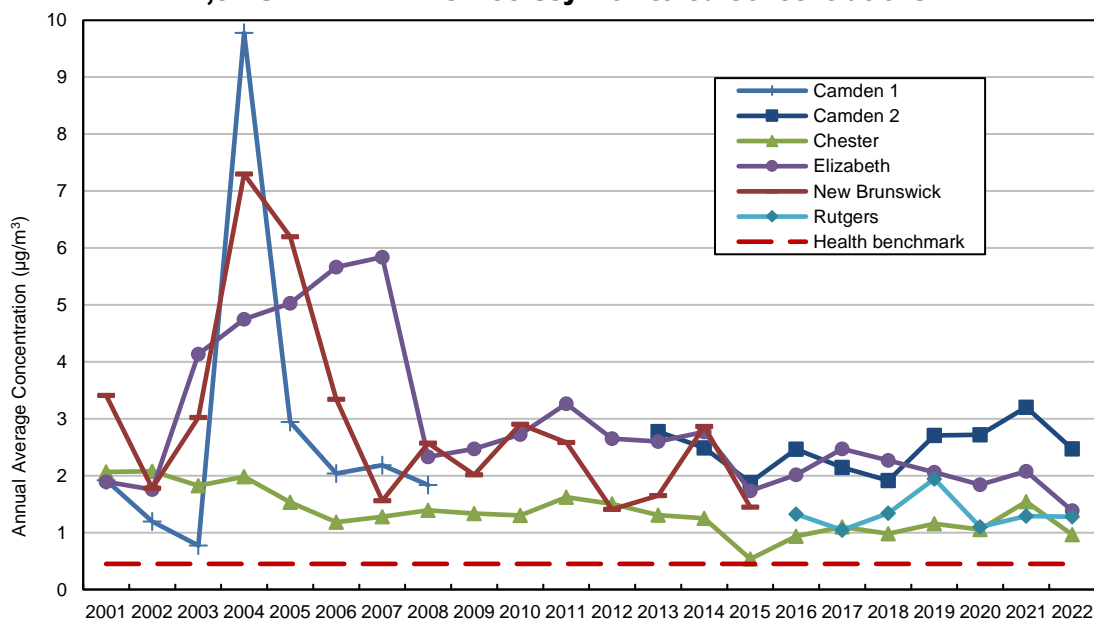


Figures 10-6 and 10-7 show a general decrease in **benzene** and **1,3-butadiene** concentrations over the past decade. Over 50% of New Jersey's ambient benzene and 1,3-butadiene comes from on-road mobile sources, and about 20% comes from non-road mobile sources.

**Figure 10-6**  
**BENZENE – New Jersey Monitored Concentrations**



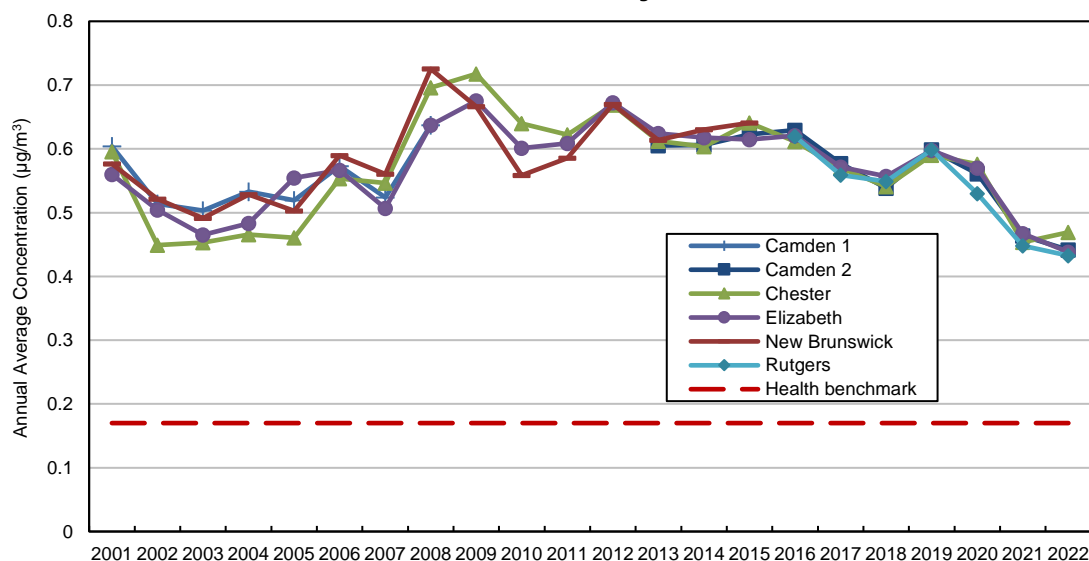
**Figure 10-7**  
**1,3-BUTADIENE – New Jersey Monitored Concentrations**





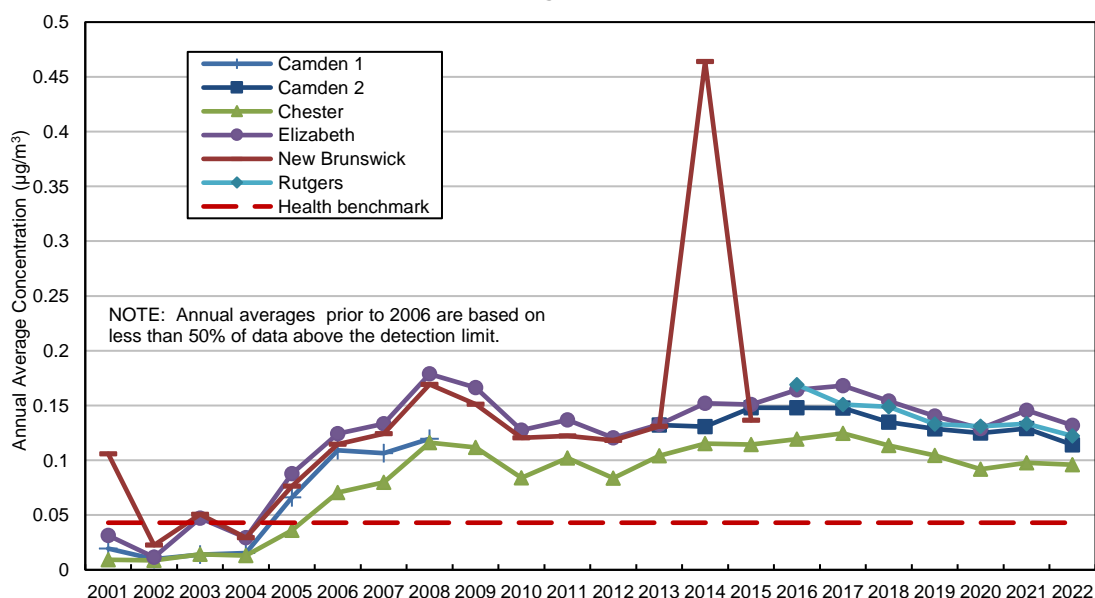
**Carbon tetrachloride** (Figure 10-8) was once used widely as a degreaser, household cleaner, propellant, refrigerant, and fumigant. It has been phased out of most production and use because of its toxicity and because it depletes stratospheric ozone. However, about 100 tons are still emitted annually by industry in the U.S., although no emissions have been reported in New Jersey for years. It degrades slowly in the environment, so it can be transported from other areas, and levels in the air can remain relatively steady for a long time.

**Figure 10-8**  
**CARBON TETRACHLORIDE – New Jersey Monitored Concentrations**

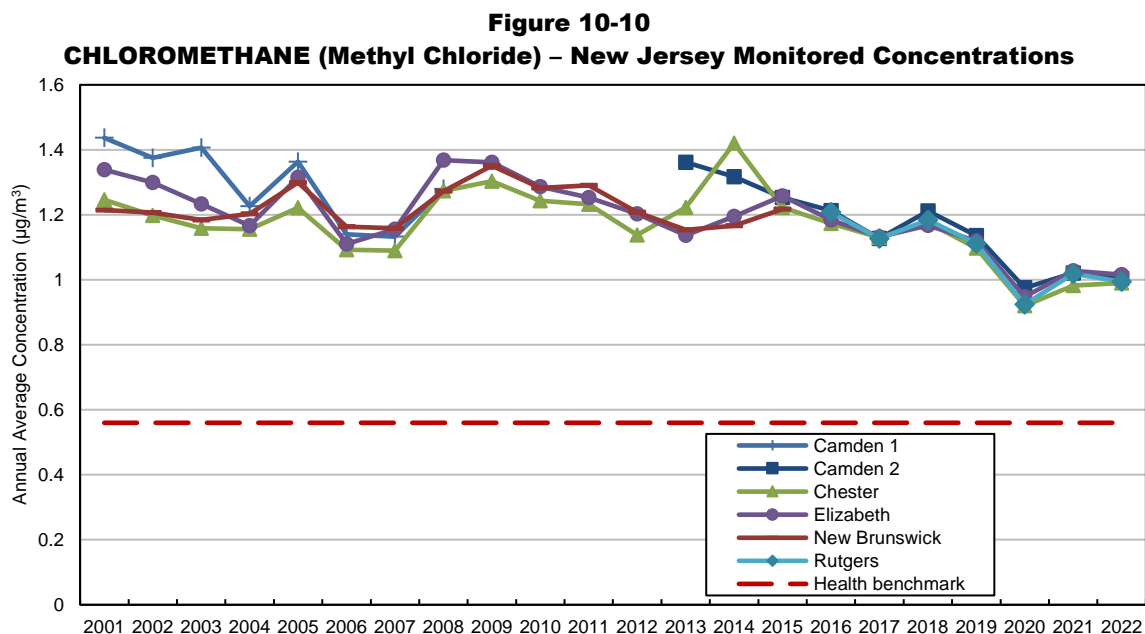


Some of the increase in the **chloroform** concentration shown in Figure 10-9 is believed to be from improvements in the laboratory detection limit. The high annual average concentration for New Brunswick in 2014 is attributable to a period of high values in May and June. Point and nonpoint sources (related to waste disposal) are the major contributors to ambient chloroform levels in New Jersey. Chloroform can be formed in small amounts by chlorination of water. It breaks down slowly in ambient air.

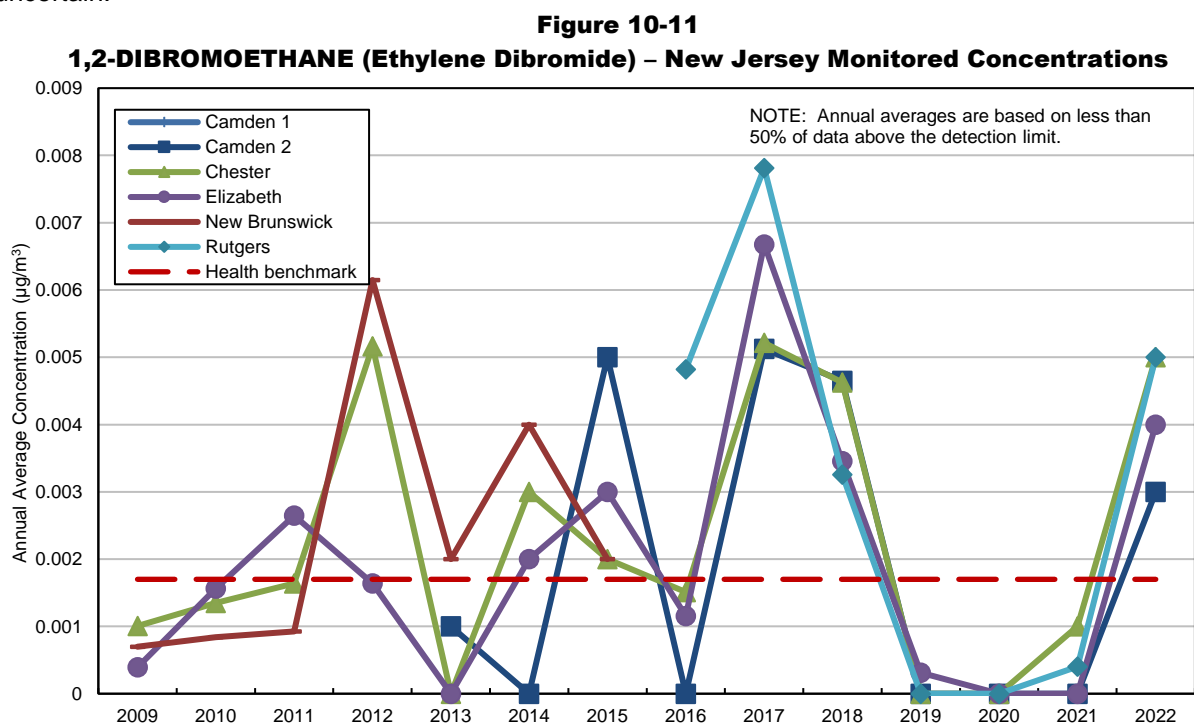
**Figure 10-9**  
**CHLOROFORM – New Jersey Monitored Concentrations**



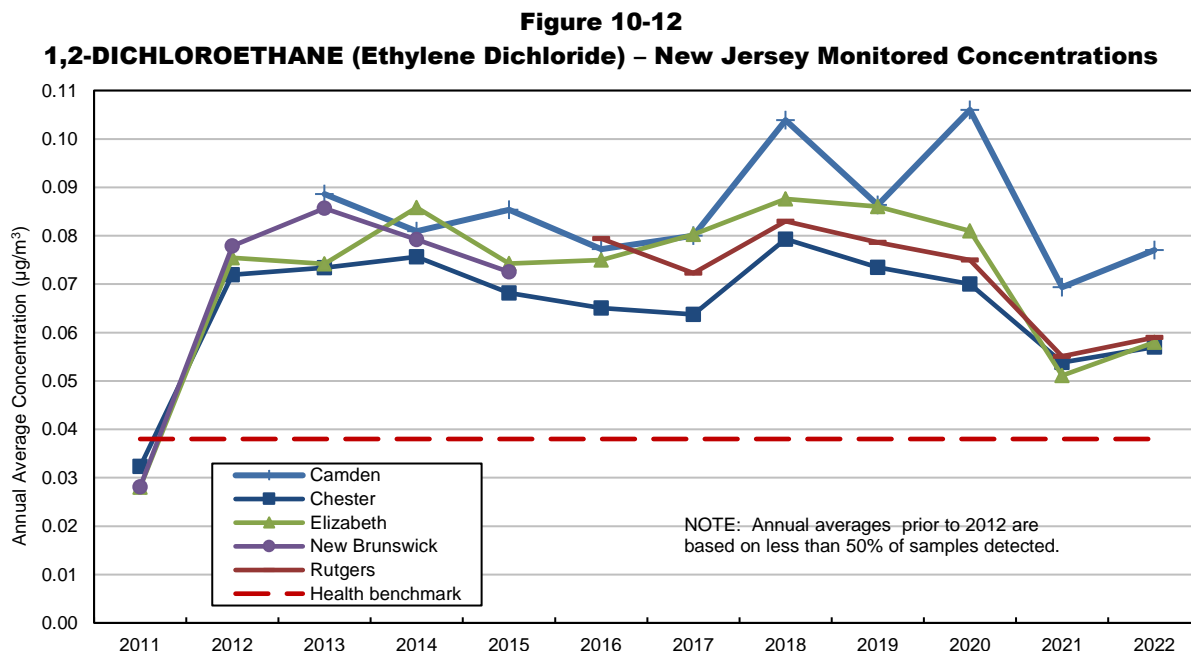
As seen in Figure 10-10, **chloromethane** (also known as methyl chloride) levels have remained relatively stable from year to year, and all the sites show similar levels. It was once commonly used as a refrigerant and in the chemical industry, but was phased out because of its toxicity. According to the USEPA's 2014 National Emissions Inventory, about 73% of the chloromethane in New Jersey's air is from nonpoint sources, primarily waste disposal, while 27% is from point sources.



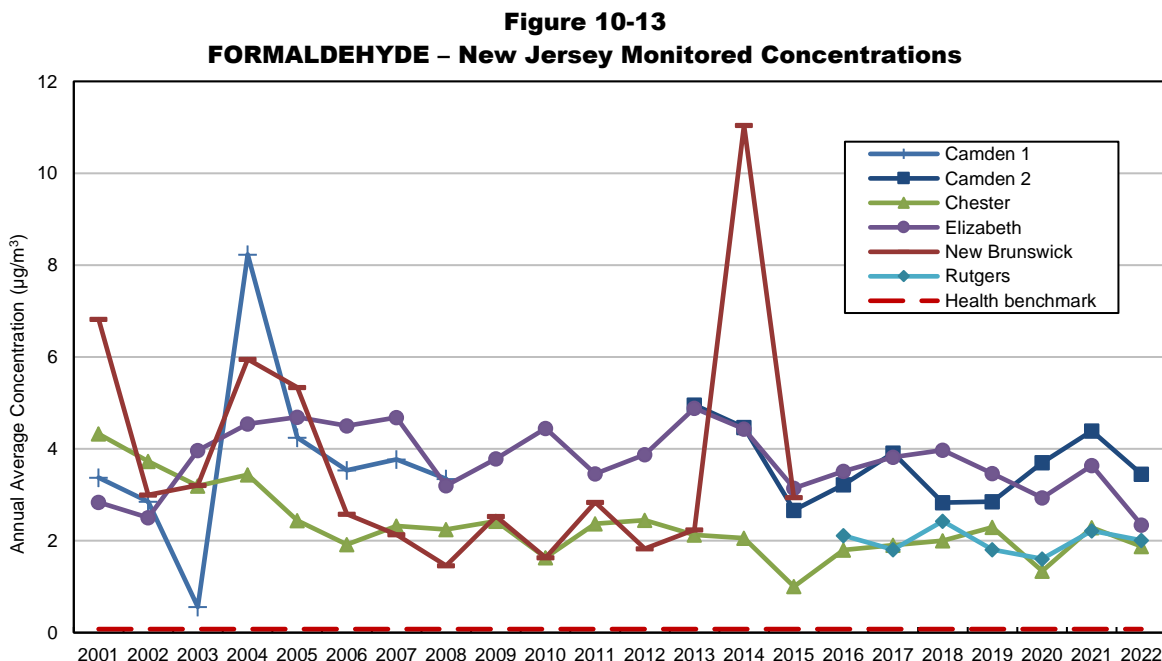
**1,2-Dibromoethane** (also known as ethylene dibromide) (Figure 10-11) is currently used as a pesticide in the treatment of felled logs for bark beetles and termites, and control of wax moths in beehives. It was once used as an additive to leaded gasoline and as a soil and grain fumigant, but those uses have been banned by USEPA. Most of the monitoring results fall below the detection limit, so the data in the graph is fairly uncertain.



**1,2-Dichloroethane** (also known as ethylene dichloride) (Figure 10-12) is primarily used in the production of chemicals, as a solvent, dispersant and wetting and penetrating agent. The increase in concentrations after 2011 is related to an improvement in the laboratory detection limit, resulting in over 90% of samples having detectable levels. The 2014 National Emissions Inventory estimates that 93% of 1,2-dichloroethane in New Jersey's air is from point sources, and 7% from nonpoint sources.



**Formaldehyde** (Figure 10-13) is a ubiquitous pollutant that is often found at higher concentrations indoors rather than outdoors because of its use in many consumer goods. It is used in the production of fertilizer, paper, plywood, urea-formaldehyde resins, and many other products. In New Jersey the primary emitters of formaldehyde are mobile sources, although high outdoor levels are mostly the result of secondary formation. In 2014, concentrations at the New Brunswick site were consistently higher than at the other monitors, although levels subsequently dropped to the range of the other monitoring sites.



**Table 10-4**  
**CAMDEN SPRUCE STREET – 2022 NJ Air Toxics Monitoring Data**

Pollutant		Annual Mean	Annual Median	24-Hour Max.	Annual Mean	Annual Median	24-Hour Max.	% Detected	Annual Risk Ratio
		ppbv			µg/m³				
1	Acetaldehyde	1.371	1.300	2.880	2.470	2.342	5.189	100	5.5
2	Acetone	0.971	0.968	2.230	2.307	2.299	5.297	100	0.0001
3	Acetonitrile	0.237	0.197	1.550	0.398	0.330	2.603	82	0.01
4	Acetylene	0.682	0.567	3.280	0.726	0.604	3.493	100	
5	Acrolein	0.371	0.319	0.959	0.851	0.731	2.199	100	42.6
6	Acrylonitrile	0.003	0	0.040	0.007	0	0.087	22	0.4
7	tert-Amyl Methyl Ether	0.0003	0	0.011	0.001	0	0.045	7	
8	Benzaldehyde	0.047	0.031	0.359	0.204	0.136	1.558	100	
9	Benzene	0.205	0.180	0.652	0.655	0.575	2.083	100	5.0
10	Bromochloromethane	0.0004	0	0.013	0.002	0	0.070	23	0.0001
11	Bromodichloromethane	0.001	0	0.016	0.007	0	0.106	42	0.3
12	Bromoform	0.002	0.002	0.015	0.020	0.017	0.153	96	0.02
13	Bromomethane	0.174	0.010	4.530	0.677	0.041	17.590	100	0.1
14	1,3-Butadiene	0.021	0.018	0.064	0.045	0.040	0.141	100	1.4
15	Butyraldehyde	0.082	0.081	0.154	0.241	0.238	0.454	100	
16	Carbon Disulfide	0.021	0.015	0.318	0.065	0.045	0.990	98	0.0001
17	Carbon Tetrachloride	0.070	0.075	0.103	0.442	0.474	0.648	100	2.6
18	Chlorobenzene	0.0004	0	0.012	0.002	0	0.057	7	0.000002
19	Chloroethane	0.012	0.009	0.053	0.033	0.024	0.141	95	0.000003
20	Chloroform	0.023	0.023	0.050	0.114	0.113	0.245	96	2.7
21	Chloromethane	0.486	0.496	0.611	1.004	1.024	1.262	100	1.8
22	Chloroprene	0.0003	0	0.016	0.001	0	0.056	5	0.6
23	Crotonaldehyde	0.060	0.003	3.280	0.173	0.007	9.403	53	
24	Dibromochloromethane	0.001	0.001	0.012	0.007	0.004	0.106	80	0.2
25	1,2-Dibromoethane	0.0004	0	0.010	0.003	0	0.078	23	1.8
26	m-Dichlorobenzene	0.001	0	0.010	0.003	0	0.060	43	
27	o-Dichlorobenzene	0.001	0.0003	0.010	0.004	0.002	0.062	57	0.00002
28	p-Dichlorobenzene	0.007	0.005	0.038	0.041	0.033	0.227	100	0.5
29	Dichlorodifluoromethane	0.516	0.513	0.653	2.553	2.537	3.229	100	0.03
30	1,1-Dichloroethane	0.0003	0	0.012	0.001	0	0.048	5	0.002
31	1,2-Dichloroethane	0.019	0.016	0.055	0.077	0.066	0.224	100	2.0
32	1,1-Dichloroethylene	0.001	0.001	0.004	0.004	0.005	0.016	71	0.00002
33	cis-1,2-Dichloroethylene	0.0002	0	0.012	0.001	0	0.049	2	
34	trans-1,2-Dichloroethylene	0.004	0.003	0.076	0.017	0.011	0.301	80	
35	Dichloromethane	0.170	0.120	0.590	0.590	0.417	2.050	100	0.01
36	1,2-Dichloropropane	0.001	0	0.017	0.002	0	0.076	5	0.02
37	cis-1,3-Dichloropropylene	0.0002	0	0.011	0.001	0	0.049	2	0.004
38	trans-1,3-Dichloropropylene	0.0002	0	0.011	0.001	0	0.052	2	0.004
39	Dichlorotetrafluoroethane	0.018	0.017	0.029	0.123	0.121	0.206	100	
40	Ethyl Acrylate	0.0003	0	0.006	0.001	0.000	0.025	5	0.0002
41	Ethylbenzene	0.057	0.039	0.254	0.246	0.168	1.103	100	0.6
42	Ethyl tert-Butyl Ether	0.002	0	0.013	0.003	0	0.023	32	
43	Formaldehyde	2.809	2.600	7.650	3.450	3.193	9.395	100	44.8
44	Hexachlorobutadiene	0.001	0.0003	0.012	0.006	0.003	0.124	71	0.1
45	Hexaldehyde	0.052	0.045	0.159	0.213	0.186	0.651	100	

Continued

**Table 10-4 (continued)**  
**CAMDEN SPRUCE STREET – 2022 NJ Air Toxics Monitoring Data**

Pollutant		Annual Mean	Annual Median	24-Hour Max.	Annual Mean	Annual Median	24-Hour Max.	% Detected	Annual Mean Risk Ratio
		ppbv			µg/m³				
46	Methyl Ethyl Ketone	0.100	0.092	0.311	0.296	0.272	0.917	100	0.0001
47	Methyl Isobutyl Ketone	0.034	0.027	0.191	0.141	0.110	0.782	96	0.00005
48	Methyl Methacrylate	0.007	0	0.300	0.030	0	1.228	6	0.00004
49	Methyl tert-Butyl Ether	0.0002	0	0.004	0.001	0	0.016	5	0.0002
50	n-Octane	0.050	0.036	0.193	0.231	0.170	0.902	91	
51	Propionaldehyde	0.171	0.167	0.322	0.405	0.397	0.765	100	0.1
52	Propylene	1.871	1.105	17.800	3.220	1.902	30.635	100	0.001
53	Styrene	0.071	0.039	0.569	0.301	0.167	2.424	100	0.2
54	1,1,2,2-Tetrachloroethane	0.0004	0	0.005	0.003	0	0.036	27	0.2
55	Tetrachloroethylene	0.022	0.014	0.284	0.150	0.092	1.926	100	0.9
56	Toluene	0.476	0.325	1.660	1.795	1.223	6.256	100	0.0005
57	1,2,4-Trichlorobenzene	0.002	0.001	0.014	0.012	0.006	0.106	75	0.01
58	1,1,1-Trichloroethane	0.003	0.002	0.013	0.015	0.011	0.068	96	0.00001
59	1,1,2-Trichloroethane	0.0004	0	0.012	0.002	0	0.066	7	0.04
60	Trichloroethylene	0.005	0.006	0.025	0.027	0.030	0.136	62	0.1
61	Trichlorofluoromethane	0.322	0.277	0.877	1.809	1.553	4.927	100	0.003
62	Trichlorotrifluoroethane	0.076	0.076	0.111	0.584	0.579	0.851	100	0.00002
63	1,2,4-Trimethylbenzene	0.064	0.036	0.288	0.314	0.177	1.416	75	0.01
64	1,3,5-Trimethylbenzene	0.018	0.012	0.089	0.090	0.057	0.436	100	0.001
65	Valeraldehyde	0.051	0.049	0.189	0.179	0.171	0.666	100	
66	Vinyl Chloride	0.004	0.002	0.033	0.009	0.004	0.085	57	0.1
67	m,p-Xylene	0.167	0.106	0.754	0.723	0.460	3.274	100	0.01
68	o-Xylene	0.070	0.044	0.323	0.303	0.193	1.403	100	0.003

- Arithmetic means in *italics* had fewer than 50% of samples with detectable concentrations.
- For a valid 24-hour sampling event, when the analyzing laboratory reports the term “Not Detected” for a particular pollutant, the concentration of 0.0 ppbv is assigned to that pollutant. These zero concentrations were included in the calculation of annual averages and medians for each pollutant regardless of percent detection.
- Annual mean **risk ratios** in *italics* are based on noncancer effects.
- A risk ratio for a pollutant is calculated by dividing the annual mean air concentration by the long-term health benchmark. If the annual mean is 0, then the risk ratio is not calculated. See Table 10-9 for chemical-specific health benchmarks.

**Table 10-5**  
**CHESTER – 2022 NJ Air Toxics Monitoring Data**

Pollutant		Annual Mean	Annual Median	24-Hour Max.	Annual Mean	Annual Median	24-Hour Max.	% Detected	Annual Mean Risk Ratio
		ppbv			µg/m³				
1	Acetaldehyde	0.535	0.492	1.620	0.964	0.886	2.919	100	2.1
2	Acetone	0.909	0.926	1.680	2.159	2.198	3.991	100	0.0001
3	Acetonitrile	1.240	0.182	53.600	2.083	0.306	89.997	96	0.03
4	Acetylene	0.326	0.264	1.510	0.347	0.281	1.608	100	
5	Acrolein	0.365	0.343	1.200	0.836	0.786	2.752	100	41.8
6	Acrylonitrile	0.001	0	0.029	0.002	0	0.064	6	0.2
7	tert-Amyl Methyl Ether	0.001	0	0.015	0.002	0	0.063	13	
8	Benzaldehyde	0.022	0.015	0.092	0.096	0.066	0.400	100	
9	Benzene	0.096	0.074	0.232	0.305	0.236	0.741	100	2.3
10	Bromochloromethane	0.001	0	0.016	0.004	0	0.084	27	0.0001
11	Bromodichloromethane	0.001	0	0.018	0.006	0	0.123	35	0.2
12	Bromoform	0.002	0.002	0.021	0.023	0.016	0.221	93	0.03
13	Bromomethane	0.010	0.009	0.028	0.038	0.034	0.108	100	0.01
14	1,3-Butadiene	0.006	0.005	0.027	0.014	0.011	0.059	95	0.4
15	Butyraldehyde	0.027	0.027	0.062	0.079	0.080	0.183	100	
16	Carbon Disulfide	0.020	0.014	0.109	0.061	0.043	0.339	100	0.0001
17	Carbon Tetrachloride	0.075	0.078	0.100	0.469	0.488	0.629	100	2.8
18	Chlorobenzene	0.001	0	0.016	0.004	0	0.075	13	0.000004
19	Chloroethane	0.012	0.009	0.044	0.031	0.023	0.117	96	0.000003
20	Chloroform	0.020	0.019	0.034	0.096	0.092	0.166	100	2.2
21	Chloromethane	0.479	0.486	0.560	0.990	1.004	1.156	100	1.8
22	Chloroprene	0.001	0	0.019	0.002	0	0.067	9	0.9
23	Crotonaldehyde	0.006	0	0.044	0.017	0	0.125	40	
24	Dibromochloromethane	0.001	0.0003	0.018	0.008	0.003	0.152	56	0.2
25	1,2-Dibromoethane	0.001	0	0.015	0.005	0	0.113	31	3.2
26	m-Dichlorobenzene	0.001	0	0.015	0.005	0	0.088	42	
27	o-Dichlorobenzene	0.001	0	0.014	0.005	0	0.085	49	0.00003
28	p-Dichlorobenzene	0.002	0.002	0.015	0.014	0.010	0.091	89	0.2
29	Dichlorodifluoromethane	0.501	0.502	0.583	2.477	2.483	2.883	100	0.02
30	1,1-Dichloroethane	0.001	0	0.017	0.003	0	0.068	13	0.004
31	1,2-Dichloroethane	0.014	0.014	0.029	0.057	0.056	0.118	100	1.5
32	1,1-Dichloroethylene	0.002	0.002	0.017	0.008	0.006	0.068	91	0.00004
33	cis-1,2-Dichloroethylene	0.0004	0	0.017	0.002	0	0.068	4	
34	trans-1,2-Dichloroethylene	0.007	0.001	0.284	0.027	0.004	1.126	53	
35	Dichloromethane	0.137	0.109	0.517	0.475	0.379	1.796	100	0.01
36	1,2-Dichloropropane	0.001	0	0.019	0.003	0	0.090	5	0.03
37	cis-1,3-Dichloropropylene	0.001	0	0.016	0.003	0	0.071	7	0.01
38	trans-1,3-Dichloropropylene	0.001	0	0.016	0.002	0	0.074	4	0.01
39	Dichlorotetrafluoroethane	0.018	0.018	0.029	0.126	0.124	0.201	100	
40	Ethyl Acrylate	0	0	0	0	0	0	0	0
41	Ethylbenzene	0.015	0.012	0.052	0.065	0.052	0.225	100	0.2
42	Ethyl tert-Butyl Ether	0.004	0.003	0.019	0.007	0.006	0.034	67	
43	Formaldehyde	1.526	1.265	7.410	1.874	1.553	9.100	100	24.3
44	Hexachlorobutadiene	0.001	0.0002	0.014	0.010	0.002	0.144	73	0.2
45	Hexaldehyde	0.058	0.047	0.202	0.236	0.193	0.827	100	

Continued

**Table 10-5 (continued)**  
**CHESTER – 2022 NJ Air Toxics Monitoring Data**

Pollutant		Annual Mean	Annual Median	24-Hour Max.	Annual Mean	Annual Median	24-Hour Max.	% Detected	Annual Mean Risk Ratio
		ppbv			µg/m³				
46	Methyl Ethyl Ketone	0.088	0.088	0.154	0.258	0.258	0.454	100	0.0001
47	Methyl Isobutyl Ketone	0.021	0.018	0.187	0.086	0.075	0.766	96	0.00003
48	Methyl Methacrylate	0.0002	0	0.013	0.001	0	0.054	2	0.000001
49	Methyl tert-Butyl Ether	0.0004	0	0.007	0.002	0	0.026	11	0.0004
50	n-Octane	0.019	0.016	0.070	0.088	0.076	0.328	84	
51	Propionaldehyde	0.103	0.095	0.233	0.244	0.226	0.553	100	0.03
52	Propylene	0.414	0.356	1.090	0.713	0.613	1.876	100	0.0002
53	Styrene	0.012	0.005	0.096	0.051	0.020	0.411	94	0.03
54	1,1,2,2-Tetrachloroethane	0.001	0	0.008	0.004	0	0.052	25	0.2
55	Tetrachloroethylene	0.008	0.006	0.029	0.052	0.041	0.196	100	0.3
56	Toluene	0.100	0.084	0.463	0.376	0.318	1.745	100	0.0001
57	1,2,4-Trichlorobenzene	0.002	0.000	0.016	0.012	0.002	0.120	64	0.01
58	1,1,1-Trichloroethane	0.002	0.002	0.018	0.013	0.010	0.098	96	0.00001
59	1,1,2-Trichloroethane	0.001	0	0.016	0.004	0	0.085	13	0.1
60	Trichloroethylene	0.005	0.005	0.023	0.028	0.026	0.121	92	0.1
61	Trichlorofluoromethane	0.239	0.239	0.289	1.344	1.343	1.624	100	0.002
62	Trichlorotrifluoroethane	0.076	0.076	0.097	0.581	0.584	0.743	100	0.00002
63	1,2,4-Trimethylbenzene	0.007	0.002	0.031	0.034	0.011	0.152	50	0.001
64	1,3,5-Trimethylbenzene	0.003	0.002	0.020	0.016	0.012	0.097	100	0.0003
65	Valeraldehyde	0.025	0.021	0.083	0.087	0.072	0.291	100	
66	Vinyl Chloride	0.001	0	0.017	0.003	0	0.044	40	0.03
67	m,p-Xylene	0.030	0.025	0.095	0.131	0.108	0.413	100	0.001
68	o-Xylene	0.013	0.010	0.040	0.058	0.045	0.175	100	0.001

- Arithmetic means in *italics* had fewer than 50% of samples with detectable concentrations.
- For a valid 24-hour sampling event, when the analyzing laboratory reports the term “Not Detected” for a particular pollutant, the concentration of 0.0 ppbv is assigned to that pollutant. These zero concentrations were included in the calculation of annual averages and medians for each pollutant regardless of percent detection.
- Annual mean **risk ratios** in *italics* are based on noncancer effects.
- A risk ratio for a pollutant is calculated by dividing the annual mean air concentration by the long-term health benchmark. If the annual mean is 0, then the risk ratio is not calculated. See Table 10-9 for chemical-specific health benchmarks.

**Table 10-6**  
**ELIZABETH – 2022 NJ Air Toxics Monitoring Data**

Pollutant		Annual Mean	Annual Median	24-Hour Max.	Annual Mean	Annual Median	24-Hour Max.	% Detected	Annual Mean Risk Ratio
		ppbv			µg/m³				
1	Acetaldehyde	0.768	0.788	1.670	1.383	1.420	3.009	100	3.1
2	Acetone	0.962	0.881	2.080	2.285	2.093	4.941	100	0.0001
3	Acetonitrile	0.473	0.329	2.110	0.794	0.552	3.543	91	0.01
4	Acetylene	1.620	0.621	23.000	1.726	0.661	24.494	100	
5	Acrolein	0.378	0.282	1.070	0.868	0.647	2.453	100	43.4
6	Acrylonitrile	0.002	0	0.014	0.004	0	0.030	22	0.3
7	tert-Amyl Methyl Ether	0.0004	0	0.011	0.002	0	0.046	11	
8	Benzaldehyde	0.019	0.019	0.056	0.084	0.081	0.244	98	
9	Benzene	0.220	0.199	0.604	0.704	0.636	1.930	100	5.4
10	Bromochloromethane	0.0005	0	0.014	0.002	0	0.072	21	0.0001
11	Bromodichloromethane	0.001	0	0.013	0.007	0	0.084	47	0.3
12	Bromoform	0.002	0.002	0.017	0.025	0.021	0.174	96	0.03
13	Bromomethane	0.010	0.009	0.023	0.038	0.035	0.089	100	0.01
14	1,3-Butadiene	0.026	0.021	0.092	0.057	0.047	0.204	98	1.7
15	Butyraldehyde	0.034	0.031	0.092	0.100	0.090	0.272	97	
16	Carbon Disulfide	0.035	0.028	0.146	0.110	0.087	0.455	100	0.0002
17	Carbon Tetrachloride	0.070	0.076	0.108	0.438	0.481	0.679	98	2.6
18	Chlorobenzene	0.001	0	0.014	0.003	0	0.066	14	0.000003
19	Chloroethane	0.012	0.009	0.070	0.031	0.024	0.183	91	0.000003
20	Chloroform	0.027	0.025	0.049	0.132	0.124	0.237	98	3.1
21	Chloromethane	0.492	0.490	0.621	1.016	1.012	1.282	100	1.8
22	Chloroprene	0.0003	0	0.012	0.001	0	0.044	4	0.6
23	Crotonaldehyde	0.004	0.003	0.020	0.013	0.009	0.058	59	
24	Dibromochloromethane	0.001	0.001	0.013	0.008	0.004	0.112	68	0.2
25	1,2-Dibromoethane	0.001	0	0.011	0.004	0	0.088	20	2.5
26	m-Dichlorobenzene	0.001	0.0002	0.011	0.004	0.001	0.064	51	
27	o-Dichlorobenzene	0.001	0.0004	0.011	0.006	0.002	0.065	63	0.00003
28	p-Dichlorobenzene	0.006	0.005	0.027	0.035	0.028	0.161	100	0.4
29	Dichlorodifluoromethane	0.505	0.500	0.602	2.496	2.473	2.977	100	0.02
30	1,1-Dichloroethane	0.0004	0	0.014	0.002	0	0.055	11	0.003
31	1,2-Dichloroethane	0.014	0.014	0.028	0.058	0.057	0.114	95	1.5
32	1,1-Dichloroethylene	0.001	0.001	0.014	0.005	0.005	0.054	58	0.00003
33	cis-1,2-Dichloroethylene	0.0003	0	0.014	0.001	0	0.056	4	
34	trans-1,2-Dichloroethylene	0.003	0.002	0.018	0.012	0.009	0.070	70	
35	Dichloromethane	0.197	0.140	1.150	0.684	0.486	3.995	100	0.01
36	1,2-Dichloropropane	0.001	0	0.017	0.002	0	0.079	5	0.02
37	cis-1,3-Dichloropropylene	0.0003	0	0.012	0.001	0	0.054	4	0.01
38	trans-1,3-Dichloropropylene	0.0002	0	0.011	0.001	0	0.050	2	0.004
39	Dichlorotetrafluoroethane	0.015	0.016	0.029	0.106	0.113	0.200	91	
40	Ethyl Acrylate	0.0002	0	0.009	0.001	0	0.036	2	0.0001
41	Ethylbenzene	0.060	0.052	0.286	0.259	0.228	1.242	100	0.6
42	Ethyl tert-Butyl Ether	0.0004	0	0.012	0.001	0	0.021	5	
43	Formaldehyde	1.910	1.690	5.830	2.346	2.075	7.160	100	30.5
44	Hexachlorobutadiene	0.001	0.000	0.011	0.008	0.003	0.117	74	0.2
45	Hexaldehyde	0.067	0.063	0.304	0.276	0.259	1.245	100	

Continued



**Table 10-6 (continued)**  
**ELIZABETH – 2022 NJ Air Toxics Monitoring Data**

Pollutant		Annual Mean	Annual Median	24-Hour Max.	Annual Mean	Annual Median	24-Hour Max.	% Detected	Annual Mean Risk Ratio
		ppbv			µg/m³				
46	Methyl Ethyl Ketone	0.099	0.089	0.248	0.291	0.261	0.731	100	0.0001
47	Methyl Isobutyl Ketone	0.032	0.027	0.122	0.130	0.112	0.500	100	0.00004
48	Methyl Methacrylate	0.001	0	0.021	0.005	0	0.086	7	0.00001
49	Methyl tert-Butyl Ether	0.0003	0	0.008	0.001	0	0.027	5	0.0003
50	n-Octane	0.065	0.053	0.432	0.304	0.247	2.018	91	
51	Propionaldehyde	0.109	0.110	0.285	0.258	0.261	0.677	100	0.03
52	Propylene	2.393	1.140	15.300	4.119	1.962	26.333	100	0.001
53	Styrene	0.026	0.018	0.100	0.113	0.078	0.426	94	0.1
54	1,1,2,2-Tetrachloroethane	0.0005	0	0.007	0.003	0	0.047	19	0.2
55	Tetrachloroethylene	0.015	0.010	0.071	0.101	0.070	0.480	100	0.6
56	Toluene	0.390	0.296	1.240	1.470	1.115	4.673	100	0.0004
57	1,2,4-Trichlorobenzene	0.001	0.001	0.015	0.011	0.004	0.113	70	0.01
58	1,1,1-Trichloroethane	0.003	0.002	0.013	0.015	0.012	0.071	98	0.00001
59	1,1,2-Trichloroethane	0.002	0	0.039	0.011	0	0.210	11	0.2
60	Trichloroethylene	0.006	0.006	0.020	0.033	0.031	0.107	93	0.2
61	Trichlorofluoromethane	0.243	0.242	0.296	1.364	1.360	1.663	100	0.002
62	Trichlorotrifluoroethane	0.076	0.076	0.093	0.585	0.581	0.710	100	0.00002
63	1,2,4-Trimethylbenzene	0.051	0.041	0.144	0.250	0.204	0.708	90	0.004
64	1,3,5-Trimethylbenzene	0.013	0.010	0.042	0.066	0.049	0.205	100	0.001
65	Valeraldehyde	0.027	0.022	0.135	0.094	0.078	0.476	100	
66	Vinyl Chloride	0.001	0	0.012	0.003	0	0.031	33	0.02
67	m,p-Xylene	0.163	0.131	0.620	0.708	0.569	2.692	100	0.01
68	o-Xylene	0.068	0.056	0.321	0.294	0.244	1.394	100	0.003

- Arithmetic means in *italics* had fewer than 50% of samples with detectable concentrations.
- For a valid 24-hour sampling event, when the analyzing laboratory reports the term “Not Detected” for a particular pollutant, the concentration of 0.0 ppbv is assigned to that pollutant. These zero concentrations were included in the calculation of annual averages and medians for each pollutant regardless of percent detection.
- Annual mean **risk ratios** in *italics* are based on noncancer effects.
- A risk ratio for a pollutant is calculated by dividing the annual mean air concentration by the long-term health benchmark. If the annual mean is 0, then the risk ratio is not calculated. See Table 10-9 for chemical-specific health benchmarks.

**Table 10-7**  
**RUTGERS – 2022 NJ Air Toxics Monitoring Data**

Pollutant		Annual Mean	Annual Median	24-Hour Max.	Annual Mean	Annual Median	24-Hour Max.	% Detected	Annual Mean Risk Ratio
		ppbv			µg/m³				
1	Acetaldehyde	0.709	0.631	1.550	1.277	1.136	2.793	100	2.8
2	Acetone	0.918	0.831	1.910	2.182	1.974	4.537	100	0.0001
3	Acetonitrile	0.361	0.244	4.880	0.605	0.410	8.194	88	0.01
4	Acetylene	0.518	0.346	2.880	0.551	0.368	3.067	100	
5	Acrolein	0.632	0.314	15.600	1.450	0.719	35.771	100	72.5
6	Acrylonitrile	0.002	0	0.034	0.005	0	0.073	21	0.3
7	tert-Amyl Methyl Ether	0.001	0	0.012	0.002	0	0.051	19	
8	Benzaldehyde	0.015	0.013	0.077	0.063	0.058	0.333	100	
9	Benzene	0.122	0.097	0.340	0.389	0.310	1.086	100	3.0
10	Bromochloromethane	0.001	0	0.012	0.004	0	0.064	27	0.0001
11	Bromodichloromethane	0.001	0	0.015	0.009	0	0.101	41	0.3
12	Bromoform	0.002	0.002	0.017	0.023	0.018	0.173	95	0.03
13	Bromomethane	0.009	0.009	0.020	0.036	0.034	0.078	100	0.01
14	1,3-Butadiene	0.012	0.009	0.058	0.025	0.021	0.129	98	0.8
15	Butyraldehyde	0.039	0.031	0.128	0.116	0.093	0.377	100	
16	Carbon Disulfide	0.020	0.015	0.096	0.061	0.046	0.297	100	0.0001
17	Carbon Tetrachloride	0.069	0.076	0.098	0.433	0.478	0.619	100	2.5
18	Chlorobenzene	0.001	0	0.013	0.003	0	0.058	12	0.000003
19	Chloroethane	0.017	0.009	0.222	0.045	0.024	0.586	90	0.000005
20	Chloroform	0.025	0.023	0.063	0.122	0.113	0.305	100	2.8
21	Chloromethane	0.482	0.486	0.594	0.994	1.004	1.227	100	1.8
22	Chloroprene	0.001	0	0.015	0.002	0	0.055	12	1.0
23	Crotonaldehyde	0.005	0	0.044	0.014	0	0.125	45	
24	Dibromochloromethane	0.001	0.0004	0.015	0.009	0.003	0.126	73	0.2
25	1,2-Dibromoethane	0.001	0	0.012	0.005	0	0.090	32	3.1
26	m-Dichlorobenzene	0.001	0.0003	0.012	0.006	0.002	0.071	56	
27	o-Dichlorobenzene	0.001	0.0003	0.012	0.006	0.002	0.073	59	0.00003
28	p-Dichlorobenzene	0.004	0.003	0.018	0.023	0.016	0.106	100	0.2
29	Dichlorodifluoromethane	0.504	0.504	0.586	2.491	2.492	2.898	100	0.02
30	1,1-Dichloroethane	0.001	0	0.014	0.003	0	0.057	20	0.005
31	1,2-Dichloroethane	0.015	0.014	0.029	0.059	0.056	0.118	98	1.6
32	1,1-Dichloroethylene	0.002	0.001	0.014	0.006	0.005	0.056	78	0.00003
33	cis-1,2-Dichloroethylene	0.0005	0	0.015	0.002	0	0.057	5	
34	trans-1,2-Dichloroethylene	0.002	0.002	0.024	0.009	0.006	0.097	63	
35	Dichloromethane	0.164	0.124	0.712	0.570	0.431	2.473	100	0.01
36	1,2-Dichloropropane	0.001	0	0.018	0.002	0	0.081	5	0.02
37	cis-1,3-Dichloropropylene	0.0005	0	0.012	0.002	0	0.055	7	0.01
38	trans-1,3-Dichloropropylene	0.0002	0	0.012	0.001	0	0.054	2	0.004
39	Dichlorotetrafluoroethane	0.018	0.018	0.030	0.126	0.123	0.208	100	
40	Ethyl Acrylate	0.0002	0	0.010	0.001	0	0.039	2	0.0001
41	Ethylbenzene	0.025	0.022	0.082	0.108	0.097	0.357	100	0.3
42	Ethyl tert-Butyl Ether	0.005	0.006	0.021	0.010	0.010	0.037	68	
43	Formaldehyde	1.637	1.525	5.540	2.010	1.873	6.803	100	26.1
44	Hexachlorobutadiene	0.001	0.000	0.013	0.010	0.003	0.140	75	0.2
45	Hexaldehyde	0.045	0.035	0.266	0.183	0.145	1.090	100	

Continued

**Table 10-7 (continued)**  
**RUTGERS – 2022 NJ Air Toxics Monitoring Data**

Pollutant		Annual Mean	Annual Median	24-Hour Max.	Annual Mean	Annual Median	24-Hour Max.	% Detected	Annual Mean Risk Ratio
		ppbv			µg/m³				
46	Methyl Ethyl Ketone	0.101	0.098	0.329	0.296	0.289	0.970	100	0.0001
47	Methyl Isobutyl Ketone	0.023	0.020	0.090	0.094	0.080	0.369	98	0.00003
48	Methyl Methacrylate	0.0001	0	0.009	0.001	0	0.036	2	0.000001
49	Methyl tert-Butyl Ether	0.001	0	0.008	0.002	0	0.028	12	0.001
50	n-Octane	0.023	0.021	0.098	0.108	0.097	0.459	80	
51	Propionaldehyde	0.115	0.095	0.305	0.273	0.226	0.725	100	0.03
52	Propylene	0.590	0.554	1.940	1.016	0.953	3.339	100	0.0003
53	Styrene	0.012	0.010	0.044	0.051	0.043	0.186	97	0.03
54	1,1,2,2-Tetrachloroethane	0.001	0	0.008	0.004	0	0.054	34	0.3
55	Tetrachloroethylene	0.009	0.007	0.033	0.061	0.050	0.226	100	0.4
56	Toluene	0.167	0.144	0.556	0.629	0.543	2.095	100	0.0002
57	1,2,4-Trichlorobenzene	0.002	0.001	0.023	0.015	0.004	0.169	69	0.01
58	1,1,1-Trichloroethane	0.003	0.002	0.015	0.014	0.011	0.080	95	0.00001
59	1,1,2-Trichloroethane	0.001	0	0.013	0.004	0	0.073	10	0.1
60	Trichloroethylene	0.006	0.005	0.021	0.030	0.029	0.110	92	0.2
61	Trichlorofluoromethane	0.242	0.239	0.295	1.357	1.343	1.657	100	0.002
62	Trichlorotrifluoroethane	0.076	0.076	0.100	0.586	0.579	0.764	100	0.00002
63	1,2,4-Trimethylbenzene	0.014	0.009	0.084	0.068	0.046	0.413	59	0.001
64	1,3,5-Trimethylbenzene	0.005	0.004	0.025	0.026	0.022	0.122	100	0.0004
65	Valeraldehyde	0.028	0.024	0.068	0.099	0.086	0.238	98	
66	Vinyl Chloride	0.002	0	0.015	0.004	0	0.038	49	0.04
67	m,p-Xylene	0.059	0.051	0.230	0.255	0.221	0.999	100	0.003
68	o-Xylene	0.024	0.022	0.087	0.106	0.093	0.379	100	0.001

- Arithmetic means in *italics* had fewer than 50% of samples with detectable concentrations.
- For a valid 24-hour sampling event, when the analyzing laboratory reports the term “Not Detected” for a particular pollutant, the concentration of 0.0 ppbv is assigned to that pollutant. These zero concentrations were included in the calculation of annual averages and medians for each pollutant regardless of percent detection.
- Annual mean **risk ratios** in *italics* are based on noncancer effects.
- A risk ratio for a pollutant is calculated by dividing the annual mean air concentration by the long-term health benchmark. If the annual mean is 0, then the risk ratio is not calculated. See Table 10-9 for chemical-specific health benchmarks.

**Table 10-8**  
**2022 Air Toxics Detection Limits and Health Benchmarks**

	Pollutant	CAS No.	Detection Limit (ppbv)	Detection Limit (µg/m³)	Health Benchmark (µg/m³)
1	Acetaldehyde	75-07-0	0.0796	0.143	0.45
2	Acetone	67-64-1	0.2990	0.710	31000
3	Acetonitrile	75-05-8	0.1378	0.231	60
4	Acetylene	74-86-2	0.1100	0.117	
5	Acrolein	107-02-8	0.1775	0.407	0.02
6	Acrylonitrile	107-13-1	0.0171	0.037	0.015
7	tert-Amyl Methyl Ether	994-05-8	0.0144	0.060	
8	Benzaldehyde	100-52-7	0.0624	0.271	
9	Benzene	71-43-2	0.0117	0.037	0.13
10	Bromochloromethane	74-97-5	0.0114	0.060	40
11	Bromodichloromethane	75-27-4	0.0254	0.170	0.027
12	Bromoform	75-25-2	0.0136	0.141	0.91
13	Bromomethane	74-83-9	0.0205	0.080	5
14	1,3-Butadiene	106-99-0	0.0207	0.046	0.033
15	Butyraldehyde	123-72-8	0.0606	0.179	
16	Carbon Disulfide	75-15-0	0.0190	0.059	700
17	Carbon Tetrachloride	56-23-5	0.0112	0.070	0.17
18	Chlorobenzene	108-90-7	0.0132	0.061	1000
19	Chloroethane	75-00-3	0.0164	0.043	10000
20	Chloroform	67-66-3	0.0076	0.037	0.043
21	Chloromethane	74-87-3	0.0507	0.105	0.56
22	Chloroprene	126-99-8	0.0213	0.077	0.002
23	Crotonaldehyde	123-73-9	0.0187	0.054	
24	Dibromochloromethane	124-48-1	0.0143	0.122	0.037
25	1,2-Dibromoethane	106-93-4	0.0154	0.118	0.0017
26	m-Dichlorobenzene	541-73-1	0.0162	0.098	
27	o-Dichlorobenzene	95-50-1	0.0187	0.112	200
28	p-Dichlorobenzene	106-46-7	0.0169	0.102	0.091
29	Dichlorodifluoromethane	75-71-8	0.0244	0.121	100
30	1,1-Dichloroethane	75-34-3	0.0078	0.032	0.63
31	1,2-Dichloroethane	107-06-2	0.0072	0.029	0.038
32	1,1-Dichloroethylene	75-35-4	0.0115	0.046	200
33	cis-1,2-Dichloroethylene	156-59-2	0.0228	0.090	
34	trans-1,2-Dichloroethylene	156-60-5	0.0087	0.034	
35	Dichloromethane	75-09-2	0.1975	0.686	77
36	1,2-Dichloropropane	78-87-5	0.0091	0.042	0.1
37	cis-1,3-Dichloropropylene	10061-01-5	0.0077	0.035	0.25
38	trans-1,3-Dichloropropylene	10061-02-6	0.0159	0.072	0.25
39	Dichlorotetrafluoroethane	76-14-2	0.0083	0.058	
40	Ethyl Acrylate	140-88-5	0.0132	0.054	8
41	Ethylbenzene	100-41-4	0.0112	0.049	0.4
42	Ethyl tert-Butyl Ether	637-92-3	0.0100	0.018	
43	Formaldehyde	50-00-0	0.1110	0.136	0.077
44	Hexachlorobutadiene	87-68-3	0.0736	0.785	0.045
45	Hexaldehyde	66-25-1	0.1320	0.541	

Continued

**Table 10-8 (continued)**  
**Air Toxics Detection Limits and Health Benchmarks**

	Pollutant	CAS No.	Detection Limit (ppbv)	Detection Limit (µg/m³)	Health Benchmark (µg/m³)
46	Methyl Ethyl Ketone	78-93-3	0.0262	0.077	5000
47	Methyl Isobutyl Ketone	108-10-1	0.0308	0.126	3000
48	Methyl Methacrylate	80-62-6	0.0667	0.273	700
49	Methyl tert-Butyl Ether	1634-04-4	0.0092	0.033	3.8
50	n-Octane	111-65-9	0.0144	0.067	
51	Propionaldehyde	123-38-6	0.1020	0.242	8
52	Propylene	115-07-1	0.1300	0.224	3000
53	Styrene	100-42-5	0.0164	0.070	1.8
54	1,1,2,2-Tetrachloroethane	79-34-5	0.0158	0.108	0.017
55	Tetrachloroethylene	127-18-4	0.0287	0.194	0.16
56	Toluene	108-88-3	0.0592	0.223	420
57	1,2,4-Trichlorobenzene	120-82-1	0.0686	0.509	2
58	1,1,1-Trichloroethane	71-55-6	0.0119	0.065	1000
59	1,1,2-Trichloroethane	79-00-5	0.0108	0.059	0.063
60	Trichloroethylene	79-01-6	0.0146	0.078	0.2
61	Trichlorofluoromethane	75-69-4	0.0235	0.132	700
62	Trichlorotrifluoroethane	76-13-1	0.0119	0.091	30000
63	1,2,4-Trimethylbenzene	95-63-6	0.0129	0.063	60
64	1,3,5-Trimethylbenzene	108-67-8	0.0150	0.074	60
65	Valeraldehyde	110-62-3	0.0582	0.205	
66	Vinyl chloride	75-01-4	0.0086	0.022	0.11
67	m,p-Xylene	108-38-3 106-42-3	0.0320	0.139	100
68	o-Xylene	95-47-6	0.0154	0.067	100

- **Detection limits** are from ERG analytic lab, Morrisville, NC.
- **Health benchmark** - the chemical-specific air concentration above which there may be human health concerns. Not available for all chemicals. Those presented here are for long-term exposure.
- For a carcinogen (cancer-causing chemical), the health benchmark is set at the air concentration that would cause no more than a one-in-a-million increase in the likelihood of getting cancer, even after a lifetime of exposure.
- For a noncarcinogen, the health benchmark is the maximum air concentration to which exposure is likely to cause no harm, even if that exposure occurs on a daily basis for a lifetime.
- Health benchmarks in *italics* are based on noncancer effects.
- Health benchmarks are from *Toxicity Values for Inhalation Exposure*, NJDEP Bureau of Evaluation & Planning, April 2023.

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# 2022 PAMS Summary

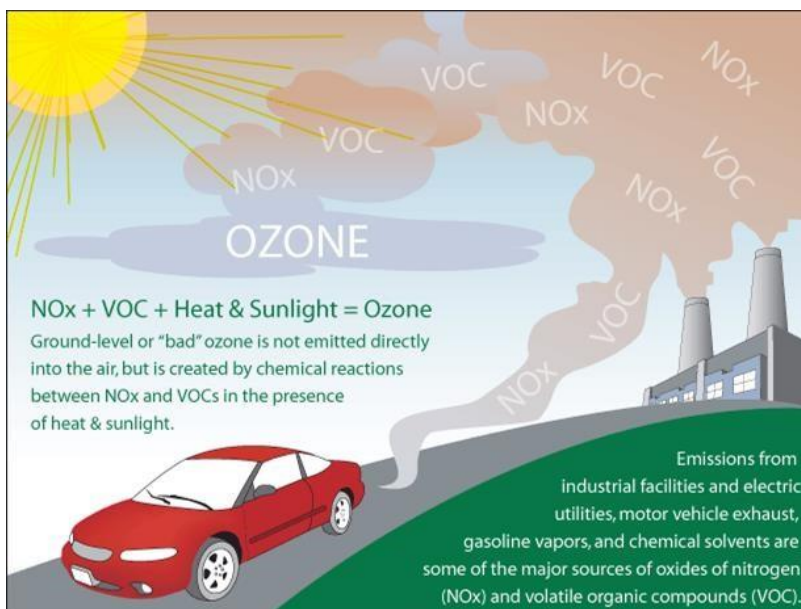
New Jersey Department of Environmental Protection

## PHOTOCHEMICAL ASSESSMENT MONITORING

Most ground-level ozone ( $O_3$ ) is formed when volatile organic compounds (VOCs) and oxides of nitrogen ( $NO_x$ ) react in the presence of sunlight, as shown in Figure 11-1. Therefore, to effectively evaluate strategies for reducing ozone levels, it is necessary to measure these ozone-forming pollutants, also known as precursor pollutants. The Photochemical Assessment Monitoring Stations (PAMS) network was established by the U.S. Environmental Protection Agency (USEPA) for this purpose. Data from the PAMS network is used to better characterize the nature and extent of the ozone problem, track VOC and  $NO_x$  emissions, assess air quality trends, and make planning decisions.

PAMS monitor both criteria and non-criteria pollutants. These include ozone, nitric oxide (NO), nitrogen dioxide ( $NO_2$ ), total reactive oxides of nitrogen ( $NO_y$ ), and specific VOCs, such as several that are carbonyls and are important in ozone formation. In addition, the measurement of specific weather parameters is required at all PAMS: wind speed and direction; temperature; barometric pressure; relative humidity; precipitation; solar radiation; UV radiation; and mixing layer height. The VOCs and carbonyls are measured only during peak ozone season, from June 1st to August 31st each year. The PAMS VOC and carbonyl data is the focus of this section of the annual Air Quality Report.

**Figure 11-1**  
**Ozone Formation**



<https://www.epa.gov/ground-level-ozone-pollution/ground-level-ozone-basics#wwh>

In 2015, USEPA revised the National Ambient Air Quality Standard (NAAQS) for ozone, including the requirements for monitoring. This led to an overhaul of the PAMS program, with changes to the methodology for measuring the PAMS target pollutants and also to the locations of PAMS sites in the U.S. To support the implementation, USEPA designated specific equipment for the continuous hourly measurement of ozone precursor VOCs, provided funding to the states for purchasing the equipment, and set a date of June 1, 2021, to begin monitoring using the approved instruments. NJDEP purchased the new instruments in 2018, evaluated and tested them through 2020, and met the deadline of June 1, 2021, for measuring the target VOCs.

New Jersey once had a number of PAMS sites around the state. Currently, its sole PAMS station is on Rutgers University agricultural land in East Brunswick.

**Figure 11-2**  
**2022 Photochemical Assessment Monitoring Station**





New Jersey collects samples using the Markes-Agilent system, which electrically cools and freezes humidified air, allowing the detection of more polar compounds, and the reporting of concentrations of both alpha- and beta-pinenes. The concentrations of the 2022 PAMS target compounds are presented in Table 11-1 below.

**Table 11-1**  
**2022 PAMS Target Compounds in New Jersey**  
**Annual Average and Hourly Maximum Concentrations**  
 Parts per Billion Carbon (ppbC)  
 Parts per Billion by Volume (ppbv)  
 Micrograms per Cubic Meter ( $\mu\text{g}/\text{m}^3$ )

Compound	Annual Average			Hourly Maximum		
	ppbC	ppbv	$\mu\text{g}/\text{m}^3$	ppbC	ppbv	$\mu\text{g}/\text{m}^3$
Acetylene	0.18	0.09	0.10	0.80	0.40	0.43
Benzene	0.35	0.06	0.19	1.38	0.23	0.73
1,3-Butadiene	0.42	0.10	0.23	70.40	17.60	38.94
n-Butane	1.51	0.38	0.90	19.64	4.91	11.67
1-Butene	0.21	0.05	0.12	0.52	0.13	0.30
c-2 Butene	0.02	0.00	0.01	0.53	0.13	0.30
t-2 Butene	0.00	0.00	0.00	0.27	0.07	0.15
Cyclohexane	0.07	0.01	0.04	1.30	0.22	0.75
Cyclopentane	0.08	0.02	0.05	0.55	0.11	0.32
n-Decane	0.08	0.01	0.05	0.95	0.10	0.55
m-Diethylbenzene	0.06	0.01	0.03	0.53	0.05	0.29
p-Diethylbenzene	0.02	0.00	0.01	0.32	0.03	0.18
2 2-Dimethylbutane	0.10	0.02	0.06	0.88	0.15	0.52
2 3-Dimethylbutane	0.18	0.03	0.10	0.91	0.15	0.53
2 3-Dimethylpentane	0.08	0.01	0.04	1.13	0.16	0.66
2 4-Dimethylpentane	0.04	0.01	0.03	0.42	0.06	0.25
n-Dodecane	0.02	0.00	0.01	1.12	0.09	0.65
Ethane	3.78	1.89	2.32	120.64	60.32	74.18
Ethylbenzene	0.17	0.02	0.09	1.18	0.15	0.64
Ethylene	0.68	0.34	0.39	3.36	1.68	1.93
m-Ethyltoluene	0.09	0.01	0.05	0.84	0.09	0.46
o-Ethyltoluene	0.02	0.00	0.01	1.58	0.18	0.86
p-Ethyltoluene	0.17	0.02	0.10	3.62	0.40	1.98
Hexane	0.42	0.07	0.25	3.62	0.60	2.13
1-Hexene	0.05	0.01	0.03	0.25	0.04	0.14
n-Heptane	0.20	0.03	0.11	4.44	0.63	2.60
Isobutane	0.70	0.18	0.42	37.01	9.25	21.99
Isopentane	1.55	0.31	0.91	14.28	2.86	8.43
Isoprene	3.79	0.76	2.11	23.28	4.66	12.97
Isopropylbenzene	0.03	0.00	0.01	0.49	0.05	0.27
Methylcyclohexane	0.12	0.02	0.07	0.76	0.11	0.44
Methylcyclopentane	0.13	0.02	0.07	0.88	0.15	0.50
2-Methylheptane	0.03	0.00	0.02	0.38	0.05	0.22
3-Methylheptane	0.03	0.00	0.02	1.15	0.14	0.67

Continued

**Table 11-1 (continued)**  
**2022 PAMS Target Compounds in New Jersey**  
**Annual Average and Hourly Maximum Concentrations**

Parts per Billion Carbon (ppbC)  
Parts per Billion by Volume (ppbv)  
Micrograms per Cubic Meter ( $\mu\text{g}/\text{m}^3$ )

Compound	Annual Average			Hourly Maximum		
	ppbC	ppbv	$\mu\text{g}/\text{m}^3$	ppbC	ppbv	$\mu\text{g}/\text{m}^3$
2-Methylhexane	0.24	0.03	0.14	2.83	0.40	1.66
3-Methylhexane	0.25	0.04	0.14	3.97	0.57	2.32
2-Methylpentane	0.55	0.09	0.32	4.07	0.68	2.39
3-Methylpentane	0.29	0.05	0.17	1.63	0.27	0.96
n-Nonane	0.09	0.01	0.05	0.61	0.07	0.36
n-Octane	0.09	0.01	0.05	0.53	0.07	0.31
n-Pentane	1.03	0.21	0.61	19.49	3.90	11.50
1-Pentene	0.15	0.03	0.09	0.50	0.10	0.29
c2-Pentene	0.02	0.00	0.01	0.39	0.08	0.22
t2-Pentene	0.04	0.01	0.02	0.52	0.10	0.30
a-Pinene	0.08	0.01	0.04	2.63	0.26	1.47
b-Pinene	0.00	0.00	0.00	0.26	0.03	0.14
Propane	2.50	0.83	1.50	14.88	4.96	8.95
n-Propylbenzene	0.06	0.01	0.03	1.15	0.13	0.63
Propylene	0.43	0.14	0.25	3.79	1.26	2.17
Styrene	0.07	0.01	0.04	0.48	0.06	0.26
Toluene	0.94	0.13	0.51	16.47	2.35	8.87
1 2 3-Trimethylbenzene	0.17	0.02	0.09	2.11	0.23	1.15
1 2 4-Trimethylbenzene	0.17	0.02	0.09	3.79	0.42	2.07
1 3 5-Trimethylbenzene	0.03	0.00	0.02	1.54	0.17	0.84
2 2 4-Trimethylpentane	0.40	0.05	0.23	5.52	0.69	3.22
2 3 4-Trimehtylpentane	0.11	0.01	0.06	0.86	0.11	0.50
n-Undecane	0.09	0.01	0.05	0.59	0.05	0.34
m,p-Xylene	0.42	0.05	0.23	4.51	0.56	2.45
o-Xylene	0.17	0.02	0.09	1.43	0.18	0.78
PAMHC	23.77	x	x	159.94	x	x
T-NMOC	26.01	x	x	161.55	x	x
Unknowns	2.24	x	x	14.80	x	x

The three main contributors to total PAMS hydrocarbons at the Rutgers site were isoprene, ethane, and propane, as shown in the pie chart in Figure 11-3. These compounds made up more than one-third of the total measured hydrocarbons during the summer ozone season. Crude oil refining and natural gas processing are major sources of ethane and propane. Isoprene is a biogenic compound released from plants, particularly trees.

**Figure 11-3**  
**Percentage of Components of Total Non-Methane Organic Carbon (TNMOC)**  
**for the 2022 PAMS Season in New Jersey**

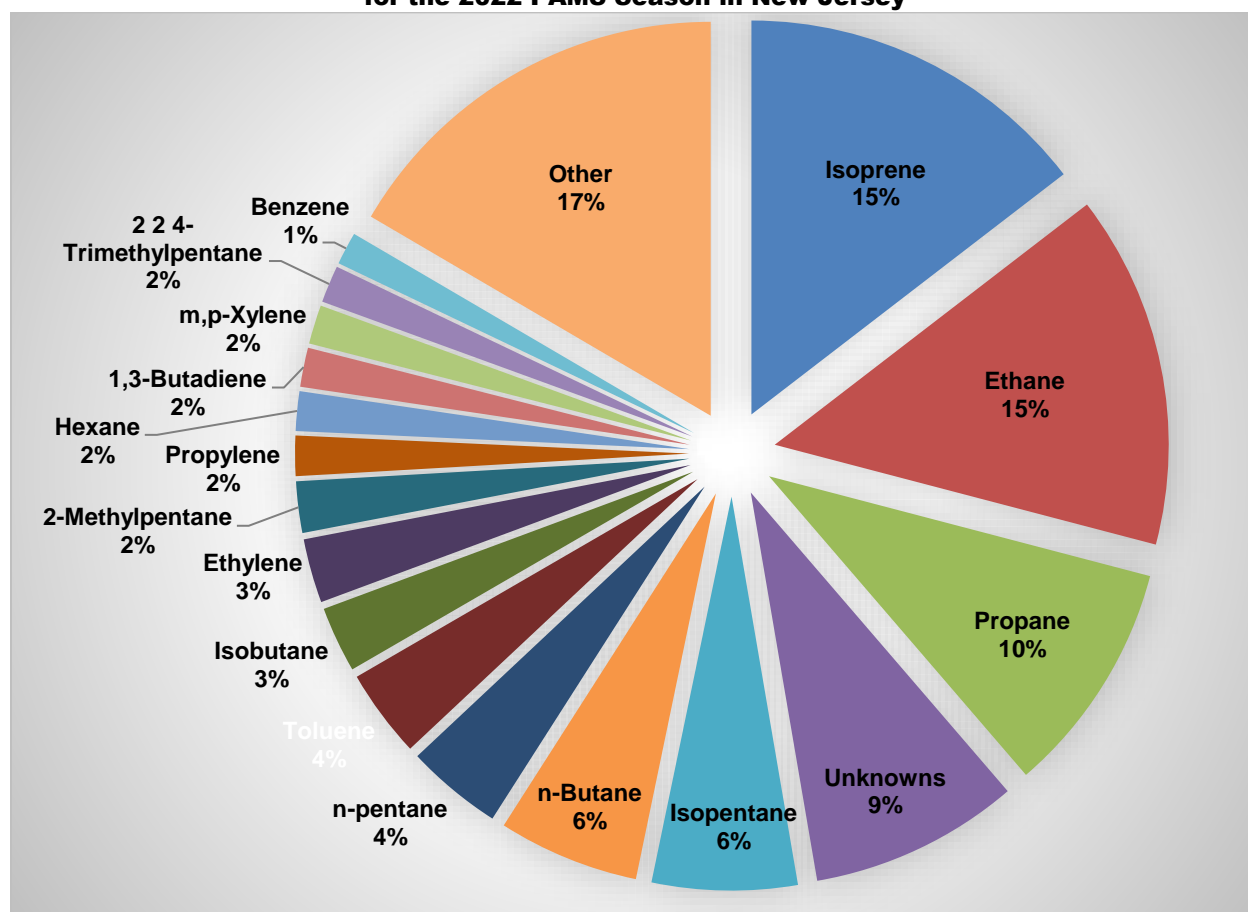
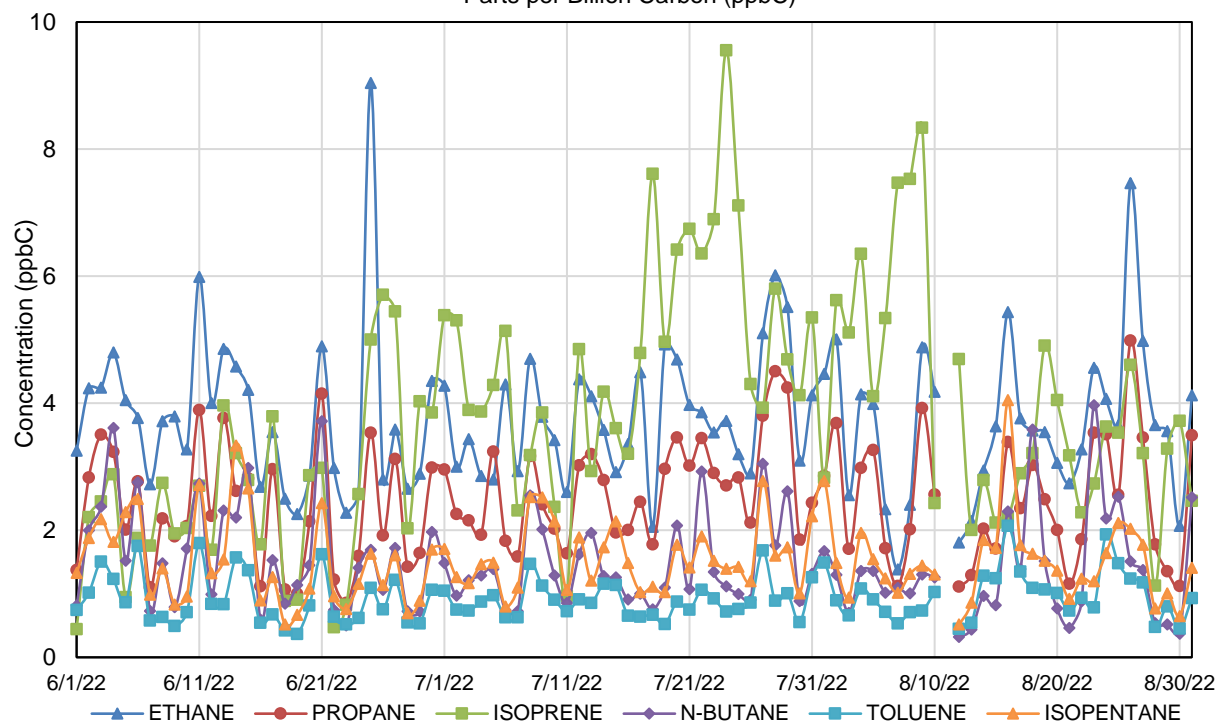
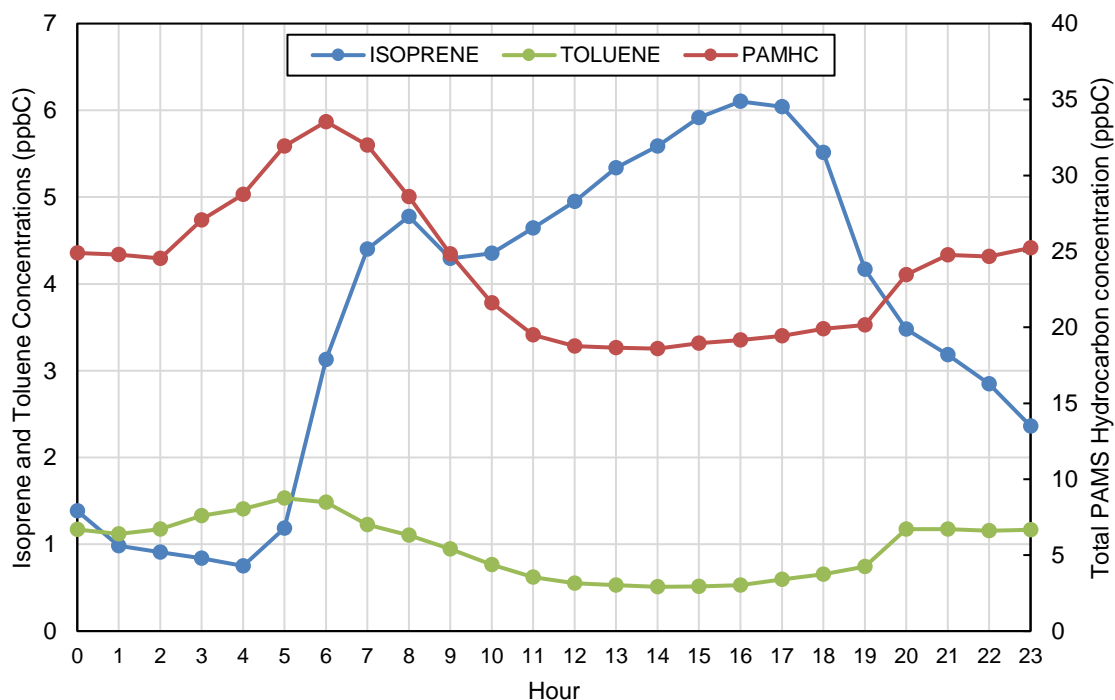


Figure 11-4 below is a comparison of the daily average concentrations of ethane, isoprene, propane, butane, toluene, and isopentane. Figure 11-5 shows the diurnal trend for isoprene, toluene, and the average of all of the other PAMS hydrocarbons (PAMHC). Most of the anthropogenic compounds show peaks in the early morning and late evening because the mixing layer height is lower at those times. In contrast, the biogenic compounds, like isoprene, tend to peak during the day, when sunlight drives plant photosynthesis.

**Figure 11-4**  
**2022 PAMS Daily Averages for Ethane, Isoprene, Propane, N-Butane, Toluene, and Isopentane in New Jersey**  
 Parts per Billion Carbon (ppbC)



**Figure 11-5**  
**2022 PAMS Diurnal Trends in New Jersey**  
**Hourly Average Concentrations**  
 Parts per Billion Carbon (ppbC)

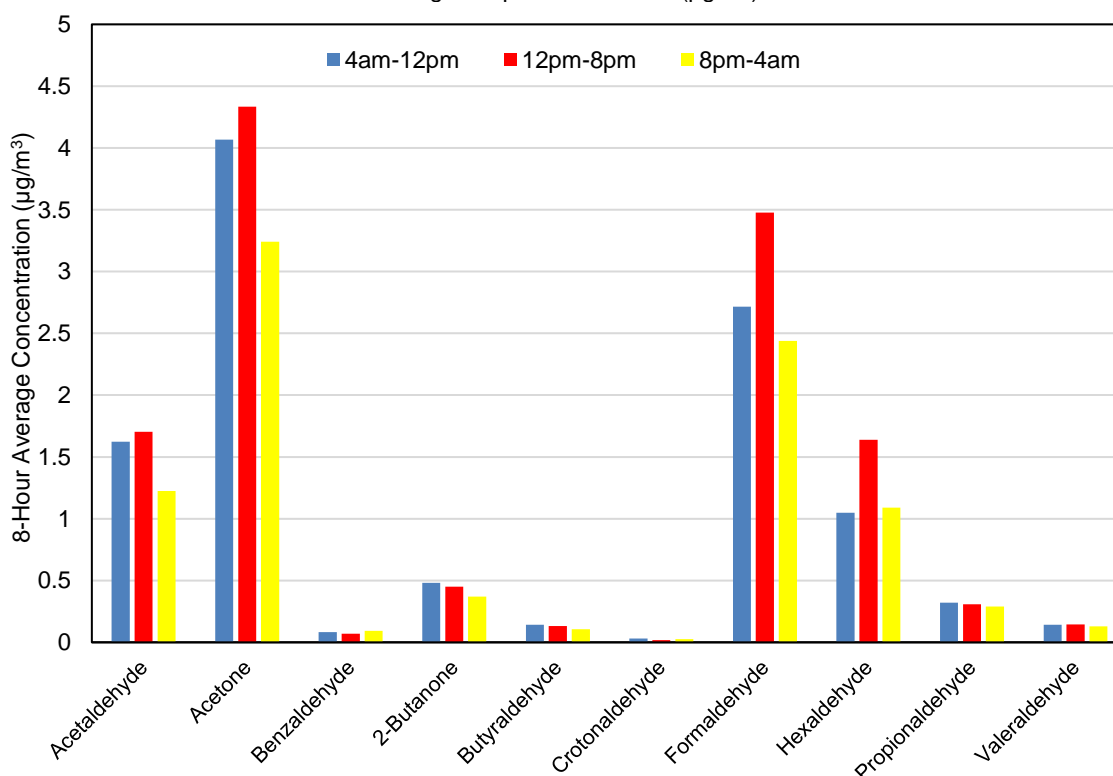


Carbonyls, a subset of VOCs, are also measured at the PAMS site during the summer. Every third day, three consecutive 8-hour samples are collected on cartridges, beginning at 4 am. These are sent to a laboratory for analysis according to USEPA's Method TO-11A protocols. Average daily concentrations for each of the ten carbonyls are shown in Table 11-2. The three carbonyls with the highest concentrations were acetone, formaldehyde, and acetaldehyde. All three of these showed higher concentrations during the afternoon. In general, the overnight hours had the lowest concentrations for most of the carbonyls (Figure 11-6).

**Table 11-2**  
**2022 PAMS 8-Hour Average Carbonyl Concentrations in New Jersey**  
 Micrograms per Cubic Meter ( $\mu\text{g}/\text{m}^3$ )

Carbonyl	4 am-12 pm	12 pm-8 pm	8 pm-4 am	Overall Average
Acetaldehyde	1.62	1.70	1.22	1.52
Acetone	4.07	4.33	3.24	3.88
Benzaldehyde	0.08	0.07	0.09	0.08
2-Butanone	0.48	0.45	0.37	0.43
Butyraldehyde	0.14	0.13	0.11	0.13
Crotonaldehyde	0.03	0.02	0.03	0.02
Formaldehyde	2.72	3.48	2.44	2.88
Hexaldehyde	1.05	1.64	1.09	1.26
Propionaldehyde	0.32	0.31	0.29	0.31
Valeraldehyde	0.14	0.14	0.13	0.14

**Figure 11-6**  
**2022 PAMS 8-Hour Average Carbonyl Concentrations in New Jersey**  
 Micrograms per Cubic Meter ( $\mu\text{g}/\text{m}^3$ )



## REFERENCES

U.S. Environmental Protection Agency (USEPA). Photochemical Assessment Monitoring Stations (PAMS). <https://www.epa.gov/amtic/photochemical-assessment-monitoring-stations-pams>. Accessed 7/20/23.

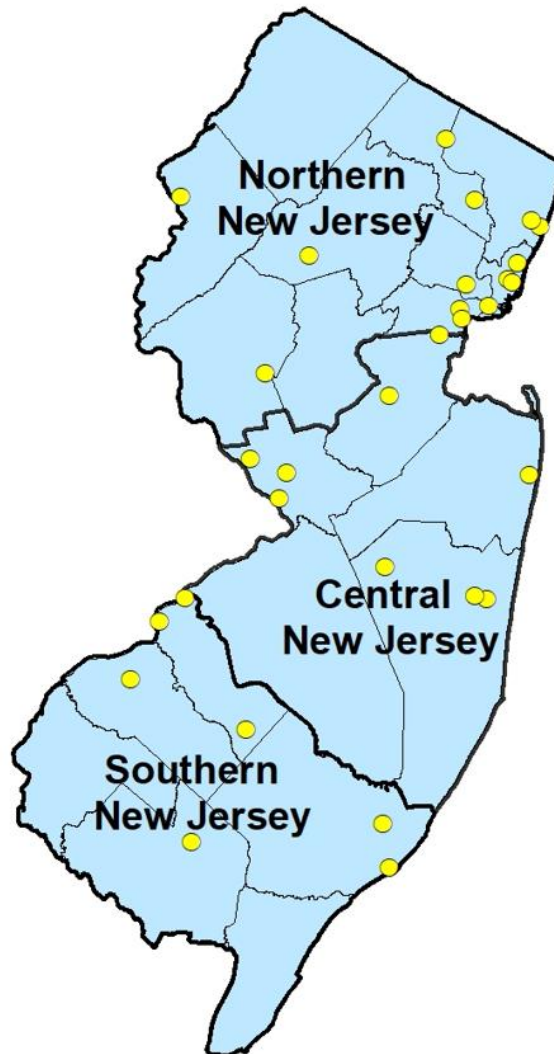
USEPA. Ground-Level Ozone Basics. <https://www.epa.gov/ground-level-ozone-pollution/ground-level-ozone-basics>. Accessed 7/20/23.



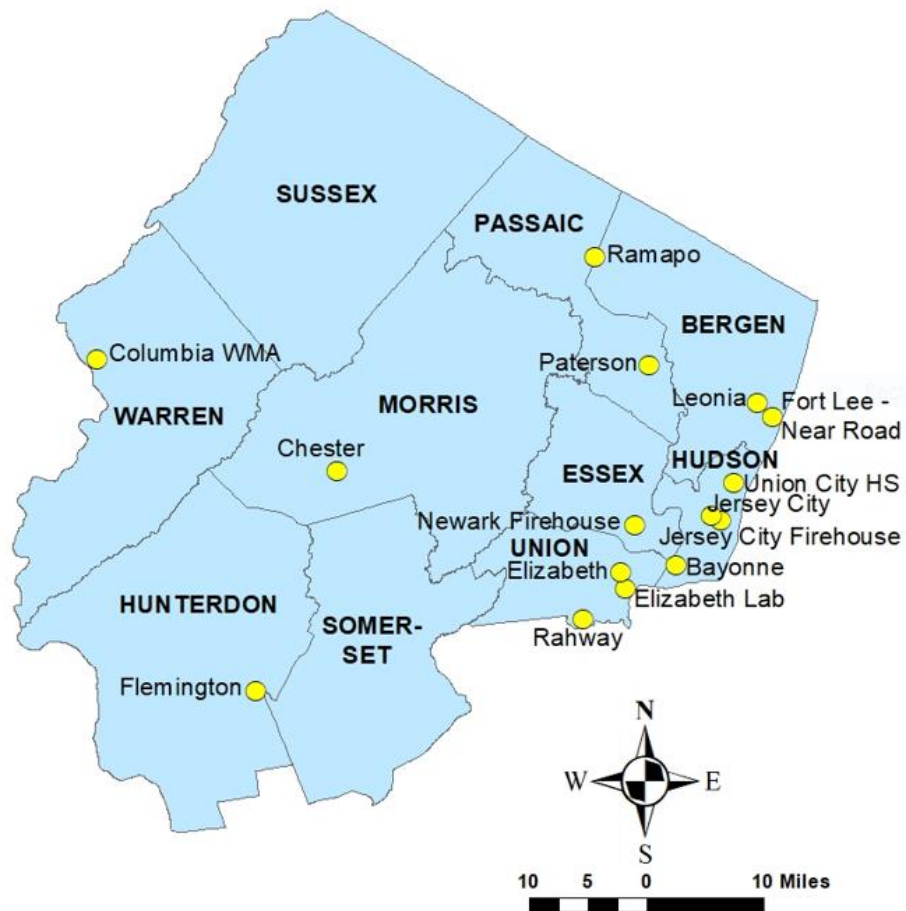
# Appendix A

## 2022 Air Monitoring Sites

New Jersey Department of Environmental Protection



**FIGURE A-1**  
**2022 NORTHERN NEW JERSEY**  
**AIR MONITORING SITES**



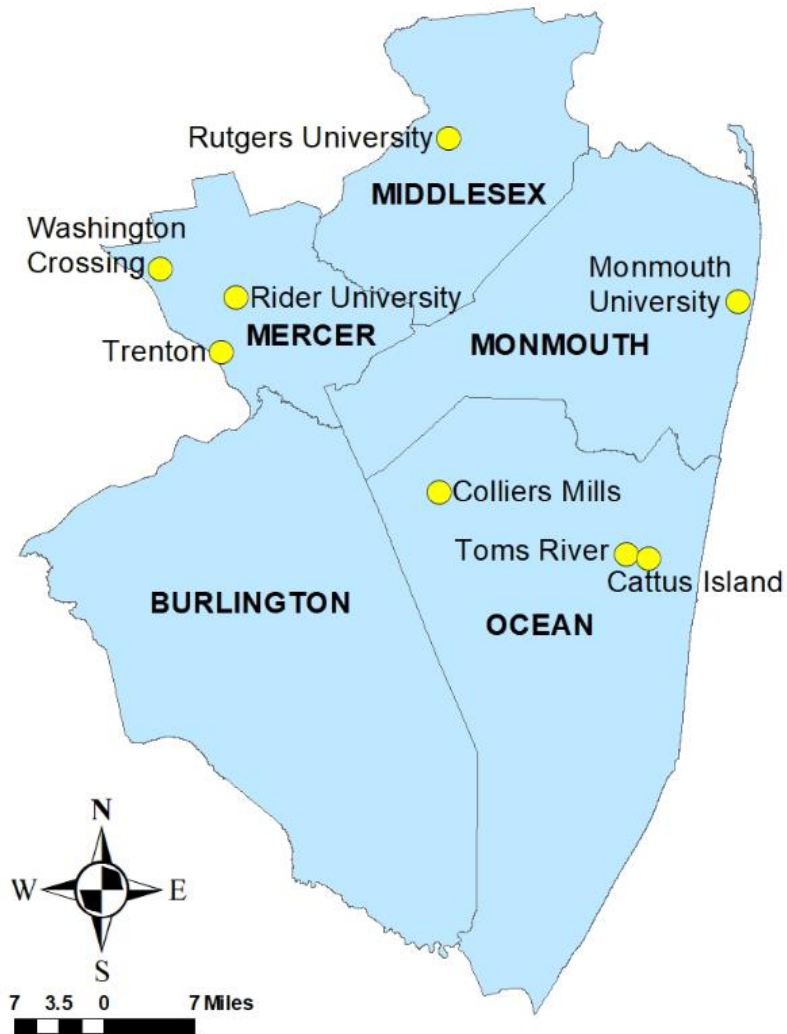


**Table A-1**  
**2022 Northern New Jersey Air Monitoring Sites**

County	Monitoring Site	AQS Code	Parameter(s) Measured <sup>1</sup>	Coordinates (Decimal degrees)		Address
				Latitude	Longitude	
BERGEN	Fort Lee Near Road	34 003 0010	CO, NO <sub>x</sub> , Beta, BTEX, BC, Met	40.853550	-73.966180	Hoyt Ave & Hudson St, south of toll plaza
	Leonia	34 003 0006	O <sub>3</sub>	40.870436	-73.991994	Overpeck Park, 40 Fort Lee Road
ESSEX	Newark Firehouse	34 013 0003	CO, O <sub>3</sub> , SO <sub>2</sub> , PM <sub>2.5</sub> , PM <sub>10</sub> , Spec, NO <sub>y</sub> , NO <sub>x</sub> , BTEX, Pb, Beta, BC, Met	40.720989	-74.192892	360 Clinton Avenue
HUDSON	Bayonne	34 017 0006	NO <sub>x</sub> , O <sub>3</sub> , SO <sub>2</sub> , BTEX, BC, Met	40.670250	-74.126081	Veterans Park, Park Rd at end of W. 25th St.
	Jersey City	34 017 1002	CO, NO <sub>x</sub> , SO <sub>2</sub>	40.731645	-74.066308	2828 John F. Kennedy Boulevard
	Jersey City Firehouse	34 017 1003	PM <sub>2.5</sub> , PM <sub>10</sub> , Beta	40.725454	-74.052290	Jersey City Fire Dept. Engine 5/Ladder 6, 355 Newark Avenue
	Union City High School	34 017 0008	Beta	40.770908	-74.036218	2500 John F. Kennedy Blvd.
HUNTERDON	Flemington	34 019 0001	O <sub>3</sub> , Met, Beta	40.515262	-74.806671	Raritan Twp. Municipal Utilities Authority, 365 Old York Road
MORRIS	Chester	34 027 3001	NO <sub>x</sub> , O <sub>3</sub> , SO <sub>2</sub> , PM <sub>2.5</sub> , Toxics, Spec	40.787628	-74.676301	Department of Public Works Bldg. #1, 50 North Road
PASSAIC	Paterson	34 031 0005	Beta	40.918381	-74.168092	Paterson Board of Health, 176 Broadway
	Ramapo	34 031 5001	O <sub>3</sub>	41.058617	-74.255544	Ramapo Station Fire Tower, Ramapo Park Drive, Wanaque
UNION	Elizabeth	34 039 0003	CO, SO <sub>2</sub>	40.662493	-74.214800	7 Broad Street
	Elizabeth Lab	34 039 0004	CO, NO <sub>x</sub> , SO <sub>2</sub> , Met, PM <sub>2.5</sub> , Toxics, Hg, Spec, BTEX, BC, Beta	40.641440	-74.208365	New Jersey Turnpike Interchange 13 Toll Plaza
	Rahway	34 039 2003	Beta	40.603943	-74.276174	Rahway Fire Department, 1300 Main Street
WARREN	Columbia	34 041 0007	NO <sub>x</sub> , O <sub>3</sub> , SO <sub>2</sub> , Met, Beta	40.924580	-75.067815	Columbia Wildlife Management Area, 105 Delaware Road, Knowlton Twp.

<sup>1</sup> See abbreviations and acronyms in Table A-4 (page A-8).

**FIGURE A-2**  
**2022 CENTRAL NEW JERSEY**  
**AIR MONITORING SITES**

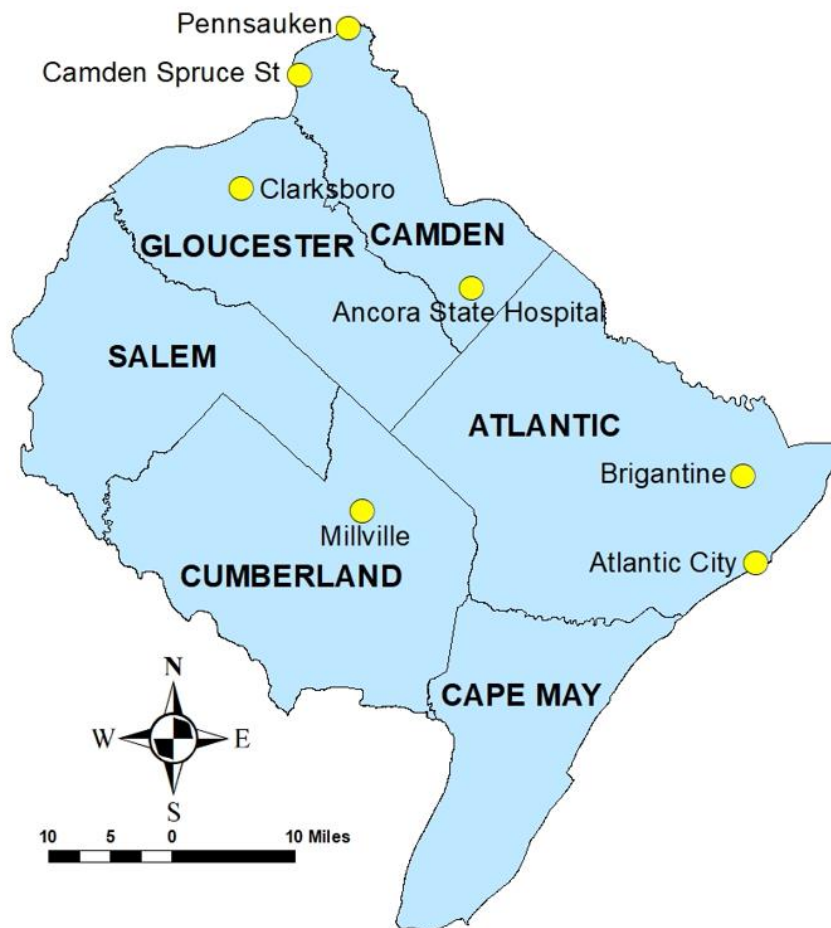


**Table A-2  
2022 Central New Jersey Air Monitoring Sites**

County	Monitoring Site	AQS Code	Parameter(s) Measured <sup>1</sup>	Coordinates (Decimal degrees)		Address
				Latitude	Longitude	
MERCER	Rider University	34 021 0005	O <sub>3</sub> , Met, Beta	40.283092	-74.742644	Athletic Fields, off of 2083 Lawrenceville Rd, Lawrence Twp.
	Trenton Library	34 021 0008	Beta	40.222411	-74.763167	120 Academy Street
	Washington Crossing	N/A	ACID	40.315359	-74.853613	Washington Crossing State Park, Philips Farm Group Area, 1239 Bear Tavern Rd., Titusville
MIDDLESEX	Rutgers University	34 023 0011	NO <sub>2</sub> , NO, NO <sub>y</sub> , O <sub>3</sub> , PAMS, Beta, PM <sub>2.5</sub> , Toxics, Spec, Hg, Met	40.462182	-74.429439	Vegetable Farm 3, 67 Ryders Lane, East Brunswick
MONMOUTH	Monmouth University	34 025 0005	O <sub>3</sub>	40.277647	-74.005100	Edison Science Hall, off of 400 Cedar Avenue, West Long Branch
OCEAN	Cattus Island	N/A	ACID	39.989636	-74.134132	Cattus Island County Park behind Administrative Office, end of Bandon Road, Toms River
	Colliers Mills	34 029 0006	O <sub>3</sub>	40.064830	-74.444050	JPTD Training Center, south of Success Rd., east of Hawkin Rd., Jackson Twp.
	Toms River	34 029 2002	Beta	39.994908	-74.170447	Hooper Avenue Elementary School, 1517 Hooper Avenue

<sup>1</sup> See abbreviations and acronyms in Table A-4 (page A-8).

**FIGURE A-3**  
**2022 SOUTHERN NEW JERSEY**  
**AIR MONITORING SITES**



**Table A-3**  
**2022 Southern New Jersey Air Monitoring Sites**

County	Monitoring Site	AQS Code	Parameter(s) Measured <sup>1</sup>	Coordinates (Decimal degrees)		Address
				Latitude	Longitude	
ATLANTIC	Atlantic City	34 001 1006	PM <sub>2.5</sub>	39.363260	-74.431000	Atlantic Cape Community College, 1535 Bacharach Boulevard
	Brigantine	34 001 0006	Visibility, O <sub>3</sub> , SO <sub>2</sub> , Beta, PM <sub>2.5</sub> , ACID <sup>2</sup>	39.464872	-74.448736	Edwin B. Forsythe National Wildlife Refuge Visitor Center, 800 Great Creek Road, Galloway
CAMDEN	Ancora State Hospital	34 007 1001	O <sub>3</sub>	39.684250	-74.861491	301 Spring Garden Road, Hammonton
	Camden Spruce Street	34 007 0002	CO, NO <sub>x</sub> , O <sub>3</sub> , SO <sub>2</sub> , PM <sub>2.5</sub> , PM <sub>10</sub> , Spec, BTEX, BC, Toxics, Met, Beta	39.934446	-75.125291	266-298 Spruce Street
	Pennsauken	34 007 1007	PM <sub>2.5</sub>	39.989036	-75.050008	Camden Water Inc., 8999 Zimmerman Ave.
CUMBERLAND	Millville	34 011 0007	NO <sub>x</sub> , O <sub>3</sub> , Beta	39.422273	-75.025204	Behind 4401 S. Main Road
GLOUCESTER	Clarksboro	34 015 0002	O <sub>3</sub> , PM <sub>2.5</sub>	39.800339	-75.212119	Shady Lane Nursing Home, 256 County House Road

<sup>1</sup> See abbreviations and acronyms in Table A-4 (page A-8).

<sup>2</sup> United States Fish and Wildlife Service-Air Quality Branch (USFWS-AQB) is responsible for ACID sample collection.

**Table A-4**  
**Abbreviations & Acronyms**

ACID	Acid deposition
AQS	Air Quality System (USEPA's ambient air quality data repository)
BC	Black carbon measured by aethalometer
Beta	Real-time PM <sub>2.5</sub> analyzer
BTEX	Measurement of benzene, toluene, ethylbenzene and xylenes
CO	Carbon monoxide
Hg	Mercury
Met	Meteorological parameters
NO <sub>x</sub>	Nitrogen dioxide and nitric oxide
NO <sub>y</sub>	Total reactive oxides of nitrogen
O <sub>3</sub>	Ozone
PAMS	Photochemical Assessment Monitoring Station, measures ozone precursors
Pb	Lead
PM <sub>2.5</sub>	Fine particles (2.5 microns or less) collected by a Federal Reference Method PM <sub>2.5</sub> filter sampler
PM <sub>10</sub>	Inhalable particles (10 microns or less) collected by a Federal Reference Method PM <sub>10</sub> filter sampler
SO <sub>2</sub>	Sulfur dioxide
Spec	Speciated fine particles (2.5 microns or less)
Toxics	Air toxics
Visibility	Measured by nephelometer



## Appendix B: 2022 Fine Particulate Speciation Summary

New Jersey Department of Environmental Protection

### CHEMICAL SPECIATION NETWORK

The data presented in this section is collected as part of the U.S Environmental Protection Agency Chemical Speciation Network. This program uses 24-hour filter-based PM<sub>2.5</sub> samples to determine the concentrations of the chemicals (metals, ions and carbon compounds) that make up the particles. Teflon filters are analyzed for 33 elements, nylon filters are analyzed for ions, and quartz filters are analyzed for carbon. For details, see <https://www.epa.gov/amtic/chemical-speciation-network-csn>.

New Jersey has been collecting this data since 2001, as indicated in Table B-1. A map of the current monitoring locations can be seen in Figure B-1.

**Table B-1**  
**Speciation Monitoring in New Jersey**

Monitoring Site	Location	Start	End	Frequency
Camden Lab (Camden 1)	Copewood & Davis Sts.	2001	2008	---
Camden Spruce St (Camden 2)	266-298 Spruce St.	2013		Every 6th day
Chester	50 North Rd.	2001		Every 6th day
Elizabeth Lab	NJ Turnpike Exit 13 Toll Plaza	2001		Every 3 <sup>rd</sup> day
Newark Firehouse**	360 Clinton Ave.	2010	2022	Every 3 <sup>rd</sup> day
New Brunswick*	Log Cabin Rd.	2001	2015	---
Rutgers*	67 Ryders Lane	2016		Every 3 <sup>rd</sup> day

\*Also has a co-located monitor for quality assurance.

\*\*Note that the Newark Firehouse station was shut down on 9/26/2022.

A 24-hour sample is collected every third or sixth day, and sent to USEPA's contract laboratory for analysis. Tables B-2 through B-6 present the 2022 averages, maximums, and the percent of samples detected. Some analytes are regularly below the limit detectable by the lab. In Tables B-2 through B-6, annual averages in italics have been calculated with fewer than 50% of samples detected.

The most recent detection limits are listed in Table B-7.

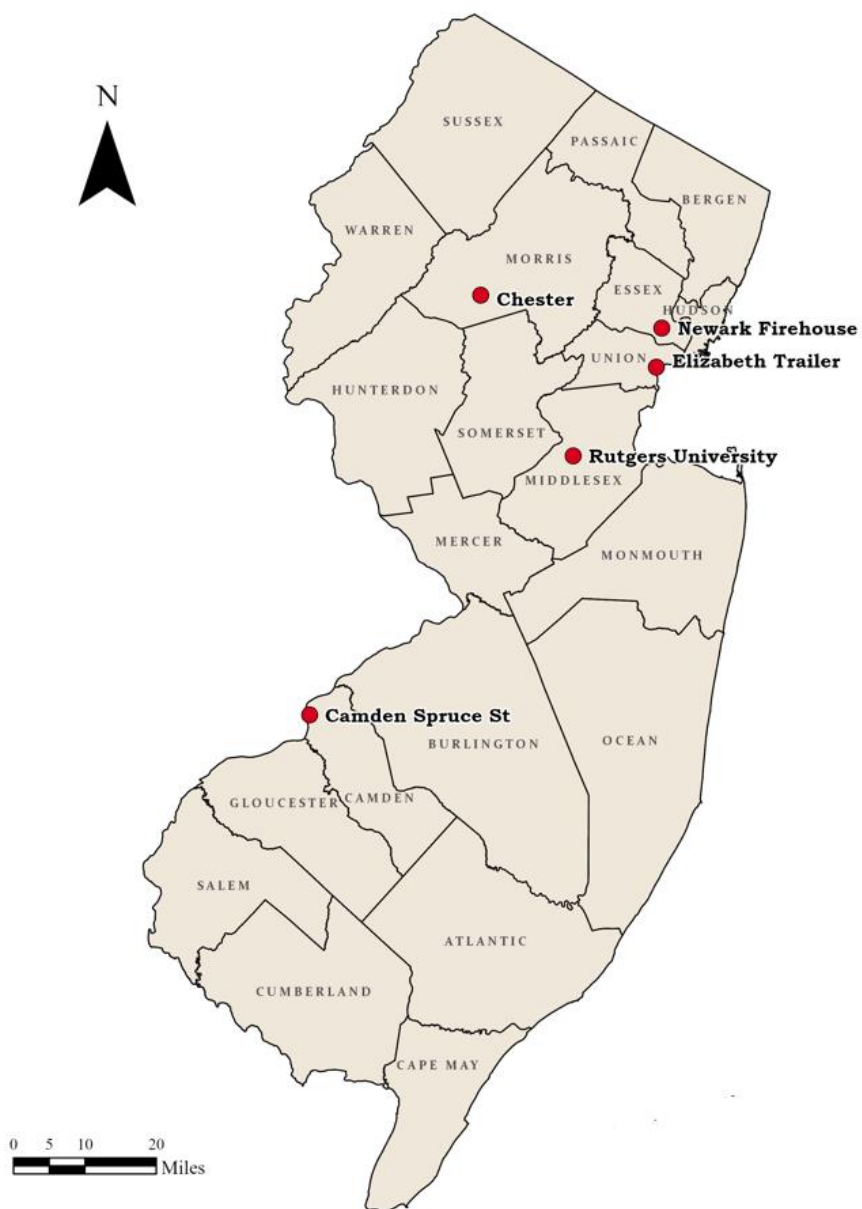
### TRENDS

Trends in the concentrations of elemental carbon and organic carbon are shown in Figure B-2 and B-3. Nitrate and sulfate concentrations can be found in Figures B-4 and B-5.

Carbon compounds make up a large portion of particulate matter. They are released into the air primarily from combustion of fuels. Significant sources of elemental carbon are on-road vehicles and non-road equipment. While these sources also emit substantial quantities of organic carbon, other major sources of organic carbon include the burning of wood and other solid fuels.

Nitrate and sulfate are other major components of ambient particles. They are mostly formed by chemical reactions in the atmosphere by nitrogen and sulfur compounds emitted by fuel combustion. In addition to contributing to health effects as particulate matter, they affect visibility and cause acidification of surface water, rain and soil.

**Figure B-1**  
**2022 Fine Particulate Speciation Monitoring Network**





**Table B-2**  
**2022 Fine Particulate Speciation Concentrations**  
**CAMDEN SPRUCE STREET NJ**  
Micrograms per Cubic Meter ( $\mu\text{g}/\text{m}^3$ )

	Species	Annual Average*	Maximum Daily Average	% Samples Detected
1	Aluminum	0.034	0.280	79
2	Ammonium Ion	0.380	1.553	100
3	Antimony	0.001	0.024	49
4	Arsenic	0	0.0001	25
5	Barium	0.006	0.066	61
6	Bromine	0.001	0.028	36
7	Cadmium	0	0.020	44
8	Calcium	0.060	0.448	100
9	Carbon, Elemental	0.823	2.298	100
10	Carbon, Organic	2.019	4.190	100
11	Cerium	0.005	0.069	49
12	Cesium	0.001	0.052	41
13	Chloride	0.566	3.318	100
14	Chlorine	0.133	1.900	92
15	Chromium	0.002	0.013	87
16	Cobalt	0	0.002	28
17	Copper	0.007	0.032	84
18	Indium	0.003	0.024	61
19	Iron	0.203	1.079	100
20	Lead	0.006	0.023	74
21	Magnesium	0.018	0.096	67
22	Manganese	0.006	0.064	80
23	Nickel	0.001	0.019	77
24	Nitrate	0.962	4.226	100
25	Phosphorus	0.0004	0.005	82
26	Potassium	0.150	0.801	100
27	Potassium Ion	0.125	0.799	100
28	Rubidium	0.0002	0.006	43
29	Selenium	0.0003	0.005	49
30	Silicon	0.074	0.547	98
31	Silver	0.002	0.020	57
32	Sodium	0.136	1.026	69
33	Sodium Ion	0.109	0.613	100
34	Strontium	0.001	0.020	67
35	Sulfate	0.877	2.319	100
36	Sulfur	0.301	0.981	100
37	Tin	0.004	0.026	62
38	Titanium	0.005	0.037	98
39	Vanadium	0.0001	0.001	28
40	Zinc	0.025	0.112	100
41	Zirconium	0	0.023	39

**Table B-3**  
**2022 Fine Particulate Speciation Concentrations**  
**CHESTER NJ**  
Micrograms per Cubic Meter ( $\mu\text{g}/\text{m}^3$ )

	Species	Annual Average*	Maximum Daily Average	% Samples Detected
1	Aluminum	0.010	0.176	63
2	Ammonium Ion	0.226	1.351	100
3	Antimony	0.003	0.031	63
4	Arsenic	0	0.0001	42
5	Barium	0.003	0.046	56
6	Bromine	0.0002	0.003	20
7	Cadmium	0.001	0.026	49
8	Calcium	0.014	0.058	100
9	Carbon, Elemental	0.323	0.898	100
10	Carbon, Organic	1.503	3.746	100
11	Cerium	0.002	0.056	51
12	Cesium	0.005	0.055	56
13	Chloride	0.043	0.454	100
14	Chlorine	0.005	0.183	49
15	Chromium	0.001	0.006	59
16	Cobalt	0.0001	0.007	47
17	Copper	0.002	0.017	64
18	Indium	0.002	0.027	54
19	Iron	0.023	0.088	100
20	Lead	0.001	0.013	51
21	Magnesium	0.012	0.111	46
22	Manganese	0.001	0.012	53
23	Nickel	0.0003	0.002	61
24	Nitrate	0.731	4.643	100
25	Phosphorus	0.0001	0.003	61
26	Potassium	0.036	0.381	100
27	Potassium Ion	0.024	0.371	98
28	Rubidium	0.0001	0.005	47
29	Selenium	0.0002	0.004	58
30	Silicon	0.025	0.137	90
31	Silver	0.002	0.027	56
32	Sodium	0.045	0.497	59
33	Sodium Ion	0.025	0.318	97
34	Strontium	0.001	0.007	69
35	Sulfate	0.695	1.720	100
36	Sulfur	0.232	0.654	100
37	Tin	0.001	0.022	51
38	Titanium	0.002	0.010	83
39	Vanadium	0.0001	0.002	24
40	Zinc	0.007	0.033	100
41	Zirconium	0	0.020	41

**Table B-4**  
**2022 Fine Particulate Speciation Concentrations**  
**ELIZABETH LAB NJ**  
Micrograms per Cubic Meter ( $\mu\text{g}/\text{m}^3$ )

	Species	Annual Average*	Maximum Daily Average	% Samples Detected
1	Aluminum	0.027	0.169	80
2	Ammonium Ion	0.297	1.502	100
3	Antimony	0.002	0.034	52
4	Arsenic	0	0.0001	23
5	Barium	0.013	0.104	73
6	Bromine	0.001	0.008	31
7	Cadmium	0.001	0.023	50
8	Calcium	0.033	0.131	100
9	Carbon, Elemental	0.999	2.919	100
10	Carbon, Organic	2.034	4.201	100
11	Cerium	0.003	0.074	52
12	Cesium	0.001	0.044	42
13	Chloride	0.128	0.822	100
14	Chlorine	0.019	0.734	86
15	Chromium	0.002	0.020	84
16	Cobalt	0	0.003	14
17	Copper	0.004	0.094	81
18	Indium	0.002	0.035	52
19	Iron	0.129	0.408	100
20	Lead	0.002	0.023	73
21	Magnesium	0.011	0.181	48
22	Manganese	0.002	0.013	75
23	Nickel	0.001	0.005	88
24	Nitrate	1.038	4.999	100
25	Phosphorus	0.001	0.009	73
26	Potassium	0.056	2.151	100
27	Potassium Ion	0.041	2.200	98
28	Rubidium	0.0002	0.006	51
29	Selenium	0.0001	0.005	48
30	Silicon	0.052	0.222	100
31	Silver	0.002	0.022	64
32	Sodium	0.076	0.591	70
33	Sodium Ion	0.067	0.356	99
34	Strontium	0.001	0.054	68
35	Sulfate	0.857	1.955	100
36	Sulfur	0.294	0.746	100
37	Tin	0.002	0.028	57
38	Titanium	0.006	0.025	98
39	Vanadium	0.0001	0.004	19
40	Zinc	0.013	0.053	100
41	Zirconium	0.002	0.020	57

**Table B-5**  
**2022 Fine Particulate Speciation Concentrations**  
**NEWARK FIREHOUSE NJ**  
Micrograms per Cubic Meter ( $\mu\text{g}/\text{m}^3$ )

	Species	Annual Average*	Maximum Daily Average	% Samples Detected
1	Aluminum	0.023	0.216	70
2	Ammonium Ion	0.301	1.426	100
3	Antimony	0.002	0.035	52
4	Arsenic	0	0.00004	19
5	Barium	0.010	0.191	64
6	Bromine	0.0003	0.004	26
7	Cadmium	0.002	0.023	56
8	Calcium	0.031	0.185	100
9	Carbon, Elemental	0.609	1.499	100
10	Carbon, Organic	1.983	4.471	100
11	Cerium	0.002	0.048	57
12	Cesium	0.002	0.057	53
13	Chloride	0.143	1.399	100
14	Chlorine	0.031	1.235	84
15	Chromium	0.001	0.007	80
16	Cobalt	0	0.002	24
17	Copper	0.006	0.144	69
18	Indium	0.002	0.035	52
19	Iron	0.071	0.327	100
20	Lead	0.003	0.017	74
21	Magnesium	0.018	0.372	67
22	Manganese	0.001	0.007	65
23	Nickel	0.001	0.004	76
24	Nitrate	1.026	5.452	100
25	Phosphorus	0.0005	0.008	69
26	Potassium	0.079	3.542	99
27	Potassium Ion	0.068	3.896	100
28	Rubidium	0.0001	0.005	49
29	Selenium	0.0003	0.004	48
30	Silicon	0.052	0.357	95
31	Silver	0.001	0.034	57
32	Sodium	0.069	0.447	65
33	Sodium Ion	0.078	0.499	100
34	Strontium	0.002	0.097	65
35	Sulfate	0.821	2.870	100
36	Sulfur	0.276	0.861	100
37	Tin	0.003	0.025	66
38	Titanium	0.004	0.026	93
39	Vanadium	0.0002	0.004	28
40	Zinc	0.013	0.052	100
41	Zirconium	0.002	0.027	58

Newark Firehouse had no data after 9/26/2023, when the site was shut down.

**Table B-6**  
**2022 Fine Particulate Speciation Concentrations**  
**RUTGERS UNIVERSITY NJ**  
Micrograms per Cubic Meter ( $\mu\text{g}/\text{m}^3$ )

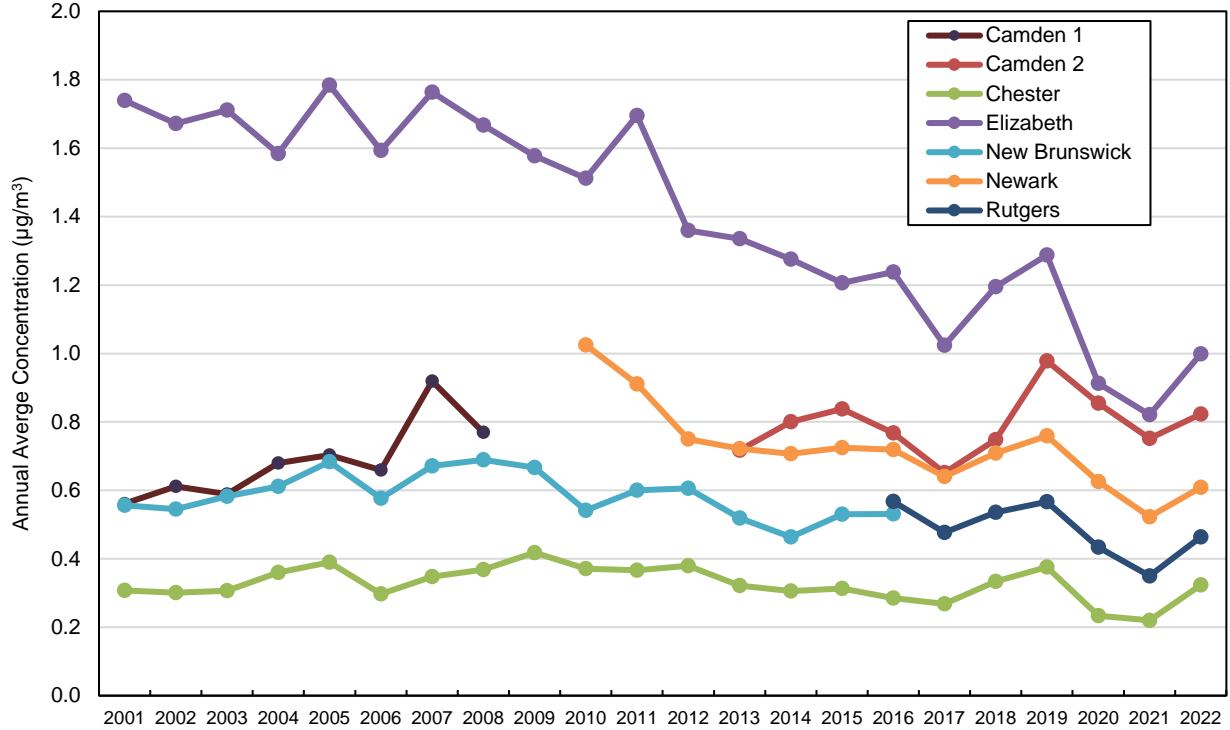
	Species	Annual Average*	Maximum Daily Average	% Samples Detected
1	Aluminum	0.017	0.085	73
2	Ammonium Ion	0.213	1.408	99
3	Antimony	0.001	0.030	55
4	Arsenic	0	0.0001	21
5	Barium	0.006	0.088	63
6	Bromine	0.0003	0.004	22
7	Cadmium	0.0002	0.015	53
8	Calcium	0.024	0.076	100
9	Carbon, Elemental	0.464	1.520	100
10	Carbon, Organic	1.802	4.180	100
11	Cerium	0.003	0.056	48
12	Cesium	0.005	0.071	57
13	Chloride	0.095	0.617	99
14	Chlorine	0.019	0.499	81
15	Chromium	0.002	0.009	80
16	Cobalt	0	0.005	29
17	Copper	0.003	0.066	62
18	Indium	0.002	0.029	51
19	Iron	0.054	0.171	100
20	Lead	0.003	0.028	72
21	Magnesium	0.015	0.194	50
22	Manganese	0.003	0.051	67
23	Nickel	0.001	0.005	80
24	Nitrate	0.779	4.769	100
25	Phosphorus	0.001	0.013	73
26	Potassium	0.052	1.607	100
27	Potassium Ion	0.037	1.631	99
28	Rubidium	0.0001	0.007	48
29	Selenium	0.0002	0.003	53
30	Silicon	0.043	0.215	98
31	Silver	0.001	0.027	52
32	Sodium	0.062	0.584	63
33	Sodium Ion	0.061	0.466	99
34	Strontium	0.001	0.039	58
35	Sulfate	0.739	1.773	100
36	Sulfur	0.250	0.690	100
37	Tin	0.002	0.040	56
38	Titanium	0.003	0.013	87
39	Vanadium	0.0001	0.002	25
40	Zinc	0.013	0.178	100
41	Zirconium	0.0004	0.027	48

**Table B-7**  
**CSN Average Minimum Detection Limits (MDL) ( $\mu\text{g}/\text{m}^3$ )**

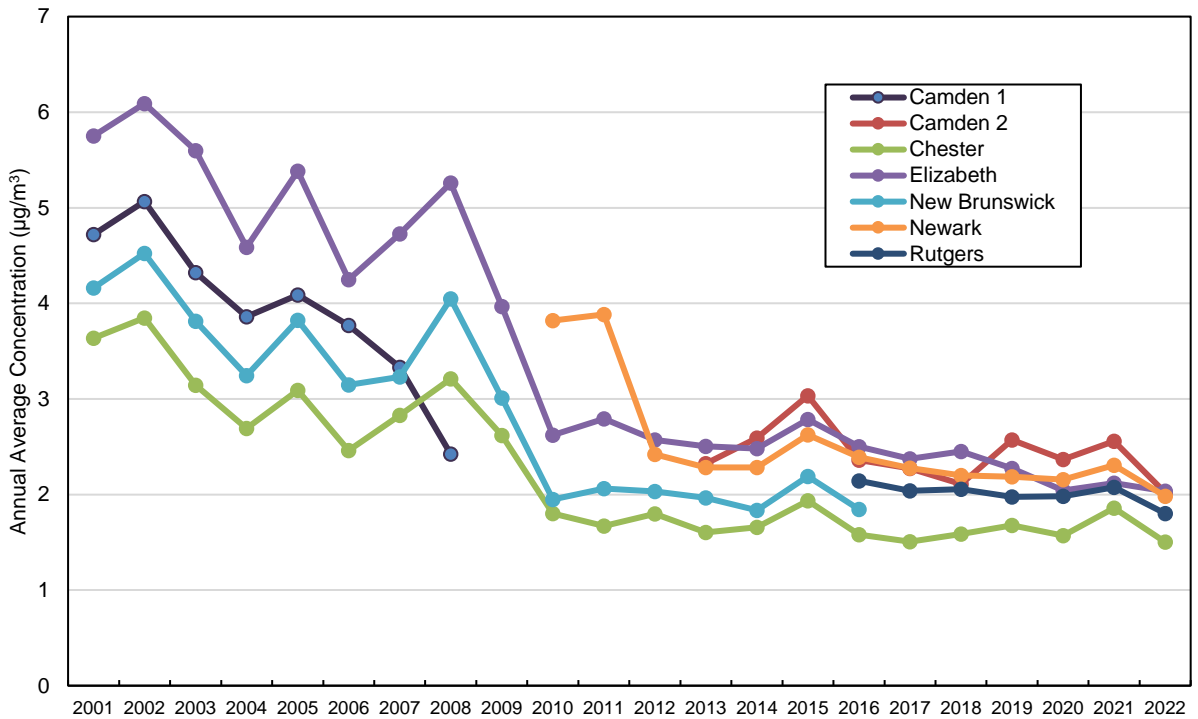
	Species	MDL ( $\mu\text{g}/\text{m}^3$ )
1	Aluminum	0.023
2	Ammonium Ion	0.013
3	Antimony	0.016
4	Arsenic	0.0001
5	Barium	0.028
6	Bromine	0.0001
7	Cadmium	0.014
8	Calcium	0.010
9	Carbon, Elemental	0.0003
10	Carbon, Organic	0.644
11	Cerium	0.036
12	Cesium	0.027
13	Chloride	0.025
14	Chlorine	0.004
15	Chromium	0.002
16	Cobalt	0.002
17	Copper	0.004
18	Indium	0.015
19	Iron	0.009
20	Lead	0.007
21	Magnesium	0.045
22	Manganese	0.003
23	Nickel	0.001
24	Nitrate	0.039
25	Phosphorus	0.002
26	Potassium	0.005
27	Potassium Ion	0.013
28	Rubidium	0.003
29	Selenium	0.003
30	Silicon	0.014
31	Silver	0.013
32	Sodium	0.081
33	Sodium Ion	0.014
34	Strontium	0.003
35	Sulfate	0.029
36	Sulfur	0.001
37	Tin	0.016
38	Titanium	0.003
39	Vanadium	0.001
40	Zinc	0.002
41	Zirconium	0.014

Most recent detection limit information from: Chemical Speciation Network (CSN) Annual Quality Report, prepared by Air Quality Research Center, University of California, Davis, pp. 32-33, 01/31/2023.  
[https://www.epa.gov/system/files/documents/2023-02/CSN\\_2021AnnualQualityReport.pdf](https://www.epa.gov/system/files/documents/2023-02/CSN_2021AnnualQualityReport.pdf)

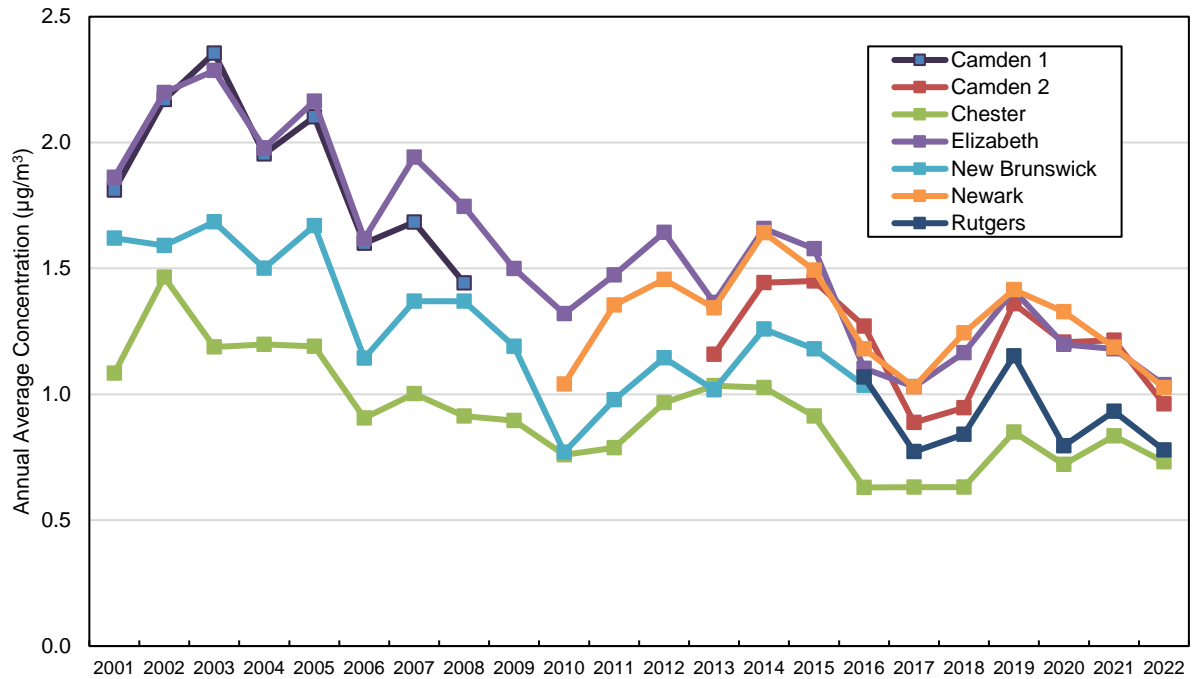
**Figure B-2**  
**ELEMENTAL CARBON – New Jersey Monitored Concentrations**



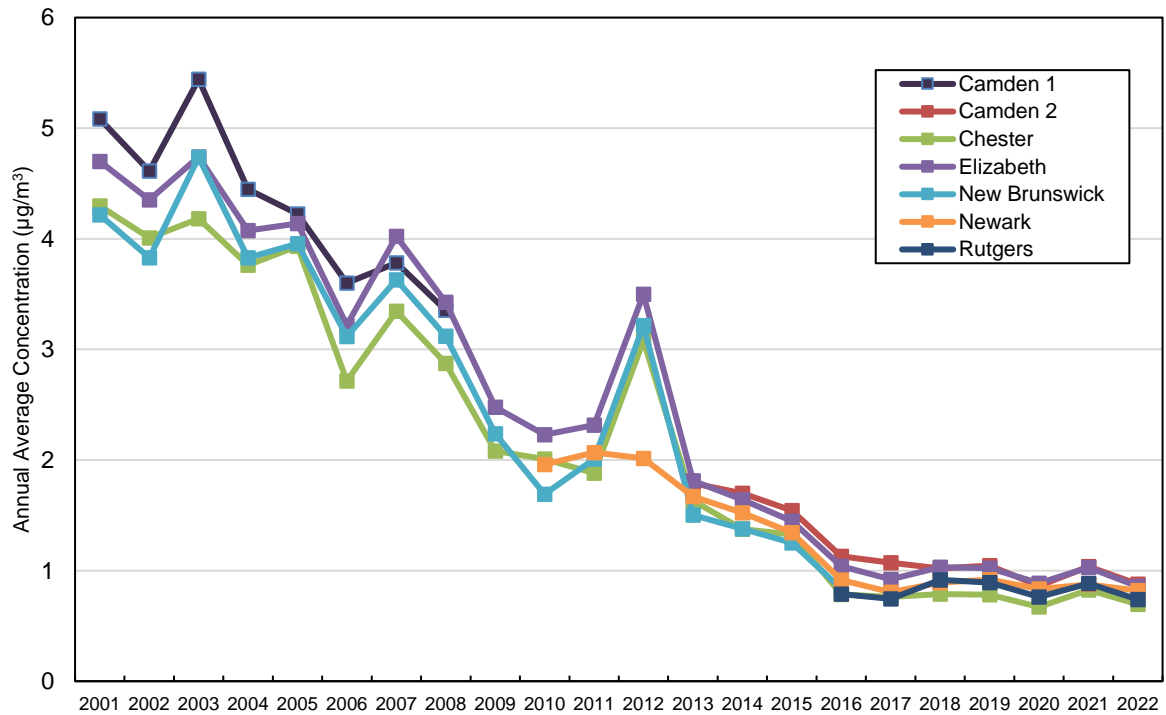
**Figure B-3**  
**ORGANIC CARBON – New Jersey Monitored Concentrations**



**Figure B-4**  
**NITRATE – New Jersey Monitored Concentrations**



**Figure B-5**  
**SULFATE – New Jersey Monitored Concentrations**







# Appendix C: Glossary

New Jersey Department of Environmental Protection

## GLOSSARY OF AIR MONITORING ABBREVIATIONS AND TERMS

**Air Quality Index (AQI)** – a national rating system for reporting daily air quality to the public

**Air toxics** – air pollutants that may cause adverse health effects in humans, but do not have a NAAQS

**Ambient air** – air in outdoor areas that are accessible to the general public

**AQS** – Air Quality System, USEPA's nationwide database for air quality data

**BAM** – NJDEP Bureau of Air Monitoring

**CAMNET** – a network of real-time cameras established to raise public awareness of the effects of air pollution on visibility.

**Canister** – a stainless steel container used for collecting an air sample to be analyzed in a lab.

**Carcinogen** – a chemical which may cause cancer

**CO** – Carbon monoxide, a criteria pollutant

**Continuous monitor** – an instrument that collects data around the clock, throughout the year, and transmits the data to a central data acquisition system every minute or hour.

**Criteria pollutant** – an air pollutant for which a National Ambient Air Quality Standard (NAAQS) has been set (ozone, particulate matter, nitrogen dioxide, sulfur dioxide, carbon monoxide & lead).

**CSN** – USEPA's Chemical Speciation Network

**Design value (DV)** – a pollutant-specific statistic applied to air monitoring data that determines whether a National Ambient Air Quality Standard is being met or exceeded

**Detection limit** – lowest quantity of a chemical that can be reliably measured by a laboratory method or sampling instrument

**Fine particles** – see PM<sub>2.5</sub>

**Hazardous Air Pollutant (HAP)** – an “air toxic” pollutant that is listed in the 1990 Clean Air Act Amendments and is subject to emissions limits for specific source types.

**Health benchmark** – a chemical-specific air concentration above which there may be human health concerns

**Inhalable particles** – see PM<sub>10</sub>

**Lead** – see Pb

**Manual sampler** – an instrument that collects an air sample over a specific time period on a filter, adsorbent cartridge or canister, which is then manually retrieved for analysis.

**Median** – the middle value in a list of numerical values, sorted in ascending or descending order

**NAAQS** – National Ambient Air Quality Standard; for specific air pollutants, a concentration allowable in ambient air.

**NJAAQS** – New Jersey Ambient Air Quality Standard; for specific air pollutants, a concentration allowable in ambient air (N.J. Administrative Code, Title 7, Chapter 27, Subchapter 13).

**NJDEP** – New Jersey Department of Environmental Protection

**NO** – Nitric oxide

**NO<sub>2</sub>** – Nitrogen dioxide, a criteria pollutant

**NO<sub>x</sub>** – Oxides of nitrogen

**NO<sub>y</sub>** – Total reactive oxides of nitrogen

**O<sub>3</sub>** – Ozone, a criteria pollutant

**Ozone precursors** – a group of volatile organic compounds (VOCs) that affect ozone formation and destruction in the atmosphere; also called PAMS pollutants.

**PAMS** – Photochemical Assessment Monitoring Station; a site which measures ozone precursors.

**Particulate matter (PM)** – a complex mix of liquid and/or solid particles in the atmosphere

**Pb** – Lead, a criteria pollutant and a HAP

**PM<sub>2.5</sub>** – Fine particles, 2.5 micrometers in aerodynamic diameter or smaller; a criteria pollutant

**PM<sub>10</sub>** – Inhalable particles, 10 micrometers in aerodynamic diameter or smaller; a criteria pollutant

**PM<sub>2.5</sub>-Speciation** – a group of elements, ionic compounds and carbon compounds that are analyzed from fine particles.

**ppb** - parts per billion, a concentration measurement usually used for gaseous pollutants

**ppm** – parts per million, a concentration measurement usually used for gaseous pollutants

**Real-time** – a system in which data is collected and (almost) immediately presented, usually every hour.

**Risk ratio** – a chemical-specific air concentration divided by its health benchmark; ratios greater than one may indicate a public health concern

**SO<sub>2</sub>** – Sulfur dioxide, a criteria pollutant

**USEPA** - United States Environmental Protection Agency

**VOC** – Volatile organic compound, a carbon-based chemical compound that is normally gaseous

**µg/m<sup>3</sup>** - micrograms per cubic meter, a concentration measurement, usually used for particulate matter and air toxics